

SCHOOL OF BASIC AND APPLIED SCIENCES

(SBAS)

Programme Handbook (Programme Structure & Evaluation Scheme)

Bachelor of Science (Honours/ Honours with Research) in Physics

Programme Code: 209

FOUR YEAR UNDERGRADUATE PROGRAMME

As per National Education Policy 2020 (Multiple Entry and Exit in Academic Programmes) (with effect from 2024-25 session)

Approved in the 34th Meeting of Academic Council Held on 29 June 2024

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1. Preface

Introduction

The objective of any programme at Higher Education Institute is to prepare their students for the society at large. The K. R. Mangalam University visualizes all its programmes in the best interest of their students and in this endeavour; it offers a new vision to all its Under-Graduate courses.

We are committed to implementing the National Education Policy (NEP) 2020 in its entirety, and to creating a more inclusive, holistic, and relevant education system that will prepare our students for the challenges of the 21st century. With the focus on Outcome-Based Education (OBE), our university is continuously evolving an innovative, flexible, and multidisciplinary curriculum, allowing students to explore a creative combination of credit-based courses in variegated disciplines along with value-addition courses, Indian Knowledge Systems, vocational courses, projects in community engagement and service, value education, environmental education, and acquiring skill sets, thereby designing their own learning trajectory.

All academic programmes offered by the University focus on employability, entrepreneurship and skill development and their course syllabi are adequately revised to incorporate contemporary requirements based on feedback received from students, alumni, faculty, parents, employers, industry and academic experts

The School of Basic and Applied Sciences presents four years undergraduate programme Bachelor of Sciences (Honours/ Honours with Research) Physics according to the New Education Policy-2020. We have designed a flexible choice-based credit system, multidisciplinary approach, and multiple entry and exit options for the duration of 2024-2028.

We are following Curriculum and Credit Framework for Undergraduate Programmes (CCFUP)" incorporating a flexible choice-based credit system (CBCS), Learning Outcomebased Curriculum Framework (LOCF), multidisciplinary approach, and multiple entry and exit options. This will facilitate students to pursue their career path by choosing the subject/field of their interest. The curricula are aligned with the needs of the industry and the job market and is flexible enough to adapt to changing trends and technologies. It integrates cross-cutting issues relevant to professional ethics, gender, human values, environment and Sustainable Development Goals (SDGs).

All the courses are having defined objectives and Learning Outcomes, which will help prospective students in choosing the elective courses to broaden their skills in the field of Physics and interdisciplinary areas. The courses will train students with sound theoretical and experimental knowledge that suits the need of academics and industry. The courses also offer ample skills to pursue research as career in the field of physics. The K. R. Mangalam University hopes the NEP-2020 approach of this four-year undergraduate programme B.Sc. (Hons. / Hons. with Research) Physics will help students in making an informed decision regarding the goals that they wish to pursue in further education and life, at large.

2. NEP-2020: Important features integrated in the curriculum

K.R. Mangalam University has adopted the National Education Policy NEP-2020 to establish a holistic and multidisciplinary undergraduate education environment, aiming to equip our students for the demands of the 21st century. Following the guidelines of NEP-2020 regarding curriculum structure and duration of the undergraduate programme, we now offer a Four-Year Undergraduate Programme with multiple entry and exit points, along with re-entry options, and relevant certifications.

- UG Certificate after completing 1 year (2 semesters with the required number of credits) of study, and an additional vocational course/internship of 4 credits during the summer vacation of the first year.
- UG Diploma after completing 2 years (4 semesters with the required number of credits) of study, and an additional vocational course/internship of 4 credits during the summer vacation of the second year.
- **Bachelor's Degree** after completing 3-year (6 semesters with the required number of credits) programme of study.
- 4-year **bachelor's degree (Honours/Honours with Research)** with the required number of credits after eight semesters programme of study.

• Students who secure 75% marks and above in the first six semesters and wish to undertake research at the undergraduate level can choose a research stream in the fourth year. Upon completing a research project in their major area(s) of study in the 4th year, a student will be awarded **bachelor's degree (Honours/Honours with Research)**.

Advantage of pursuing 4-year bachelor's degree programme with Honours/Honours with Research is that the master's degree will be of one year duration. Also, a 4-year degree programme will facilitate admission to foreign universities.

S. No.	Broad Categories of Courses	Minimum Credit Requirement for Four Year UG
		Program
1	Major (Core)	80
2	Minor	32
3	Multidisciplinary	09
4	Ability Enhancement Course (AEC)	08
5	Skill Enhancement Course (SEC)	09
6	Value-Added Course (VAC)	06-08
7	Summer Internship	02-04
8	Research Project/Dissertation	12
9	Total	160

Table 1: Minimum Credit Requirement for Four Year UG Program

2.1 Categories of Courses

Major: The major would provide the opportunity for a student to pursue in-depth study of a particular subject or discipline.

Minor: Students will have the option to choose courses from disciplinary/interdisciplinary minors and skill-based courses. Students who take enough courses in a discipline or an interdisciplinary area of study other than the chosen major will qualify for a minor in that discipline or in the chosen interdisciplinary area of study.

Students have multiple minor streams to choose from. They can select one minor stream from the available options, which will be pursued for the entire duration of the programme.

Multidisciplinary (Open Elective): These courses are intended to broaden the intellectual experience and form part of liberal arts and science education. These introductory-level courses may be related to any of the broad disciplines given below:

- Natural and Physical Sciences
- Mathematics, Statistics, and Computer Applications
- Library, Information, and Media Sciences
- Commerce and Management
- Humanities and Social Sciences

A diverse array of Open Elective Courses, distributed across different semesters and aligned with the aforementioned categories, is offered to the students. These courses enable students to expand their perspectives and gain a holistic understanding of various disciplines. Students can choose courses based on their areas of interest.

Ability Enhancement Course (AEC): Students are required to achieve competency in a Modern Indian Language (MIL) and in the English language with special emphasis on language and communication skills. The courses aim at enabling the students to acquire and demonstrate the core linguistic skills, including critical reading and expository and academic writing skills, that help students articulate their arguments and present their thinking clearly and coherently and recognize the importance of language as a mediator of knowledge and identity.

Skills Enhancement Courses (SEC): These courses are aimed at imparting practical skills, hands-on training, soft skills, etc., to enhance the employability of students.

Value-Added Course (VAC): The Value-Added Courses (VAC) are aimed at inculcating Humanistic, Ethical, Constitutional and Universal human values of truth, righteous conduct,

peace, love, non-violence, scientific and technological advancements, global citizenship values and life-skills falling under below given categories:

- Understanding India
- Environmental Science/Education
- Digital and Technological Solutions
- Health & Wellness, Yoga education, Sports, and Fitness

Research Project / **Dissertation**: Students choosing a 4-Year Bachelor's degree (Honours/Honours with Research) are required to take up research projects under the guidance of a faculty member. The students are expected to complete the Research Project in the eighth semester. The research outcomes of their project work may be published in peer-reviewed journals or may be presented in conferences /seminars or may be patented.

3. University Vision and Mission

3.1 Vision

K.R. Mangalam University aspires to become an internationally recognized institution of higher learning through excellence in inter-disciplinary education, research, and innovation, preparing socially responsible life-long learners contributing to nation building.

3.2 Mission

> Foster employability and entrepreneurship through futuristic curriculum and progressive

pedagogy with cutting-edge technology

- Instill the notion of lifelong learning through stimulating research, Outcomes-based education, and innovative thinking
- Integrate global needs and expectations through collaborative programs with premier universities, research centers, industries, and professional bodies.
- Enhance leadership qualities among the youth understanding ethical values and environmental realities

4. About the School of Basic and Applied Sciences

The School of Basic and Applied Science imparts both teaching and research through its four disciplines of Physics, Chemistry, Mathematics and Forensic science.

SBAS imparts students' disciplinary knowledge, enhances their skills and ability, motivating them to think ingeniously, helping them to act independently and take decisions accordingly in all their scientific pursuits and other endeavours. It strives to empower its students and faculty members to contribute for the development of society and Nation.

The faculty is in constant touch with various experts in the relevant fields and is willing to experiment with latest ideas in teaching and research.

5. School Vision and Mission

5.1 Vision

To be a premier school for advance learning and research in the field of basic and applied sciences.

5.2 Mission

1. Collaborations with national, international academic & research organisations and industries for knowledge creation, advancement, and application of innovative practises in sciences.

2. Create conducive environment for lifelong learning.

3. Empower students to be socially responsible and ethically strong individuals through valuebased science education.

6. About the Programme: Bachelor of Science (Honours / Honours with Research) in Physics

The B.Sc. (Hons. / Hons. with Research) Physics is an undergraduate academic program designed to provide students with a comprehensive foundation in the fascinating world of physics. This program offers a rigorous and engaging curriculum that covers diverse areas such as classical mechanics, electromagnetism, quantum mechanics, thermodynamics, and more. Students in this program will study fundamental principles of the universe, engage in complex problem-solving, and enhance their analytical and critical thinking skills. Combining hands-on lab work with theoretical learning, they'll gain practical experience and a solid grasp of scientific research methods. The B.Sc. (Hons. / Hons. with Research) Physics program prepares students for exciting career opportunities in scientific research, technology, education, and various other fields that require a strong grasp of physics principles and applications.

6.1. Nature of B.Sc. (Hons. / Hons. with Research) Physics Programme

Taking the NEP-2020 as an opportunity to review our existing academic programs and redesign them for a more holistic, multidisciplinary and inclusive education, SBAS, K.R. Mangalam University is transforming its academic structure in a phased manner. School of Basic and Applied Sciences is offering Four Year Undergraduate Degree programme B.Sc. (Hons. / Hons. with Research) Physics with Multiple Entry- Multiple Exit option from the academic session 2023-24. Through multiple entry/exit option, students will be able to enter and exit the program at various stages. This course emphasized hands on practice, innovative thought process and project-based learning.

6.2. Aim of B.Sc. (Hons. / Hons. with Research) Physics Programme

The aims of the B.Sc. (Hons. / Hons. with Research) Physics program, in accordance with the National Education Policy (NEP), are multifaceted and comprehensive. The program aims to cultivate a strong foundation in physics principles and foster a deep understanding of the subject. It seeks to promote critical thinking, analytical skills, and problem-solving abilities among students, enabling them to address real-world challenges effectively. The B.Sc. (Hons. / Hons. with Research) Physics program also encourages research-oriented thinking and provides opportunities for students to engage in scientific inquiry and exploration. By emphasizing hands-on laboratory work and practical applications, the program aims to equip students with the necessary skills for conducting experiments and analysing data. Moreover, the program seeks to foster an interdisciplinary approach, enabling students to connect physics with other scientific disciplines and societal issues. Overall, the B.Sc. (Hons. / Hons. with Research) Physics program aspires to produce well-rounded graduates with a passion for learning and a strong foundation in physics, ready to make significant contributions to the scientific community and society at large.

6.3 Definitions

Programme Outcomes (POs)

Programme Outcomes are statements that describe what the students are expected to know and would be able to do upon the graduation. These relate to the skills, knowledge, and behavior that students acquire through the programme.

Programme Specific Outcomes (PSOs)

Programme Specific Outcomes are statements about the various levels of knowledge specific to the given program which the student would be acquiring during the program.

Programme Educational Objectives (PEOs)

Programme Educational Objectives of a degree are the statements that describe the expected achievements of graduates in their career, and what the graduates are expected to perform, achieve and how they will conduct professionally during the first few years after graduation.

> Credit

Credit refers to a unit of contact hours/tutorial hours per week or 02 hours of lab/practical work per week.

6.4 Programme Educational Objectives (PEO)

These are deferred outcomes measured a few years after completion of the programme, where:

PEO 1: Graduates will become globally competent professionals, suitable for careers in government, corporate, and research sectors, while also possessing skills for entrepreneurial ventures in multidisciplinary fields.

PEO 2: Graduates will demonstrate strong technical knowledge in physics and will be able to think critically and creatively to develop practical solutions to scientific and technological challenges.

PEO 3: Graduates will be ethical professionals who can lead and collaborate effectively in teams, contributing to both their own growth and the success of their organizations.

PEO4: Graduates will engage in lifelong learning and professional development, applying their expertise to benefit society and make meaningful contributions.

PEO5: Graduates will be well-equipped to pursue higher education and advanced research in physics and related disciplines.

6.5 Programme Outcomes (PO)

At the time of graduation, students of undergraduate degree program will be able:

PO1: To apply physics principles and mathematical techniques to effectively solve complex real-world problems.

PO2: To develop strong critical thinking skills, enabling them to analyze and evaluate evidence, arguments, and methodologies in physics research and applications.

PO3: To promote teamwork and collaborative learning, enabling students to contribute to and lead interdisciplinary research.

PO4: To build strong interpersonal skills to navigate diverse professional environments, collaborate effectively, and take on leadership roles.

PO5: To communicate complex scientific ideas and research findings clearly in both writing and presentations to scientific and non-scientific audiences.

PO6: To learn independently and stay updated with scientific advancements throughout their careers.

PO7: To acquire a strong commitment to ethical standards, responsible conduct in research, and professionalism in academic and industry settings.

PO8: To understand the impact of physics research on the environment and society and show a commitment to using sustainable and ethical practices in work.

6.6 Programme Specific Outcomes (PSO)

At the time of graduation, students of undergraduate degree program will be:

PSO1: Understanding fundamental concepts, laws, theories, tools and techniques in the field of physics.

PSO2: Applying mathematical techniques, theoretical models, and physical laws to solve complex scientific problems.

PSO3: Analyzing various real-life situations, problems, and challenges related to physics.

PSO4: Evaluating the validity of scientific arguments, theories, and experimental results, making reasoned judgments about the quality and reliability of research findings in physics.

PSO5: Operating laboratory instruments and experimental setups with accuracy, demonstrating skill in measuring, observing, and manipulating equipment to gather precise data.

PSO6: Exhibiting coordinated manual skills in conducting physics experiments, ensuring safe and efficient handling of materials, tools, and equipment while following detailed procedural steps.

6.7 Career Avenues

Graduates can pursue careers in research and development (R&D) in industries, work as research scientists or assistants in academic and research institutions, or continue with higher education (M.Sc., PhD) leading to academic or specialized roles in physics. Opportunities also exist in sectors like govt. jobs, defence services, data science, finance, and competitive examinations.

6.8 Duration

The duration of this programme is four years (eight semesters) with multiple entry/exit options.

Name of Degree	Credits requirement	Completion Year
UG Certificate Physics		First Year
	43	
UG Diploma Physics	96/94	Second Year
B.Sc. Physics	149/145	Third Year
B.Sc. (Hons. / Hons. With	173	Fourth Year
Research) Physics		

6.9 Criteria for award of certificates and degree

7. Student's Structured Learning Experience from Entry to Exit in the Programme

7.1 Education Philosophy and Purpose:

Learn to Earn a Living: At KRMU we believe in equipping students with the skills, knowledge, and qualifications necessary to succeed in the job market and achieve financial

stability. All the programmes are tailored to meet industry demands, preparing students to enter specific careers and contributing to economic development.

Learn to Live: The university believes in the holistic development of learners, fostering sensitivity towards society, and promoting a social and emotional understanding of the world. Our aim is to nurture well-rounded individuals who can contribute meaningfully to society, lead fulfilling lives, and engage with the complexities of the human experience.

7.2 University Education Objective: Focus on Employability and Entrepreneurship through Holistic Education using Bloom's Taxonomy

By targeting all levels of Bloom's Taxonomy—remembering, understanding, applying, analyzing, evaluating, and creating—students are equipped with the knowledge, skills, and attitudes necessary for the workforce and entrepreneurial success. At KRMU we emphasize on learners critical thinking, problem-solving, and innovation, ensuring application of theoretical knowledge in practical settings. This approach nurtures adaptability, creativity, and ethical decision-making, enabling graduates to excel in diverse professional environments and to innovate in entrepreneurial endeavours, contributing to economic growth and societal wellbeing.

7.3 Importance of Structured Learning Experiences

A structured learning experience (SLE) is crucial for effective education as it provides a clear and organized framework for acquiring knowledge and skills. By following a well-defined curriculum, teaching-learning methods and assessment strategies, learners can build on prior knowledge systematically, ensuring that foundational concepts are understood before moving on to more complex topics. This approach not only enhances comprehension but also fosters critical thinking by allowing learners to connect ideas and apply them in various contexts. Moreover, a structured learning experience helps in setting clear goals and benchmarks, enabling both educators and students to track progress and make necessary adjustments. Ultimately, it creates a conducive environment for sustained intellectual growth, encouraging learners to achieve their full potential. At K.R. Mangalam University SLE is designed as rigorous activities that are integrated into the curriculum and provide students with opportunities for learning in two parts:

• Inside the classroom: The approach focuses on cognitive outcomes using studentcentric learning methods. Techniques like problem-solving, simulations, and group projects promote active engagement and critical thinking. Tools such as Scilab, LaTeX, and digital simulations are integrated to enhance practical understanding. Peer reviews and discussions further solidify theoretical knowledge through collaborative learning.

• **Outside the classroom**: Activities aim to develop people and psychomotor skills through industry, community, and lab engagements. Internships, lab experiments, and community projects provide hands-on experience, fostering teamwork, communication, and technical expertise. Students apply theoretical knowledge in real-world settings, gaining essential skills for professional and interpersonal growth.

7.4 Educational Planning and Execution: What, when and how learning will happen

The School of Basic and Applied Sciences (SBAS) emphasizes a holistic approach to educational planning and execution, ensuring that both academic and personal development are seamlessly integrated into the student experience. The curriculum encompasses core subjects that establish a solid academic foundation, complemented by open electives, discipline-specific electives, Value-Added Courses (VAC), and Ability Enhancement Compulsory Courses (AECC) to expand intellectual perspectives. In addition, students are offered the opportunity to pursue a Minor in fields such as Environmental Science, Data Science, Artificial Intelligence & Machine Learning, and Nanoscience, enhancing their specialization in the four-year bachelor's degree course. The selection of these minors happens in the first semester, continuing throughout the degree program.

The learning is thoughtfully planned across the curriculum. In the early stages, foundational knowledge and skills are built through core courses. As students' progress, learning becomes more specialized, with electives and minors supporting deeper exploration of disciplines. Co-curricular activities, including sports, technical events, and cultural activities, are integrated throughout to ensure all-around growth. Leadership training, teamwork, communication skills, and discipline are emphasized through structured personality development activities. Ethical values such as truthfulness, gender sensitization, and environmental consciousness are instilled from the outset, becoming a continuous part of the student journey.

At SBAS, learning is dynamic and flexible, utilizing a variety of teaching methods including lectures, case-based learning, problem-based learning, and project-based learning, all aimed at fostering critical thinking and problem-solving abilities. Hands-on learning is reinforced through lab sessions, internships, research projects, and practical activities that connect

theoretical knowledge to real-world applications. Workshops, seminars, and guest lectures from industry experts further provide practical insights and professional exposure. We have a strong students' support system in terms of differential learning (slow & fast learning), mentormentee system and personal counselling thereby ensuring students move up on the learning curve.

In terms of infrastructure, SBAS supports its academic planning with highly qualified faculty, smart classrooms, a well-equipped library, computer labs, and experimental research facilities. The inclusion of Massive Open Online Courses (MOOCs) and experiential learning ensures that students are prepared for both academic success and professional excellence. This carefully executed planning ensures that students are engaged at all levels of Bloom's Taxonomy, progressing from foundational understanding to higher-order thinking, while also fostering emotional, social, and ethical development. Continuous stakeholder feedback, including input from faculty, industry experts, students, and alumni, ensures that the curriculum remains relevant, aligned with academic advancements, and tuned to industry needs.

7.5 Course Registration and Scheduling

- ✓ Major and Minor Selection Every student must register at the beginning of each semester for the courses offered in the given semester. Major courses are registered centrally for the students. However, for other multidisciplinary courses (Minor, VAC, OE) the students must register by themselves through ERP.
- ✓ Internships/ Research Project- Students need to do summer internship after second and fourth semesters, which carries 2 credits, during the summer breaks. The same will be evaluated in the upcoming odd semester. In the eighth semester students of B.Sc. (Hons. / Hons. with Research) Physics will do Research Project (Dissertation). Projects are also mapped along with the Lab/ Practical Courses and Experiential Learning Activities.
- ✓ Cocurricular Activities Credit Choices: Participation in Co/ Extracurricular activities is part of outside classroom learning.

Students must earn 2 credits from co/ extracurricular activities. One credit from participation in co-curricular activities like Club/Society activities and another credit

from Community Service (1 credit each) through participation in NSS/ Redcross activities or NGOs that contribute to their personal development, leadership skills, and community engagement.

- Under the category of **Club/Society**, 1 credit can be earned by registration in one of the Club/Societies of university and active participation in the events organized by the club/society **OR**
- 15 hours of active engagement in any of the recreational/sports activities Under the category of **Community Service**, 1 credit can be earned by
- 15 hours active engagement in community service through NGO/NSS/Redcross or any other society approved/ empanelled by the university.
- At the end of the semester, students are required to submit a log of hours, a report, and a certificate of participation/ completion summarizing their activities followed by a presentation.

7.6 Academic Support Services

The School of Basic and Applied Sciences offers a variety of academic support services tailored to meet the diverse learning needs of its students, ensuring success for all. These services include:

- **Personalized Tutoring**: One-on-one sessions with experienced tutors focus on specific areas such as laboratory techniques, experimental design, research projects, data analysis, and theoretical understanding. Tutoring is customized to each student's level, allowing for targeted support in areas like crystal structure analysis, magnetic properties, and dielectric behaviour.
- Workshops and Seminars: Regular workshops on topics such as advanced scientific research methods, materials characterization techniques, and the latest advancements in nanotechnology and superconductivity. These workshops, alongside industry connections, help students enhance both practical skills and theoretical knowledge.
- **Peer Mentoring Programs**: Advanced learners' mentor fellow students by leading study groups, assisting with assignments, and guiding practical projects, fostering a collaborative and supportive academic environment.

- Accessible Learning Resources: A variety of online platforms provide access to resources such as recorded lectures, research papers, interactive simulations, and experimental procedure guides, catering to different learning styles and enhancing independent study.
- Outcome-Based Activities: Students are encouraged to engage in hands-on practical, such as conducting experiments on material properties, to produce meaningful results. These outcomes are then showcased and celebrated, motivating students to further develop their skills.
- **Diversity and Inclusion Initiatives**: Programs promoting diversity and inclusion ensure that all students, regardless of background, feel valued and can contribute to a rich, collaborative learning environment.
- Feedback and Assessment: Continuous feedback mechanisms provide students with constructive evaluations of their work, allowing them to refine their techniques, improve their understanding, and achieve academic excellence.

7.7 Student Career & personal Support Services

• Mentor Mentee Relationship

Every student enrolled in the school is considered a mentee and will be assigned a faculty member as their mentor. The mentor's role is to guide and support the mentee, helping them grow both personally and professionally. Mentors act as coaches by giving feedback, sharing advice, and offering insights from their own experiences. They also challenge the mentee's thinking, help them make important decisions, and connect them to valuable resources and networks. Additionally, mentors provide emotional support, celebrating successes and offering encouragement during tough times. On the other hand, the mentee's role is to actively participate in the learning process by planning meetings, setting goals, and communicating openly with their mentor. Mentees should also apply what they learn, continue growing outside the mentor-mentee relationship, and stay proactive in seeking new opportunities. By staying committed and enthusiastic, mentees can make the most of this relationship and achieve their goals.

• Counselling and Wellness Services

Counselling and wellness services typically encompass a range of resources to support students' mental health, emotional well-being, and overall quality of life. The school has various counselling programs such as individual Counselling where one-on-one sessions with licensed counsellors or psychologists are held to address personal issues, stress, and mental health concerns, **Group Counselling** which support groups for shared experiences like anxiety, depression, or adjustment challenges, **Crisis Counselling** for Immediate support for students in urgent situations or experiencing severe emotional distress, **Career Counselling** for guidance on career planning, job search strategies, and professional development and **Academic Counselling** for managing academic stress, time management, and study strategies. School also has various Wellness Services like On-campus clinics which provides medical care, including physical exams, vaccinations, and treatment for minor illnesses. Various mental health workshops on topics like stress management, mindfulness, and coping strategies are organized. All the students have access to gyms, fitness classes to promote physical health. These services aim to support students in maintaining a balanced and healthy lifestyle while managing the demands of university life.

• Career Services and Training

Career services and training programs are designed to support students in their professional development and job search. School provides personalized advice on career paths, goal setting, and job search strategies to students. They are given proper guidance on creating and refining job application materials. Mock interviews are also held. They are given opportunities to connect with alumni, professionals, and potential employers. Students are given professional training in areas like communication, leadership, and time management. These services and programs aim to prepare students for successful careers by enhancing their skills, providing practical experience, and connecting them with potential employers.

8. Assessment and Evaluation

8.1 Evaluation scheme for theory courses

Evaluation Component	Weightage		
Internal Marks (Theory): -			
I) Continuous Assessment (30 Marks)			
(All the components to be evenly spaced)			
Projects/ Quizzes/ Assignments and Essays/			
Presentations/ Participation/ Case Studies/ Reflective	30 Marks		
Journals (minimum of five components to be			
covered)			
II) Mid Term Exam	20 Marks		
External Marks (Theory): -			
End Term Examination	50 Marks		

* (It is compulsory for a student to secure 40% marks in the Internal and End Term Examination separately to secure minimum passing grade).

Overview of Internal Evaluation (30 Marks) -

Internal evaluation is designed to assess students' ongoing learning and application of course materials through diverse assessment methods. Instructors have full autonomy within the 30 marks to employ assessment strategies that best align with the course's learning objectives.

Recommended Assessment Types: -

Projects: - Individual or group projects focusing on research, analysis, and practical application of concepts.

Quizzes: - Regular, short assessments to evaluate understanding of the material.

Assignments and Essays: - In-depth tasks to assess critical thinking and problem-solving skills.

Presentations: - Assessing knowledge dissemination and communication skills.

Participation: - Evaluation of engagement and contributions to class activities.

Case Studies: - Application of theoretical knowledge to real-world scenarios.

8.2 Evaluation scheme for practical courses

Particular	Weightage
Internal Marks (Practical): -	
I) Conduct of Experiment	10 Marks
II) Lab Records	10 Marks
III) Lab Participation	10 Marks
IV) Lab Project	20 Marks
External Marks (Practical): -	
End Term Practical and Viva Voce	50 Marks

* (It is compulsory for a student to secure 40% marks in Internal and End Term Practical's and Viva Voce separately to secure minimum passing grade).

8.3 Evaluation scheme for research project

Particular	Weightage
Internal Marks: -	
(Punctuality, Performance, Work	50 Marks
Ethics, Efforts and Research Output)	
External Marks (Practical): -	50 Marks
Presentation	20
Report Writing	10
Viva Voce	20

*(It is compulsory for the student to provide an internship certificate issued by the relevant institution or organization where they completed their internship during the evaluation process.)

8.4 Evaluation scheme for Internship

Particular	Weightage		
Internal Marks: -			
Internship completion certificate obtained from	30 Marks		
supervisor from host institute.			
External Marks (Practical): -	70 Marks		
Presentation	25		
Report Writing	25		
Viva Voce	20		

8.5 Grading System

Based on the performance in all evaluation components of a Course, each student will be awarded a final grade in the Course registered, at the end of the semester. The total marks obtained by a student in the Course will be converted to a corresponding letter grade as described below.

Marks Range (%)	Letter Grade	Grade Points	Description of the Grade
% marks > 90%	0	10.0	Outstanding
$80 < \%$ marks ≤ 90	A+	9.0	Excellent
$70 < \%$ marks ≤ 80	А	8.0	Very Good
$60 < \%$ marks ≤ 70	B+	7.0	Good
$55 < \%$ marks ≤ 60	В	6.0	Above Average
$50 < \%$ marks ≤ 55	С	5.5	Average
$40 \le \%$ marks ≤ 50	Р	5.0	Pass
% marks < 40	F	0	Fail
-	AB	0	Absent
%marks ≥ 50	S	-	Satisfactory
% marks < 50	U	-	Unsatisfactory
-	W	0	Withdrawal

9. Feedback and Continuous Improvement Mechanisms:

Teaching-learning is driven by outcomes. Assessment strategies and andragogy are aligned to course outcomes. Every CO is assessed using multiple components. The attainment of COs is calculated for every course to know the gaps between the desired and actual outcomes. These gaps are analysed to understand where does the student lags in terms of learning levels. Thereafter each student's learning levels are ascertained, if found below desirable level, and intervention strategy is affected in the following semester to make necessary corrections. To cater to the diverse learning needs of its student body, K.R. Mangalam University employs a comprehensive assessment framework to identify both slow and advanced learners. Students' learning levels are continually assessed based on their performance at various stages. If a student's performance in internal assessments falls below or equal to 55%, they are categorized as slow learners. Conversely, if a student's performance score in internal assessments is greater than or equal to 80%, they are identified as advanced learners. Such students are encouraged to participate in advanced learning activities. Through periodic evaluations and the utilization of modern management systems, the institution adeptly tracks students' performance across various courses, allowing for targeted interventions and support mechanisms.

10. Academic Integrity and Ethics

The School of Basic and Applied Sciences (SBAS) is committed to promoting safety and academic integrity by enforcing rigorous behavioural standards. Alcohol consumption and substance abuses are strictly prohibited, with escalating penalties for repeat offenders, which may include rustication. Ragging is also banned, adhering to UGC regulations and Supreme Court directives, and is managed through a comprehensive anti-ragging policy. The Anti-Ragging Committee, led by student affairs advisors and comprising diverse members, is tasked with handling ragging complaints and making recommendations. The Anti-Ragging Squad plays a proactive role by monitoring the campus, patrolling potential ragging hotspots, and investigating incidents. Penalties for violations can range from suspension and withholding benefits to expulsion and filing an FIR, in line with UGC regulations.

Sexual harassment in any form is taken very seriously and will be addressed by the Internal Committee Against Sexual Harassment in accordance with the Institute's policies. The school also enforces strict penalties for other forms of misconduct, including possession of weapons, theft, and misuse of Institute property or facilities. These actions are subject to severe disciplinary measures.

Academic integrity is a cornerstone of SBAS's research and educational missions. It encompasses honesty, responsibility, and the proper acknowledgment of others' contributions. Violations such as plagiarism and cheating are treated as serious offenses. Students are required to follow principles of academic integrity, including proper citation, ethical data collection, and respect for others' work. Examples of misconduct include copying, falsifying data, and submitting purchased materials. The Institute provides guidelines for accurate record-keeping, truthful reporting, and proper attribution to uphold high academic standards.

Both individual and collective responsibility are emphasized in maintaining academic integrity. Students must ensure their theses are free from plagiarism and original before submission and are encouraged to report any violations. Faculty members are responsible for guiding students in proper research methods, ensuring accurate data recording, and reviewing student work. Additionally, faculty must educate students on academic integrity and address any breaches.

Reporting academic violations involves several steps. Faculty members should report breaches to the School Dean, and any student-faculty conflicts are managed by the Dean with committee support. The Director may appoint a committee to investigate scientific misconduct. Penalties for academic breaches are severe, with initial offenses resulting in warnings or an "F" grade, and repeat offenses potentially leading to expulsion.

Students must also seek permission before engaging with media on behalf of the Institute or recording classroom activities. Unauthorized sharing of audio/video clippings or posting derogatory comments on social media is prohibited. Misconduct can be reported by students, staff, or faculty, and penalties may include warnings, community service, restrictions, fines, withholding grades, suspension, expulsion, or a ban on reapplying for admission. The disciplinary process involves a hearing, documentation, and recommendations by a committee, with final actions decided by the Dean and enforced by the academic office. Repeat offenders face harsher penalties.

11. Programme Structure

SEMESTER -	[
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S.N 0.	TYPE OF COU RSE	COURS E CODE	COURSE TITLE	L	Т	Р	С	Multiple Entry and Exit
1	Major 1	SCPH10 1	Mathematical Physics I	4	0	0	4	
2	Major 2	SCPH10 3	Electricity and Magnetism	4	0	0	4	
3	Major 2 (Lab)	SCPH15 1	Electricity and Magnetism LAB	0	0	4	2	Award:
5	VAC-I		VAC	2	0	0	2	UG Certificate
6	SEC-I	SEC015	Basic Instrumentation Skills	3	0	0	3	[after completing 1 year of study (2 semesters with credits as prescribed), and an
7	Minor 1	UNS101 / UEV101 / UDT101 / UDT101	Minor NS/EVS/DS/AI&M L	4	0	0	4	additional vocational course/internship of 4 credits to be covered within 6-8 weeks during the summer vacation of the first year]
8	CS001	CS001	CLUB/SOCIETY	0	0	0	1	
ΤΟΤΑ	AL (Mino	r NS/EVS/	DS/AI&ML)				20	
SEMS	SEMSTER-II							
S.No	TYPE OF COUR SE	COURS CODE	E COURSE TITLE	Ι	. T	Р	С	

1	Major 3	SCPH102	02 Mathematical Physics II 4		0	0	4
2	Major 4	SCPH104	Mechanics	4	0	0	4
3	Major 4 (Lab)	SCPH152 Mechanics Lab		0	0	4	2
4	VAC-II		VAC	2	0	0	2
5	Open Electiv e I		OE	3	0	0	3
6	SEC-II	SEC016	Physics Workshop Skill	3	0	0	3
7	Minor2	UNS102/ UEV102/ UDT102/ UDT102	Minor NS/EVS/DS/AI& ML	4	0	0	4
8	Summe r Interns hip/Pro ject	SIPH001	Summer Internship I				
9	CS002	CS002	COMMUNITY SERVICES	0	0	0	1
TOTAL (Minor NS/EVS/DS/AI&ML)							

SEMSTER-III

S.N 0	TYPE OF COURSE	COURS E CODE	COURSE TITLE	L	Т	Р	С	Multiple Entry and Exit
1	Major 5	SCPH201	Mathematical Physics III	4	0	0	4	
2	Major 6	SCPH203	SolidStatePhysics	4	0	0	4	
3	Major 6 (Lab)	SCPH253	SolidStatePhysics Lab	0	0	4	2	Award: UG Diploma
4	Minor 3	UNS103/ UEV103/ UDT103/ UDT103	Minor NS/EVS/DS/AI& ML	4	0	0	4	[after completing 2 years of study (4 semesters with credits as prescribed), and an
5	Open Elective II		OE	3	0	0	3	additional vocational
6	VAC-III		VAC	2	0	0	2	course/internship of
7	AEC-I	AEC001	Verbal Ability	3	0	0	3	4 credits to be covered within 6-8
8	Major 7	SCPH205	Modern Physics	4	0	0	4	weeks during the
9		SIPH001	Evaluation of Summer Internship I	2	0	0	2	summer vacation of the second year]
ΤΟΤΑ	AL (Minor N	S/EVS/DS/A	AI&ML)	<u>I</u>	1	<u>I</u>	28	Entry: The student who took exit after
SEM	ESTER -IV							completion of the first year (UG
S.N 0	TYPE OF COURSE	COURS E CODE	COURSE TITLE	L	Т	Р	С	Certificate) is allowed to enter the

1	Major 8	SCPH20 2	QuantummechanicsandApplications	4	0	0	4	diploma programme within five years from the first entry in the
2	Major 9	SCPH20 4	Waves and Optics	4	0	0	4	programme, four years in case of
3	Major 9 (Lab)	SCPH25 4	Waves and Optics Lab	0	0	4	2	degree program and three years in case of Hons. degree to
4	Open Elective III		OE	3	0	0	3	complete the programme within
5	SEC- III/MOOC	SEC075	Basic IT tools	3	0	0	3	the stipulated time period of seven years.
6	AEC-II	AEC002	Communication & Personality Development	3	0	0	3	
7	Minor 4	UNS104/ UEV104 / UDT104 / UDT109	Minor NS/EVS/DS/AI& ML	4	0	0	4	
8	Minor 4 (Lab) *Applicabl e for Minor stream Nanoscien ces	UNS105	Minor NS (lab)	0	0	2	2	

9	Summer Internship/ Project	SIPH002	Summer Internship II					
	AL (Minor	25 /2 3/ 23 /2 3						
ODD S.N o	semester- TYPE OF COURS E	COUR	COURSE TITLE	L	Т	Р	С	Multiple Entry and Exit
1	Major 10	SCPH3 01	Thermodynamics nd Statistical Mechanics	4	0	0	4	
2	Major 11		Digital Systems and Applications	4	0	0	4	
3	Major 11 (Lab)		Digital Systems and Applications Lab	0	0	4	2	
4	Minor 5	5/	ліпоr №/EVS/DS/AI&M _	4	0	0	4	Award: Bachelor's Degree

5	Major 12 /MOOC	SCPH3 07	Laser Fundamentals	s Z	1	2	0	6	[after completing 3- year of study (6 semesters with credits as
6	AEC-III	AEC00 3	Arithmetic Reasoning-III	3	3	0	0	3	prescribed)]
7	Major 13	SCPH3 05	Classical dynamics	2	1	0	0	4	Entry: The student who took exit after completion of two
8	Summer Internshi p/Projec t	SIPH00 2	Evaluation o Summer Internshi /Project II	of p 2	2	0	0	2	years of study (UG Diploma) are allowed to re-enter the degree
ΤΟΤΑ	AL (Minor	29	programme within three years and complete the degree						
EVE	N SEMEST								
			,111						programme within
S.N o	TYPE OF COURS E	COUR SE CODE	COURSE TITLE	L	Т	Р	C	1	programme within the stipulated maximum period of seven years.
S.N	TYPE OF COURS	COUR SE	COURSE	L 4	T	P	4	-	the stipulated maximum period of
S.N 0	TYPE OF COURS E	COUR SE CODE SCPH3	COURSE TITLE Analog Systems						the stipulated maximum period of
S.N o 1	TYPE OF COURS E Major 14	COUR SE CODE SCPH3 02 SCPH3	COURSE TITLE	4	0	0	4		the stipulated maximum period of

5	Minor 6	UNS10 7/ UEV10 6/ UDT10 6/ UDT10 6	Minor NS/EVS/DS/AI& ML	4	0	0	4		
6	Minor 6 (Lab) *Applica ble for Minor stream Nanoscie nces	UNS10 8	Minor NS (lab)	0	0	4	2		
7	Major 17	SCPH3 08	Atomic, and Molecular Physics	4	0	0	4		
			9S/AI&ML)				24/22 /22/2 2		
Odd	Semester-V	ΊI							
S.N 0	TYPE OF COURS E	COUR SE CODE	COURSE TITLE	L	Т	Р	С	Multiple and Exit	Entry
3	Major 18	SCCH4 01	Research Methodology	4	0	0	4		

4	Minor7	UNS10 9/ UEV 107/ UDT10 7/ UDT11 1	Minor NS/EVS/DS/AI& ML	4	0	0	4
5	SEC 4	SEC076	Technology in Research	2	0	0	2
6	SEC 5	SEC077	Research Ethics and Report writing	2	0	0	2
TOTA	AL (Minor	·NS/EVS/D	S/AI&ML)	<u>I</u>	1	1	12
*Bacl	helor's De	gree (Hono	urs with Research)	Sem	este	r-VI	II
S.N 0	TYPE OF COUR SE	COURS E CODE	COURSE TITLE	L	Т	Р	C
2	Major 19	SCPH40 4	Research Project				12
2	_		Research Project Minor EVS/DS/AI&ML	4	0	0	4
3	19 Minor8	4 UEV108 / UDT108 / UDT112	Minor	4	0	0	

ry: The student took exit after pletion of three s of study (UG ree) is allowed re-enter the ee programme imum within years and e plete the degree gramme within stipulated imum period of n years.

S.N 0	TYPE OF COUR SE	COURS E CODE	COURSE TITLE	L	Т	Р	С
1	Minor8	UEV108 / UDT108 / UDT112	Minor EVS/DS/AI&ML	4	0	0	4
2	Major 19	SCPH40 2	Advanced Quantum Mechanics-II	4	0	0	4
3	Major 20	SCPH40 6	Advances Solid State Physics-II	4	0	0	4
4	Major 20 (Lab)	SCPH45 2	Advances Solid State Physics-II Lab	0	0	4	2
5	SECVI	SCPH45 4	Latex Lab	0	0	4	2
TOTA	AL (Minor	r NS/EVS/D	S/AI&ML)	1	I	1	12/1 6/16/ 16

Total Credits: 173

DETAILS OF MINOR COURSES

Minor-	Nanoscience			
S.No.	Minor	Name	Code	Semester
Ι	Minor1	Study Of Materials	UNS101	Ι
II	Minor2	Elements Of Nanosciences and Nanomaterials	UNS102	II
III	Minor 3	Nanostructured Materials	UNS103	III
IV	Minor 4	Crystallography	UNS104	IV
IV	Minor 4 (Lab)	Crystallography Lab	UNS105	IV
V	Minor5	Synthesis Of Nanomaterials-I	UNS106	V
VI	Minor6	Synthesis Of Nanomaterials-II	UNS107	VI
VI	Minor 6 (Lab)	Synthesis Of Nanomaterials Lab	UNS108	VI
VII	Minor7	Characterisation Techniques of Nanomaterials	UNS109	VII

Minor- Environmental Science								
S.No.	Minor	Name	Code	Semester				
I	Minor1	Earth And Earth Surface Processes	UEV101	Ι				
II	Minor2	Hydrology And Hydrogeology	UEV102	II				
III	Minor 3	Natural Resources Management and Sustainability	UEV103	III				
IV	Minor 4	NaturalAndAnthropogenicHazards	UEV104	IV				
v	Minor5	Environment Legislation Policies and ESG's	UEV105	V				

VI	Minor6	Waste Management	UEV106	VI
VII	Minor7	Environmental Impact Assessment and Risk Assessment	UEV107	VII
VIII	Minor8	SDG's And Climate Change	UEV108	VIII

Minor- Da	ata Science				
S.No.	Minor	Name		Code	Semester
Ι	Minor1	Data Analytics Using SQL		UDT101	Ι
II	Minor2	Data Analytics using R	II	UDT102	II
III	Minor 3	Python for Data Science	III	UDT103	III
IV	Minor 4	Data Preprocessing and Visualization Using Python	IV	UDT104	IV
v	Minor5	Time Series Analysis & Forecasting Using Python	V	UDT105	V
VI	Minor6	Fundamental Of Machine Learning	VI	UDT106	VI
VII	Minor7	Data Driven Applications	VII	UDT107	VII
VIII	Minor8	Project and Case Study	VIII	UDT108	VIII

Minor-	Minor- Artificial Intelligence & Machine Learning									
S.No.	Minor	Name	Code	Semester						
Ι	Minor 1	Data Analytics using SQL	UDT101	Ι						
II	Minor 2	Data Analytics using R	UDT102	II						
III	Minor 3	Python for Data Science	UDT103	III						
IV	Minor 4	Data Structures and Algorithms	UDT109	IV						
		Fundamentals of Artificial		V						
V	Minor 5	Intelligence	UDT110							
VI	Minor 6	Fundamental of Machine Learning	UDT106	VI						

		Neural	Network	and	Deep		VII
VII	Minor 7	Learning				UDT111	
VII	Minor 8	Natural Language Processing			UDT112	VIII	

12. Syllabi

SEMESTER I									
SCPH101	Mathematical Physics-I	L	Τ	Р	С				
Version 1.0		4	0	0	4				
Category of Course	Major				•				

Total Contact Hours	60
Pre-Requisites/ Co-	Calculus
Requisites	

Course Perspective This course aims to demonstrate the use of mathematical techniques in solving problems in Physics and to provide a deeper understanding of the mathematics underpinning theoretical physics. The course is intended to develop the concepts of vector calculus and its applications. Emphasis will be on illustrative examples from Physics and Engineering.

Course Outcomes

On completion of the course, the learner will be:

CO1: Understanding key calculus concepts like limits, differentiation, and series.

- CO2: Applying vector operations and theorems such as Gauss's and Stokes' and concepts in curvilinear coordinates and probability distributions.
- **CO3:** Analyzing vector calculus concepts, including vector fields and products.
- CO4: Evaluating the applications of vector integration theorems, and probability distribution functions in solving real-world problems.

Course Content

UNIT-I Calculus

Recapitulation: Limits, continuity, average and instantaneous quantities, differentiation. Intuitive ideas of continuous, differentiable, etc. Approximation: Taylor and binomial series (statements only).

First Order and Second Order Differential equations: First Order Differential Equations and Integrating Factor. Homogeneous Equations with constant coefficients. Wronskian and general solution. Statement of existence and Uniqueness Theorem for Initial Value Problems. Integral.

UNIT-II Vector Calculus

Calculus of functions of more than one variable: Partial derivatives, exact and inexact differentials. Integrating factor, with simple illustration. Constrained Maximization using Lagrange Multipliers.

15 Lectures

15 Lectures

Recapitulation of vectors: Properties of vectors under rotations. Scalar product and its invariance under rotations. Vector product, Scalar triple product and their interpretation in terms of area and volume respectively. Scalar and Vector fields.

UNIT-IIIVectorDifferentiation&Integration15 Lectures

Vector Differentiation: Directional derivatives and normal derivatives. Gradient of a scalar field and its geometrical interpretation. Divergence and curl of a vector field. Del and Laplacian operators. Vector identities.

Vector Integration: Ordinary Integrals of Vectors. Multiple integrals, Jacobian. Notion of infinitesimal line, surface and volume elements. Line, surface and volume integrals of Vector fields. Flux of a vector field. Gauss' divergence theorem, Green's and Stokes Theorems and their applications (no rigorous proofs).

UNIT-IV orthogonal curvilinear coordinates & probability 15 Lectures

Orthogonal Curvilinear Coordinates. Derivation of Gradient, Divergence, Curl and Laplacian in Cartesian, Spherical and Cylindrical Coordinate Systems.

Probability distribution functions; binomial, Gaussian, and Poisson, with examples. Mean and variance

Learning experience

The learning experience in this course will be interactive and applied. Students will engage in problem-solving exercises and practical applications of calculus and vector analysis. They will work on assignments and projects that involve real-world scenarios, using mathematical tools and techniques to solve complex problems. Hands-on activities, group discussions, and demonstrations will help reinforce theoretical concepts, making the learning process dynamic and relevant to their future studies and careers.

Textbooks:

- 1. Mathematical Physics, H.K. Dass and R. Verma, S. Chand & Company.
- 2. Advanced Engineering Mathematics, Erwin Kreyszig, 2008, Wiley India.

Suggested Readings

- 1. Mathematical Methods for Physicists, G.B. Arfken, H.J. Weber, F.E. Harris, 2013, 7th Edn., Elsevier.
- 2. An introduction to ordinary differential equations, E.A. Coddington, 2009, PHI learning
- 3. Differential Equations, George F. Simmons, 2007, McGraw Hill.

Open Educational Resources (OER)

- 1. <u>https://www.youtube.com/playlist?list=PLZHQObOWTQDNPOjrT6KVlfJuKtYTftqH6</u> <u>https://www.youtube.com/watch?v=GLs0nXZBUYk</u>
- 2. https://archive.nptel.ac.in/courses/111/105/111105122/
- 3. <u>https://www.youtube.com/watch?v=KZzAFX54DWs</u>
- 4. https://archive.nptel.ac.in/courses/111/102/111102111/

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER I					
SCPH103	Electricity and Magnetism	L	T	Р	C
Version 1.0		4	0	0	4
Category of Course	Major		1	1	
Total Contact Hours	60				

Pre-Requisites/	Co-	Mathematical Physics
Requisites		

Course Perspective The course on Electricity and Magnetism provides a foundational understanding of the behaviour of electric and magnetic fields, which are essential to many areas of physics and engineering. It introduces students to fundamental concepts, laws, and mathematical techniques used to describe and analyse these fields. The course emphasizes the application of theoretical principles to solve real-world problems, preparing students for advanced studies and practical applications in fields such as electromagnetism, electronics, and materials science. Through this course, students will develop critical thinking and problem-solving skills, enabling them to approach complex physical phenomena with confidence.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding key concepts like electric fields, potentials, Gauss's Law, and Maxwell's equations.

CO 2: Applying Gauss's Law, the Method of Images, and Ampere's Law to solve field-related problems.

CO 3: Analysing relationships between electric and magnetic properties to predict system behaviour.

CO 4: Evaluating different problem-solving approaches in electrostatics and magnetostatics.

Course Content

Unit I: Electrostatics:

Electric field & Electric field lines. Electric flux. Gauss' Law with applications to charge distributions with spherical, cylindrical and planar symmetry. Conservative nature of Electrostatic Field. Electrostatic Potential.

Laplace's and Poisson equations. The Uniqueness Theorem. Potential and Electric Field of a dipole. Force and Torque on a dipole. Capacitance of a system of charged conductors. Parallel-plate capacitor. Capacitance of an isolated conductor.

Unit II: Electric Potential and Field

15 Lecture hours

Special techniques for the calculation of Potential and Field: The Method of Images is applied to a system of a point charge and finite continuous charge distribution (line charge and surface charge) in the presence of (i) a Plane infinite sheet maintained at constant potential, and (ii) a Sphere maintained at constant potential.

Electric Field in Matter: Polarization in matter, Bound charges and their physical interpretation. Field inside a dielectric, Displacement vector D, Gauss' Law in the presence of dielectrics, Boundary conditions for D, Linear dielectrics, Electric Susceptibility and Dielectric Constant, idea of complex dielectric constant due to varying electric field.

Unit III: Magnetostatics

15 Lecture hours

Magnetic Field: Magnetic force between current elements and definition of Magnetic Field B. Biot-Savart's Law and its simple applications. Current Loop as a Magnetic Dipole and its Dipole Moment (Analogy with Electric Dipole). Ampere's Circuital Law and its application to (1) Solenoid and (2) Toroid.

Properties of B: curl and divergence. Vector Potential. Magnetic Force on (1) point charge (2) current carrying wire (3) between current elements. Torque on a current loop in a uniform Magnetic Field.

Unit IV: Magnetic Properties & Electrodynamics 15 Lecture hours

Magnetic Properties of Matter: Magnetization vector (M). Magnetic Intensity (H). Magnetic Susceptibility and permeability. Relation between B, H, M. Ferromagnetism. B-H curve and hysteresis.

Electrodynamics

Electromagnetic Induction: Faraday's Law. Lenz's Law. Self-Inductance and Mutual Inductance. Reciprocity Theorem. Energy stored in a Magnetic Field. Introduction to Maxwell's Equations. Charge Conservation and Displacement current.

Learning Experience

The learning experience in this course is designed to be interactive and hands-on, blending theoretical knowledge with practical applications. Students will engage with complex concepts through a combination of lectures, problem-solving sessions, and laboratory experiments. The

course encourages active participation, critical thinking, and collaboration, providing opportunities for students to apply what they learn in real-world contexts. Through assignments, group discussions, and projects, students will not only deepen their understanding of electrostatics and magnetostatics but also develop skills in analytical reasoning and scientific inquiry, preparing them for future academic and professional challenges.

Textbooks

1. Physics for Scientists and Engineers (6th Ed.), Raymond A. Serway and John W. Jewett, Thomson Brooks (2004).

2. Engineering Physics Theory and Practical, A. K. Katiyar and C. K. Pandey, Wiley (2015)

Suggested Readings

1. Introduction to Electrodynamics (3rd Indian reprint), D.J. Griffiths, Pearson Education (2003).

2. Electricity, Magnetism & Electromagnetic Theory, S. Mahajan and Choudhury, 2012, Tata McGraw.

Open Educational Resources (OER)

- o https://www.youtube.com/watch?v=dGYCwt2Nqas
- o https://www.youtube.com/watch?v=qjsZTBZd1Ms
- o https://www.youtube.com/watch?v=KGTZPTnZBFE
- o https://www.youtube.com/watch?v=hJD8ywGrXks&vl=en

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER I			
SCPH151	Electricity and Magnetism Lab L T P		С
Version1.0	0 0 4 2		2
Category of Course	Major Lab		
Total Contact Hours	30		
Pre-Requisites/ Co-	Basics of Electricity and Magnetism		
Requisites			

Course Perspective

This course offers hands-on experiments designed to deepen understanding of fundamental principles in physics and electronics. Students will explore a variety of topics, such as determining the frequency of AC mains with a sonometer, measuring unknown capacitance using De-Sauty's bridge, and investigating magnetic fields with Stewart and Gee's apparatus. Additionally, the course covers advanced concepts like calculating the band gap of semiconductors using the four-probe method, determining the Hall coefficient, and measuring high resistance by substitution. Experiments on hysteresis loss using a CRO, resistance in series and parallel circuits, and verification of Ohm's Law further reinforce key theoretical concepts, providing students with a comprehensive and practical learning experience.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Observing and identifying the experimental procedures and identify key variables, equipment, and phenomena relevant to the scientific principles being studied.

CO2: Imitating replicate established experimental techniques and protocols accurately to achieve reliable and reproducible results in laboratory settings.

CO3: Practicing the theoretical knowledge by performing hands-on experiments, honing their technical skills, and demonstrating proficiency in using laboratory instruments and equipment.

Course Contents

- 1. To find the frequency of A. C. mains by using a sonometer.
- 2. To find the capacitance of unknown condensers by De Sauty's bridge
- 3. To study the variation of magnetic field with distance and to find the radius of coil by Stewart and Gee's Apparatus.
- 4. To find the band gap of intrinsic semi-conductor using four-probe method.
- 5. To find the value of Hall coefficient.
- 6. To measure high resistance by substitution method.
- 7. To determine hysteresis loss by CRO.
- 8. Resistance In Series and Parallel Apparatus
- 9. To Verify the Ohm's Law for the given circuit

Learning Experience

The learning experience in this course is designed to be both engaging and transformative, offering students a blend of theoretical knowledge and practical application. Through handson experiments and interactive demonstrations, students gain valuable insights into fundamental scientific principles and develop essential technical skills. The course encourages active participation, critical thinking, and problem-solving, fostering a deeper understanding of complex concepts. By observing, replicating, and adapting experimental techniques, students not only grasp the core content but also enhance their ability to apply knowledge in real-world scenarios, ultimately preparing them for future academic and professional endeavours.

Textbooks

- 1. Laboratory Manual Physics for class XII Published by NCERT
- 2. B. Sc Practical Physics by Geeta Sanon
- 3. B. Sc Practical Physics by CL Arora

Suggested Readings

- 1. Solid State Physics: An Introduction" by Philip Hofmann
- 2. Semiconductor Physics and Devices: Basic Principles" by Donald A. Neamen

- 3. Practical Electronics for Inventors" by Paul Scherz and Simon Monk
- 4. Electricity and Magnetism" by Edward M. Purcell and David J. Morin

Open Educational Resources (OER)

- 1. <u>https://en.wikipedia.org/wiki/Sonometer</u>
- 2. <u>http://www.indiastudychannel.com/resources/141412-How-to-measure-the-AC-source-frequency-using-sonometer-and-electromagnet.aspx</u>
- 3. http://www.schoolphysics.co.uk/age16-19/Sound/text/Sonometer /index.html
- 4. https://www.youtube.com/watch?v=F6SlpU0XkPo
- 5. https://www.youtube.com/watch?v=F0LosOk3a1g

Evaluation Scheme

Evaluation components	Weightage
Internal marks (practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical)	50 Marks
End Term Examination	

It is compulsory for a student to secure 40% marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER I					
VAC151	Environmental Studies + Disaaster Management	L	Τ	Р	C
Version 2.0		2	0	0	2
Category of Course	Value Added Course VAC-I				
Total Contact Hours	30				
Pre-Requisites/	Basics of Environment				

	Co-Requisites	isites		
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Course Perspective: This course on Environmental Sciences and Disaster Management provides a comprehensive understanding of environmental issues and disaster preparedness, crucial for students pursuing careers in environmental science, public policy, and emergency management. The curriculum covers critical topics such as land and water resources, pollution control, environmental policies, and disaster management strategies. Students will gain practical skills in assessing environmental impacts, managing resources sustainably, and preparing for and responding to various types of disasters. This knowledge is directly applicable in real-world scenarios, such as developing effective pollution control measures, implementing sustainable practices, and enhancing disaster response strategies. By integrating case studies and practical applications, the course equips students with the tools needed to address pressing environmental challenges and contribute to resilience and sustainability efforts globally.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding the types, causes, and effects of environmental pollution, and the key environmental laws in India.

CO2: Applying knowledge of environmental laws and policies and utilize the concepts of disaster preparedness,

CO3: Analysing the effects of human activities on the environment

CO4: Evaluating the effectiveness of existing environmental laws, governance frameworks, and disaster management practices in India

CO5: Creating innovative solutions for sustainable development and disaster management plans.

Course Content

Unit A: Environment and Natural Resources

Multidisciplinary nature of environmental sciences; Scope and importance; Need for public awareness.

8 Lectures

Land resources; land use change; Land degradation, soil erosion and desertification.

Deforestation: Causes and impacts due to mining, dam building on environment, forests, biodiversity and tribal populations.

Water: Use and over-exploitation of surface and ground water, floods, droughts, conflicts over water (international & inter-state).

Energy resources: Renewable and non- renewable energy sources, use of alternate energy sources, growing energy needs, case studies. Carbon Footprints.

Unit B: Environmental Pollution and Environmental Policies Lectures

Environmental pollution: types, causes, effects and controls; Air, water, soil and noise pollution nuclear hazards and human health risks; Solid waste management: Control measures of urban and industrial waste; Pollution case studies.

Sustainability and sustainable development; Climate change, global warming, ozone layer depletion, acid rain and impacts on human communities and agriculture; Environment Laws: Environment Protection Act; Air (Prevention & Control of Pollution) Act; Water (Prevention and control of Pollution) Act; wildlife Protection Act; Forest Conservation Act; Nature reserves, tribal populations and rights, and human wildlife conflicts in Indian context. Fundamentals and Application of ESG (Environment Social Governance).

> Unit C: Introduction to Disasters

Lectures

Concept and definitions- Disaster, Hazard, vulnerability, resilience, risks. Different Types of Disaster: Causes, effects and practical examples for all disasters. Natural Disaster: such as Flood, Cyclone, Earthquakes, Landslides etc. Man-made Disaster: such as Fire, Industrial Pollution, Nuclear Disaster, Biological Disasters, Accidents (Air, Sea, Rail & Road), Structural failures (Building and Bridge), War & Terrorism etc.

Unit D: Disaster Management

Disaster Preparedness Plan, Prediction, Early Warnings and Safety Measures of Disaster, Role of Government, International and NGO Bodies in Disaster Preparedness. Reconstruction and Rehabilitation, Post Disaster effects and Remedial Measures Disaster Management Act, 2005: Disaster management framework in India before and after Disaster Management Act, 2005, Applications of AI and ML in Disaster Management and risk predictions.

7 Lectures

8

7

Learning Experience

This comprehensive course integrates lectures, interactive discussions, hands-on activities, and field experiences to provide a deep understanding of environmental science, pollution, disaster types, and management strategies.

Instruction Methods:

• Lectures: Core concepts will be presented through multimedia and problem-solving sessions.

• Interactive Sessions: Includes Q&A segments, live quizzes, and discussions to enhance engagement and understanding.

• Technology Use: The Learning Management System (LMS) will provide resources, recorded lectures, and discussion forums.

Activities:

• Case Studies & Group Work: Students will analyse real-world problems related to environmental and disaster issues, working collaboratively on projects.

• Field Visits: Real-world observation of environmental and disaster management practices.

Assessments:

• Formative: Ongoing feedback through quizzes and online discussions.

• **Summative:** Exams, presentations, and case study reports to evaluate overall understanding and application of course content.

Support:

Instructors will offer additional guidance and feedback, and peer collaboration will be encouraged to support student learning and achievement of course outcomes.

Textbooks

1. Content building programme (CBP) book on Disaster Management, Forum AS.

2. Kaushik and Kaushik, Environmental Studies, New Age International Publishers (P) Ltd. New Delhi.

Suggested Readings

1. A. K. De, Environmental Chemistry, New Age International Publishers (P) Ltd. New Delhi.

2. S.E. Manahan, Environmental Chemistry, CRC Press.

3. S.S Dara and D.D. Mishra, Environmental Chemistry and Pollution Control, S.Chand& Company Ltd, New Delhi.

Open Educational Resources (OER)

- 1. OpenStax
- 2. <u>Khan Academy</u>
- 3. MIT OpenCourseWare

- 4. <u>Coursera</u> and \underline{edX}
- 5. <u>YouTube</u>
- 6. <u>Wikibooks</u>
- 7. OER Commons
- 8. <u>NOAA Education</u>
- 9. <u>UNEP Education</u>
- 10. TED Talks

Evaluation Scheme:

Evaluation components		Weightage
Internal marks (Theory)		30 Marks
I. Continuous assessment (30 marks)		
All the components to be evenly spaced		
Project/quizzes/assignment and	essays/presentation/	
II. Internal marks (Theory):		20 Marks
Mid Term Examination		
III. External Marks (Theory):		50 Marks
End Term Examination		

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER I					
SEC015	Basic Instrumentation SkillsLTP		С		
Version 1.0		3	0	0	3
Category of Course	Skill Enhancement Course				
Total Contact Hours	30				
Pre-Requisites/ Co-	Basics of Electronics				
Requisites					

Course Perspective The "Basic Instrumentation Skills" course is designed to equip students with essential skills in using and understanding measurement instruments. It provides a foundational knowledge of how these instruments work, their key specifications, and their applications in real-world scenarios. The course emphasizes the importance of precision and accuracy in measurements, critical for any scientific or engineering work. By focusing on practical skills, this course prepares students for advanced studies and technical careers, where accurate measurement and data analysis are crucial.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding concepts like accuracy, precision, and measurement errors.

CO 2: Using multimeters and voltmeters, understanding their specifications.

CO 3: Analyzing the advantages of electronic voltmeters over conventional ones.

CO 4: Evaluating instrument performance based on specifications for specific tasks.

Course Content

UNIT-I Basic of Measurement

Basic of Measurement: Instruments accuracy, precision, sensitivity, resolution range etc. Errors in measurements and loading effects. Multimeter: Specifications of a multimeter and their significance.

Unit-II Electronic Voltmeter

Electronic Voltmeter: Advantage over conventional multimeter for voltage measurement with respect to input impedance and sensitivity. Principles of voltage, measurement (block diagram only). Specifications of an electronic Voltmeter/ Multimeter and their significance. AC millivoltmeter: Type of AC millivoltmeters: Amplifier- rectifier, and rectifier- amplifier. Block diagram ac millivoltmeter, specifications and their significance.

UNIT-III Cathode Ray Oscilloscope

Cathode Ray Oscilloscope: Block diagram of basic CRO. Construction of CRT, Electron gun, electrostatic focusing and acceleration (Explanation only– no mathematical treatment), brief discussion on screen phosphor, visual persistence & chemical composition. Time base

10 Lectures

12 Lectures

8 Lectures

operation, synchronization. Front panel controls. Specifications of a CRO and their significance.

Use of CRO for the measurement of voltage (dc and ac frequency, time period. Special features of dual trace, introduction to digital oscilloscope, probes. Digital storage Oscilloscope: Block diagram and principle of working.

Learning Experience: The learning experience in this course will be hands-on and practical, focusing on the use and understanding of various measurement instruments. Students will engage in interactive sessions, working directly with multimeters, voltmeters, and oscilloscopes to gain real-world skills. Through guided lab activities, demonstrations, and problem-solving exercises, students will develop a deep understanding of how to operate and interpret readings from these instruments, preparing them for practical applications in technical and engineering fields.

Textbooks:

- 1. Textbook in Electrical Technology B L Theraja S Chand and Co.
- 2. Digital Circuits and systems, Venugopal, 2011, Tata McGraw Hill.

Suggested Readings:

- 1. Logic circuit design, Shimon P. Vingron, 2012, Springer.
- 2. Digital Electronics, Subrata Ghoshal, 2012, Cengage Learning.
- Electronic Devices and circuits, S. Salivahanan & N. S.Kumar, 3rd Ed., 2012, Tata Mc-Graw Hill.

Open Educational Resources (OER)

- 1. https://www.youtube.com/watch?v=kxKOU3hmNug
- 2. <u>https://www.youtube.com/watch?app=desktop&v=dGtwItnYYqk</u>

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Theory)			30 Marks
I. continuous assessment (30 i	narks)		
All the components to be even	y spaced		
Project/quizzes/assignment II. internal marks (Theory):	and	essays/presentation/	20 Marks
Mid Term Examination III. External Marks (Theory)			50 Marks
III. External Marks (Theory) End Term Examination	:		50 Marks

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER II					
SCPH102	Mathematical Physics-II	L	Т	Р	С
Version 1.0		4	0	0	4
Category of Course	Major	1	1	1	
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Mathematical Physics-I				

Course Perspective This course aims to demonstrate the use of mathematical techniques in solving problems in Physics and to provide a deeper understanding of the mathematics underpinning theoretical physics. The course is intended to develop the theory of errors, Fourier series, special functions and partial differential equations. Emphasis will be on illustrative examples from Physics and Engineering

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding Fourier series theory, including expansions and Parseval Identity.CO2: Applying the Frobenius method and special functions (Legendre, Bessel, etc.) and

separation of variables method to solve partial differential equations for various applications.

CO3: Analyzing special integrals like Beta and Gamma functions and their applications.

CO4: Evaluating errors, including systematic and random errors, and use least-squares fitting.

Course Content

UNIT-I Fourier Series

15 Lecture Hours

Periodic functions. Orthogonality of sine and cosine functions, Dirichlet Conditions (Statement only). Expansion of periodic functions in a series of sine and cosine functions and determination of Fourier coefficients. Complex representation of Fourier series. Expansion of functions with arbitrary period. Expansion of non-periodic functions over an interval. Even and odd functions and their Fourier expansions. Application. Summing of Infinite Series. Term-by-Term differentiation and integration of Fourier Series. Parseval Identity.

UNIT-I Frobenius Method and Special Functions 15 Lecture Hours

of Singular Points Second Order Linear Differential Equations and their importance. Frobenius method and its applications to differential equations. Legendre, Bessel, Hermite and Laguerre Differential Properties Equations. of Legendre Polynomials: Rodrigues Formula, Generating Function, Orthogonality. Simple recurrence relations. Expansion of function in a series of Legendre Polynomials. Bessel Functions of the First Kind: Generating Function, simple recurrence relations. Zeros of Bessel Functions (Jo(x) and J1(x)) and Orthogonality.

UNIT-III Integrals & Errors

15 Lecture Hours

Some Special Integrals: Beta and Gamma Functions and Relation between them. Expression of Integrals in terms of Gamma Functions. Error Function (Probability Integral).

Theory of Errors: Systematic and Random Errors. Propagation of Errors. Normal Law of Errors. Standard and Probable Error. Least-squares fit. Error on the slope and intercept of a fitted line.

UNIT-IVPartialDifferentialEquations15 Lecture Hours

Solutions to partial differential equations, using separation of variables: Laplace's Equation in problems of rectangular, cylindrical and spherical symmetry. Wave equation and its solution

for vibrational modes of a stretched string, rectangular and circular membranes. Diffusion Equation.

Learning experience

The learning experience in this course will involve a combination of theoretical understanding and practical application. Students will engage with mathematical techniques through lectures, problem-solving sessions, and hands-on exercises. They will apply Fourier series, special functions, and special integrals to real-world problems, and use software tools for error analysis and fitting. Practical exercises and assignments will help them understand and apply partial differential equations to physical scenarios, enhancing their problem-solving skills and mathematical proficiency.

Textbooks:

- 1. Mathematical Physics, H.K. Dass and R. Verma, S. Chand & Company.
- 2. Advanced Engineering Mathematics, Erwin Kreyszig, 2008, Wiley India.

Suggested Readings:

- 1. Mathematical Methods for Physicists: Arfken, Weber, 2005, Harris, Elsevier.
- 2. Fourier Analysis by M.R. Spiegel, 2004, Tata McGraw-Hill.
- 3. Mathematics for Physicists, Susan M. Lea, 2004, Thomson Brooks/Cole.
- 4. Differential Equations, George F. Simmons, 2006, Tata McGraw-Hill.

Open Educational Resources (OER)

- 1. https://math.mit.edu/~gs/cse/websections/cse41.pdf
- 2. <u>https://www.researchgate.net/publication/250156802_Notes_on_Special_Functions</u>
- 3. <u>https://ocw.mit.edu/courses/18-152-introduction-to-partial-differential-equations-fall-</u> 2011/resources/mit18_152f11_lec_01/

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Theory)			30 Marks
I. continuous assessment (30 ma	arks)		
All the components to be evenly s	spaced		
Project/quizzes/assignment II. internal marks (Theory):	and	essays/presentation/	20 Marks
Mid Term Examination III. External Marks (Theory):			50 Marks
End Term Examination			

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER II					
SCPH104	Mechanics	L	Τ	Р	C
		4	0	0	4
Version1.0					
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Basic Physics and Mathematics				

Course Perspective This course offers a comprehensive exploration of classical mechanics and relativity, covering fundamental dynamics, rotational motion, and harmonic oscillations. Students will gain a deep understanding of concepts such as energy conservation, rotational dynamics, and relativistic effects, and will apply these principles to solve complex physical problems. Emphasis is placed on analytical skills and practical applications, preparing students to tackle advanced topics in physics and engineering.

Course Outcomes

Upon completion of the course the learner will be able:

CO1: Understanding the concepts of inertial reference frames, Newton's Laws of Motion, different types of forces, energy interactions, collisions, principles of conservation of momentum and relativity.

CO2: Applying principles of translation and rotational dynamics to solve problems involving force, energy, momentum, torque, angular momentum, moment of inertia, elasticity and relativity.

CO3: Analyzing the motion of projectiles, rockets, systems of particles, centre of mass, relativistic effects, including time dilation, Lorentz contraction, and mass-energy equivalence.

CO4: Evaluating gravitational interactions, gravitational potential energy and central force motions, kinetic energy, potential energy, and total energy of an oscillating system to explain the concepts related to GPS and weightlessness, damping and resonance in real-world scenarios.

Course Content

UNIT-I Fundamentals of Dynamics

15 Lecture Hours

Reference frames. Inertial frames; Review of Newton's Laws of Motion. Galilean transformations; Galilean invariance. Momentum of variable-mass system: motion of rocket. Motion of a projectile in Uniform gravitational field Dynamics of a system of particles. Centre of Mass. Principle of conservation of momentum. Impulse.

Work and Kinetic Energy Theorem. Conservative and non-conservative forces. Potential Energy. Energy diagram. Stable and unstable equilibrium. Elastic potential energy. Force as gradient of potential energy. Work & Potential energy. Work done by nonconservative forces. Law of Conservation of Energy.

Elastic and inelastic collisions between particles. Centre of Mass and Laboratory frames.

UNIT-II Rotational Dynamics and Gravitational motion 15 Lecture Hours

Angular momentum of a particle and system of particles. Torque. Principle of conservation of angular momentum. Rotation about a fixed axis. Moment of Inertia. Calculation of moment of inertia for rectangular, cylindrical and spherical bodies. Kinetic energy of rotation. Motion involves both translation and rotation.

Relation between Elastic constants. Twisting torque on a Cylinder or Wire.

Kinematics of Moving Fluids: Poiseuille's Equation for Flow of a Liquid through a Capillary Tube.

Gravitation and Central Force Motion: Law of gravitation. Gravitational potential energy. Inertial and gravitational mass. Potential and field due to spherical shell and solid sphere.

Motion of a particle under a central force field. Two-body problem and its reduction to onebody problem and its solution. The energy equation and energy diagram. Kepler's Laws. Satellite in circular orbit and applications. Geosynchronous orbits. Weightlessness. Basic idea of global positioning system (GPS).

UNIT-III Oscillatory motion and Non-Inertial Systems 15 Lecture Hours

Simple Harmonic Oscillations. Differential equation of SHM and its solution. Kinetic energy, potential energy, total energy and their time-average values. Damped oscillation. Forced oscillations: Transient and steady states; Resonance, sharpness of resonance; power dissipation and Quality Factor.

Non-Inertial Systems: Non-inertial frames and fictitious forces. Uniformly rotating frame. Laws of Physics in rotating coordinate systems. Centrifugal force. Coriolis force and its applications. Components of Velocity and Acceleration in Cylindrical and Spherical Coordinate Systems.

UNIT-IV Special Theory of Relativity

15 Lecture Hours

Michelson-Morley Experiment and its outcome. Postulates of Special Theory of Relativity. Lorentz Transformations. Simultaneity and order of events. Lorentz contraction. Time dilation. Relativistic transformation of velocity, frequency and wave number. Relativistic addition of velocities. Variation of mass with velocity. Massless Particles. Mass energy Equivalence. Relativistic Doppler effect. Relativistic Kinematics. Transformation of Energy and Momentum.

Learning Experience

The course will be delivered through a combination of lectures, interactive demonstrations, and multimedia presentations to facilitate a comprehensive understanding of mechanics. Group discussions, and problem-solving workshops will provide practical applications of theoretical concepts. Regular quizzes, assignments, and project-based assessments will evaluate their grasp of the material. Additionally, support will be available through office hours, peer collaborations, and online forums, promoting a collaborative and resourceful learning environment.

Textbooks

1. Mechanics, D.S. Mathur, S. Chand and Company Limited, 2000

Suggested Readings

- 1. An introduction to mechanics, D. Kleppner, R.J. Kolenkow, 1973, McGraw-Hill.
- 2. Mechanics, Berkeley Physics, vol.1, C.Kittel, W.Knight, et.al. 2007, Tata McGraw-Hill.

Open Educational Resources (OER)

- <u>https://www.fisica.net/mecanicaclassica/introduction_to_statics_and_dynamics_by_rudra</u> _pratap.pdfhttps://www.fisica.net/mecanicaclassica/introduction_to_statics_and_dynamic s_by_rudra_pratap.pdf
- 2. <u>https://www.youtube.com/watch?v=W8_Vr7zzA84</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER II					
SCPH152	MECHANICS LAB	L	Т	Р	С
		0	0	4	2

Version 1.0	
Category of Course	Major Lab
Total Contact Hours	30
Pre-Requisites/ Co- Requisites	Mechanics

Course Perspective The course will combine practical experimentation with theoretical analysis to provide a comprehensive understanding of fundamental physics concepts. Students will conduct hands-on experiments to measure physical quantities, determine mechanical properties, and analyze error and precision in data. The course will also incorporate data analysis techniques, including statistical error analysis and result interpretation, to bridge the gap between theoretical knowledge and practical application. Through detailed experimental reports and presentations, students will develop critical thinking and problem-solving skills, enhancing their ability to apply theoretical concepts to real-world scenarios.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Observing and measuring length or diameter using vernier callipers, screw gauges, and traveling microscopes.

CO2: Identifying, analyzing, and quantifying random errors in experimental observations, applying statistical methods to assess the reliability of their data

CO3: Practicing setting up experiments, including spring motion experiments and flywheel measurements to measure physical constants such as the acceleration due to gravity (g), the spring constant, and the modulus of rigidity.

CO4: Evaluating the mechanical properties of materials, such as Young's Modulus, coefficient of viscosity, and elastic constants, through experiments.

Course Content

At least 08 experiments from the following:

1. Measurements of length (or diameter) using vernier caliper, screw gauge and travelling microscope.

2. To study random errors in observations.

3. To determine the height of a building using a Sextant.

4. To study the Motion of Spring and calculate (a) Spring constant, (b) g and (c) Modulus of rigidity.

5. To determine the Moment of Inertia of a Flywheel.

6. To determine g and velocity for a freely falling body using Digital Timing Technique

7. To determine Coefficient of Viscosity of water by Capillary Flow Method (Poiseuille's method).

8. To determine the Young's Modulus of a Wire by Optical Lever Method.

9. To determine the Modulus of Rigidity of a Wire by Maxwell's needle.

10. To determine the elastic Constants of a wire by Searle's method.

11. To determine the value of g using Bar Pendulum.

12. To determine the value of g using Kater's Pendulum.

Textbooks

1. A Text Book of Practical Physics, I.Prakash & Ramakrishna, 11th Edn, 2011, Kitab Mahal

Suggested Readings

- Engineering Practical Physics, S.Panigrahi & B.Mallick, 2015, Cengage Learning India Pvt. Ltd.
- 2. Practical Physics, G.L. Squires, 2015, 4th Edition, Cambridge University Press.

Open Educational Resources (OER)

1. phet.colorado.edu

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER II					
SEC016	Physics Workshop Skill	L	Т	Р	С
		3	0	0	3
Version 1.0					
Category of Course	Skill Enhancement Course				
Total Contact Hours	30				
Pre-Requisites/ Co- Requisites	Basic Physics				

Course Perspective This course provides an in-depth exploration of fundamental measurement techniques, workshop practices, and basic electrical and electronic skills. Students will gain practical experience in using various measuring instruments and tools, understand different manufacturing methods, and develop the ability to work with electrical circuits. Emphasis is placed on both theoretical knowledge and hands-on applications, ensuring students are well-prepared to apply these skills in real-world scenarios.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Understanding various measurement tools (e.g., meter scale, Vernier calliper, screw gauge) and their uses in practical scenarios.

CO2: Applying measurement techniques to accurately measure the dimensions of a solid block, the volume of a cylindrical beaker, the diameter of a thin wire, and the thickness of a metal sheet using appropriate tools.

CO3: Analyzing various manufacturing methods (e.g., casting, welding) and differentiate between types of welding joints and defects, identifying appropriate methods for specific materials and processes.

CO4: Evaluating the performance of electrical and electronic circuits using instruments such as multimeters and oscilloscopes and assess the quality of soldered connections and circuit functionality in terms of design and operation.

Course Content

UNIT-I Units & Measurement

and CGS. Familiarization with meter s

Introduction: Measuring units. conversion to SI and CGS. Familiarization with meter scale, Vernier calliper, Screw gauge and their utility. Measure the dimensions of a solid block, volume of cylindrical beaker/glass, diameter of a thin wire, thickness of metal sheet, etc.

UNIT-II Mechanical Skill

10 Lecture Hours

10 Lecture Hours

Concept of workshop practice. Overview of manufacturing methods: casting, foundry, machining, forming and welding. Types of welding joints and welding defects. Common materials used for manufacturing like steel, copper, iron, metal sheets, composites and alloy, wood. Concept of machine processing, introduction to common machine tools like lathe, shaper, drilling, milling and surface machines. Cutting tools, lubricating oils.

UNIT-III Electrical and Electronic Skill 10 Lecture Hours

Use of Multimeter. Soldering of electrical circuits having discrete components (R, L, C, diode) and ICs on PCB. Operation of oscilloscope. Making regulated power supply. Timer circuit, electronic switch using transistor and relay.

Learning Experience

The course will be delivered through a blend of lectures, hands-on demonstrations, and practical laboratory sessions to ensure a thorough understanding of measurement techniques,

workshop practices, and basic electronics. Interactive workshops will allow students to apply theoretical concepts in real-world scenarios, such as measuring dimensions, working with manufacturing methods, and constructing electronic circuits. Group projects and problemsolving activities will encourage collaborative learning and practical application of skills. Regular assessments through quizzes, practical exercises, and project evaluations will gauge students' comprehension and ability to apply their knowledge. Support will be available through scheduled office hours, peer study groups, and online resources, fostering an engaging and supportive learning environment.

Textbooks:

- 1. A textbook in Electrical Technology B L Theraja S. Chand and Company.
- 2. Performance and design of AC machines M.G. Say, ELBS Edn.
- 3. Mechanical workshop practice, K.C. John, 2010, PHI Learning Pvt. Ltd

Suggested Readings:

1. Workshop Processes, Practices and Materials, Bruce J Black 2005, 3rd Edn., Editor Newnes

2. New Engineering Technology, Lawrence Smyth/Liam Hennessy, The Educational Company of Ireland [ISBN: 0861674480]

Open Educational Resources (OER)

1. <u>https://www.youtube.com/watch?v=i9ixyz8y2gA</u>

2. <u>https://www.youtube.com/watch?v=ynnJVpdhX3o</u>

3.<u>https://www.youtube.com/watch?v=mc979OhitAg&list=PLWv9VM947MKi_7yJ0_FC</u> <u>fzTBXpQU-Qd3K</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	

II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER III					
SCPH201	Mathematical Physics-III	L	Т	Р	С
		4	0	0	4
Version 1.0					
Category of Course	Major			•	
Total Contact Hours	60				
Pre-Requisites/ Co-	Mathematical Physics-I, II				
Requisites					

Course Perspective This course is crucial for developing advanced mathematical skills needed in engineering and science, focusing on complex analysis, Fourier, and Laplace transforms. It enhances problem-solving abilities essential for designing and analyzing systems in real-world applications like signal processing and control systems. Mastery of these techniques prepares students for technical careers and advanced studies, making them adept at addressing complex challenges in their fields.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Understanding key concepts and formulas related to complex analysis, Fourier transforms, and Laplace transforms.

CO2: Applying Complex Algebra, Fourier and Laplace transforms to solve interdisciplinary problems in engineering and science.

CO3: Analysing the fundamental properties of complex functions, Fourier and Laplace transforms and explain their importance in solving problems.

CO4: Evaluating which transform is best suited for solving integrals and differential equations associated with a specific physical system.

Course Content

UNIT-I Complex Analysis

15 Lecture Hours

Brief Revision of Complex Numbers and their Graphical Representation. Euler's formula, De Moivre's theorem, Roots of Complex Numbers. Functions of Complex Variables. Analyticity and Cauchy-Riemann Conditions. Examples of analytic functions. Singular functions: poles and branch points, order of singularity, branch cuts.

UNIT-II Functions of a Complex variable and Integrals Transforms 20 Lecture Hours

Integration of a function of a complex variable. Cauchy's Inequality. Cauchy's Integral formula. Simply and multiply connected region. Laurent and Taylor's expansion. Residues and Residue Theorem. Application in solving Definite Integrals.

Integrals Transforms: Fourier Integral theorem. Fourier Transform. Examples. Fourier transform of trigonometric, Gaussian, finite wave train & other functions. Representation of Dirac delta function as a Fourier Integral.

UNIT-III Fourier transform

Fourier transform of derivatives, Inverse Fourier transform, Convolution theorem. Properties of Fourier transforms (translation, change of scale, complex conjugation, etc.). Three dimensional Fourier transforms with examples. Application of Fourier Transforms to differential equations: One dimensional Wave and Diffusion/Heat Flow Equations.

UNIT-IV Laplace Transform

Laplace Transform (LT) of Elementary functions. Properties of LTs: Change of Scale Theorem, Shifting Theorem. LTs of 1st and 2nd order Derivatives and Integrals of Functions, Derivatives and Integrals of LTs. LT of Unit Step function, Dirac Delta function, Periodic Functions. Convolution Theorem. Inverse LT. Application of Laplace Transforms to 2nd order Differential

10 Lecture Hours

15 Lecture Hours

Equations: Damped Harmonic Oscillator, Simple Electrical Circuits, Coupled differential equations of 1st order. Solution of heat flow along infinite bar using Laplace transform.

Learning Experience

The course will be conducted through a blend of lectures, PowerPoint presentations, and interactive discussions to ensure an engaging and practical experience. Students will use mathematical software and online resources to explore complex analysis, Fourier transforms, and Laplace transforms. They will participate in group projects, case studies, and problem-solving sessions to apply theoretical concepts to real-world problems. Regular assignments and exams will assess their understanding and application skills. Support will be provided through office hours, peer reviews, and collaborative activities, fostering a supportive learning environment where students are encouraged to seek help and collaborate.

Textbooks

1. Mathematical Physics, H.K. Dass and R. Verma, S. Chand & Company.

Suggested Readings

1. Mathematical Methods for Physics and Engineers, K.F Riley, M.P. Hobson and S. J. Bence, 3rd ed., 2006, Cambridge University Press.

2. Mathematics for Physicists, P. Dennery and A.Krzywicki, 1967, Dover Publications.

Open Educational Resources (OER)

- 1. <u>https://youtu.be/ysVcAYo7UPI</u>
- 2. https://youtu.be/QiwfF83NWNA
- 3. https://www.coursera.org/lecture/complex-analysis/complex-functions

<u>koxdh?utm_source=link&utm_medium=page_share&utm_content=vlp&utm_campaign=top_</u> <u>button</u>

- 4. https://youtu.be/OiNh2DswFt4
- 5. https://youtu.be/33TYoybjqPg

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER III					
SCPH203	Solid State Physics	L	Т	Р	С
Version 1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co	Crystallography and Mathematical Physics				
Requisites					

Course Perspective

This course is intended to cover most of the basic topics of solid-state including Crystal Structure, Space groups, Packing fraction, Miller indices, defects in crystal, X-Ray Diffraction, Diamagnetism and Para-magnetism, Superconductivity. This course gives crystallographic knowledge and enables students to analyse the solid structures, which will also be beneficial for students in research field. The Solid-State Physics course is crucial for understanding the behavior of materials at the atomic level, a key aspect of both fundamental physics. It equips students with essential knowledge about crystal structures, electronic properties, and material design, which is directly applicable to careers in electronics, materials science, and nanotechnology. Mastery of these concepts enables students to innovate in fields like

semiconductor technology and renewable energy, driving advancements in modern technology and research.

Course Outcomes

Upon completion of the solid-state physics course, the learner will be:

CO1: Understanding key concepts and principles of solid-state physics.

CO2: Applying theoretical models and principles to solve problems related to materials and their properties.

CO3: Analyzing experimental data and theoretical predictions to draw conclusions about material behaviours.

CO4: Evaluating different approaches and theories to understand their strengths and limitations in explaining solid state phenomena.

Course Content

Unit No I: Crystal Structure

Amorphous and crystalline materials, lattice translation vectors, lattice with a basis –unit cell, types of lattices symmetry elements, inter planer spacing, packing fraction, Miller Indices, Bonding in solids- ionic bond. covalent bond, metallic bonds, hydrogen bonding, van Der Waals bond, crystal defects, point defects, line defects, Burger's vector, surface imperfections.

Unit No II: Magnetic Properties of Matter

Classification of Magnetic materials - Dia-, Para-, Ferri- and Ferromagnetic Materials, Classical Langevin Theory of dia magnetic materials, Curie's law, Weiss's Theory of Ferromagnetism and Ferromagnetic domains. Discussion of B-H Curve, Hysteresis & Energy Loss, anti-ferromagnetism, ferrimagnetism.

Unit No III: Dielectrics Properties of Materials

Types of Polarization, Local Electric Field at an Atom, static, Dielectric Constant, Electric Susceptibility, Polarizability, Classical Theory of Electric Polarizability, three vectors, Clausius-Mosotti Equation. Variation of dielectric polarization with temperature and frequency, pizo-pyro and ferroelectricity properties, domain theory of ferroelectricity, Complex Dielectric Constant, dielectrics in alternating fields, relaxation in dielectrics, absorption and losses, dielectric breakdowns.

No. of Hours 15

No. of Hours 15

No. of Hours 15

Unit No IV: Superconductivity

No. of Hours 15

Critical Temperature. Critical magnetic field. Meissner effect. Type I and type II Superconductors, London's Equation and Penetration Depth. Isotope effect. Idea of BCS theory (No derivation)

Learning Experience

The Solid-State Physics course will be delivered through lectures with multimedia tools, handson lab sessions, and virtual simulations. Students will engage in group projects, case studies, and problem-solving assignments to apply their knowledge. Technology such as simulation software will enhance learning, and regular quizzes, exams, and lab reports will assess progress. Support will be provided through office hours, peer reviews, and online discussion forums, promoting collaboration and active learning.

Textbooks

1. S O Pillai, Solid State Physics (New Age International Limited).

2. B.D. Cullity and C. D. Graham, Introduction to Magnetic Materials (John Wiley and Sons, Inc.)

Suggested Readings

- 1. B.D. Cullity, Elements of X-Ray Diffraction (Addison-Wesley Metallurgy Series).
- 2. Charles Kittel, Introduction to Solid State Physics (John Wiley and Sons, Inc.).
- 3. N. W. Ascroft and N. D. Mermin, Solid State Physics (Harcourt Asia, Singapore).
- 4. M. Ali Omar, Elementary solid-state physics: principles and applications (Pearson Education)

Open Educational Resources (OER)

- 1. https://youtu.be/5h5gXoFyo64
- 2. https://youtu.be/ax rNTSI7ac
- 3. https://youtu.be/63cwdYXNIYE
- 4. <u>https://youtu.be/sEGLcpmIIBY</u>
- 5. <u>https://youtu.be/XrTJUAyolvE</u>
- 6. https://youtu.be/rkntp3 cZl4
- 7. https://youtu.be/WV2AexANG34

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER III					
SCPH253	Solid State Physics Lab	L	Τ	Р	С
Version 1.0		0	0	4	2
Category of Course	Major				
Total Contact Hours	30				
Pre-Requisites/ Co Requisites	Solid State Physics				

Course Perspective

This course imparts the basic knowledge of construction of materials, microscopically. The course deals with the factors and conditions which are required to determine various properties like dielectric constant, magnetic susceptibility, semiconductor properties. The course further delivers keen understanding of magnetism and its application in technology. The Solid-State Physics Laboratory course is crucial for B.Sc. (H) Physics students as it connects theoretical concepts with hands-on experience. It equips students with practical skills in experimental

techniques, data analysis, and the application of solid-state physics principles. This knowledge is vital for careers in research, materials science, and engineering. The course also prepares students for real-world applications, such as developing new materials and improving electronic devices, making it a foundational component of their academic and professional development.

Course Outcomes

Upon completion of the solid-state physics course, the learner will be able:

CO1: Observing and describing experimental phenomena in solid state physics, demonstrating an ability to accurately record and interpret data.

CO2: Imitating standard experimental procedures and techniques, replicating established methods with attention to accuracy and detail (Applying).

CO3: Practising using laboratory instruments and techniques, refining their skills through repeated use and procedural application.

Course Content

List of experiments

- 1. To find the band gap of intrinsic semiconductor using four probe method.
- 2. To determine hysteresis loss using CRO.
- 3. To measure the dielectric constant of a dielectric material.
- 4. To study the variation of magnetic field with distance and to find the radius of coil.
- 5. To study V-I characteristics of PN diode.
- 6. To find the energy band gap of PN junction diode.
- 7. To determine the value of Planck's constant by photocell.
- 8. To find the value of Hall coefficient and carrier concentration of a semiconductor.
- 9. To measure the Magnetic field strength in solenoid.

Learning Experience

The Solid-State Physics Laboratory course will be conducted through hands-on experiments, utilizing advanced technology and simulation software to enhance learning. Students will engage in group work, case studies, and real-world applications to apply theoretical concepts. Assessments will include lab reports, presentations, and practical exams. Support and feedback will be provided by the course instructor, with opportunities for additional help and peer

collaboration throughout the course. This approach ensures an experiential and participatory learning experience.

Textbooks

- 1. Solid State Physics, S.O.PIllai, New Age Publication.
- 2. Elements of X-Ray Diffraction, B.D. Cullity. Addison-Wesley Publishing Company.
- Introduction to Magnetic Materials (2nd Edition), B.D. Cullity and C.D. Graham, Wiley (2009).

Suggested Readings

- 1. Introduction to Solid State Physics, Charles Kittel, 8th Ed., 2004, Wiley India Pvt. Ltd.
- 2. Elements of Solid-State Physics, J.P. Srivastava, 2nd Ed., 2006, Prentice-Hall of India.
- 3. Introduction to Solids, Leonid V. Azaroff, 2004, Tata Mc-Graw Hill.
- 4. Solid State Physics, N.W. Ashcroft and N.D. Mermin, 1976, Cengage Learning.
- 5. Solid-state Physics, H. Ibach and H. Luth, 2009, Springer. Reference Books/Materials

Open Educational Resources (OER)

- 1. <u>https://youtu.be/KU3c5WaEwkI</u>
- 2. https://youtu.be/sSzpb6rOXx8
- 3. <u>https://youtu.be/N9wyxXo-c30</u>
- 4. <u>https://youtu.be/_vKeaPHXF9U</u>
- 5. https://youtu.be/NKYBc7u6hO4
- 6. <u>https://youtu.be/gDFGj0Iodug</u>
- 7. <u>https://youtu.be/RA_wGkUBetU</u>
- 8. <u>https://youtu.be/2VkUMnl3dPk</u>

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical):	50 Marks
End Term Examination	

SEMESTER III					
AEC001	Verbal Ability	L	Т	Р	С
Version 1.0		3	0	0	3
Category of Course	Ability Enhancement Course				
Total Contact Hours	45				
Pre-Requisites/ Co- Requisites					

Course Perspective:

The course aims to improve language proficiency in three key areas: grammar, vocabulary and identification of grammatical errors in writing. Language proficiency enables students to comprehend lectures, understand course materials and enhances students' ability to express themselves clearly and effectively. In many professions, strong language skills are a prerequisite. Whether in business, medicine, law, or science, being able to communicate fluently and accurately is essential for collaboration, negotiation, and advancement. A strong command of verbal abilities can significantly impact job interviews. It allows candidates to answer questions confidently, demonstrate their qualifications effectively and leave a positive impression on potential employers.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1 Understanding the grammar rules and word meaning (Vocabulary).

CO2 Applying grammar rules and vocabulary in different contexts & purpose

CO3 Analyzing situations/ context of communication and selecting appropriate grammar and words.

CO4 Developing sentences and paragraphs to describe and narrate a situation

Course Content

Unit Number: 1 Vocabulary Development and Application 10 hrs

Content Summary: Understanding the concept of root words, Prefix and suffix, Ways to enhance Vocabulary, Crosswords and word quizzes, Confusing words, One word substitution, Odd one out, Synonyms and Antonyms, commonly misspelt words, Idioms and Phrases

Unit Number:2 Fundamentals of Grammar and Sentence Structure 8 hrs

Content Summary: Introduction to Parts of Speech, Tenses and its 'rules, Sentences (Simple, Compound and Complex), Subject Verb Agreement, Pronoun Antecedent agreement, Phrases and Clauses

Unit Number: 3 Mastering Sentence Accuracy and Completion Skills 12hrs

Content Summary: Spot the error (grammatical errors in a sentence), Sentence Correction (Improvement of sentences based on Grammar rules), Sentence Completion, Cloze Tests

Unit Number: 4 Enhancing Sentence Structure and Reading Comprehension Skills 6 hrs

Logical Arrangement of Sentences, Comprehending passages, Contextual questions, Anagrams, Analogies

Learning Experience:

The course will be conducted using interactive and participatory methods such as vocabularybuilding activities, grammar exercises, and sentence correction drills. Students will engage in crosswords, word quizzes, and error-spotting exercises to enhance their vocabulary and grammatical accuracy. Through sentence arrangement and comprehension practice, they will improve their reading and writing skills. Real-time feedback and collaborative activities like anagrams and analogies will foster a deeper understanding of language structure and application in various contexts.

Textbooks:

- 1. Norman Lewis Word Power Made Easy
- 2. Wren & Martin High School English Grammar & Composition
- 3. R.S. Agarwal & Vikas Agarwal Quick Learning Objective General English
- 4. S.P. Bakshi Objective General English
- 5. Praxis Groups -Campus Recruitment Complete Reference

Suggested Readings:

- 1. https://www.indiabix.com/online-test/aptitude-test/
- 2. https://www.geeksforgeeks.org/aptitude-questions-and-answers/
- 3. https://www.hitbullseye.com/

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER III							
SIPH001	Evaluation Internship/Pro	of Dject	Summer	L	Τ	Р	С

Version 1.0		2	0	0	2
Category of Course	Internship				
Total Contact Hours					
Pre-Requisites/ Co	Practical Exposure				
Requisites					

Course Perspective:

In the end of Semester II, students will be asked to join research/academic organizations or industries to get hands on knowledge on the selected topics. The student will work on the assigned topic for 3-4 weeks in regular consultation with his/her assigned expert/guide. The student will write a report based on the work carried out during internship and prepare two copies to be submitted to the office of the Head of the Department duly signed by the student and the expert. The student will make a power point presentation based on the work carried out and mentioned in the report to the board of examiners appointed by the University in the third semester. The student will be evaluated based on a report and presentation.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Carrying out the extensive literature survey on the topic assigned by academicians and industry experts.

CO 2: Applying various methods and techniques applicable to their research topic to study and contribute to domain knowledge.

CO 3: Analyzing the result of the experiment carried out and present the results using data visualization methods.

CO 4: Evaluating the effectiveness of methods used and the significance of research findings.

CO 5: Writing and presenting technical reports/articles.

Learning Experience:

The internship course will be experiential through hands-on lab work, real-world research projects, and active participation in ongoing studies. Students will collaborate with faculty and

researchers, applying theoretical knowledge to experimental tasks and data analysis. Regular group discussions, progress presentations, and peer feedback will enhance collaborative learning. The course will also include reflective journaling to encourage self-assessment and growth throughout the internship.

Evaluation Scheme:

Particular	Weightage
Internal Marks: - (Punctuality, Performance, Work Ethics, Efforts and Research Output)	50 Marks
External Marks (Practical): -	50 Marks
Presentation	20
Report Writing/Dissertation	10
Viva Voce	20

SEMESTER III					
SCPH205	MODERN PHYSICS	L	T	Р	C
Version1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Electromagnetism				

Course Perspective – This course offers students a deep understanding of the principles governing atomic and subatomic particles. It builds a strong foundation in key concepts such as atomic structure, wave-particle duality, and nuclear transformations, which are essential for advanced studies and research in physics. The course enhances students' critical thinking and problem-solving skills through the exploration of historical experiments and theoretical

models. It also provides practical knowledge applicable in industries like nuclear energy, healthcare, and technology. This course is a stepping stone for those pursuing advanced studies or careers in physics, engineering, and related fields.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding key concepts in atomic and nuclear physics, including atomic structure, wave-particle duality, and nuclear reactions.

CO2: Analyzing experimental evidence supporting atomic models and theories, like Rutherford's experiment and the Bohr model.

CO3: Applying quantum mechanics and nuclear principles to solve problems related to atomic spectra, nuclear transformations, and energy production.

CO4: Evaluating the impact of atomic and nuclear phenomena on real-world applications, such as energy production and medical imaging.

Course Content

UNIT-I Structure of the Atom

Rutherford's Experiments on Scattering of α -particles; Theory of α -particle Scattering; Bohr Atom Model; Effect of Nuclear Motion on Atomic Spectra; Evidences in Favour of Bohr's Theory; Correspondence Principle; Critical Potentials; Sommerfeld's Relativistic Atom Model; The Vector Atom Model; Quantum Numbers Associated with the Vector Atom Model; The Pauli Exclusion Principle; The Periodic Classification of Elements; Magnetic Dipole Moment due to Orbital Motion of the Electron; Magnetic Dipole Moment due to Spin; Zeeman Effect; Larmor's Theorem; Stark Effect.

UNIT-II Particle and Wave properties of Waves 15 Lectures

Particle properties of Waves: Electromagnetic waves, Blackbody Radiation, Photo-electric effect, Einstein's Photoelectric Equation, Compton scattering, pair production, photons and gravity.

Wave properties of the Particles: De Broglie waves, description of wave, Probability. Wave amplitude and wave functions, Group and Phase velocities and relation between them, Davisson-Germer experiment, Heisenberg uncertainty principle.

15 Lectures

UNIT-III Nuclear Transformations

15 Lectures

stability of the nucleus; Law of radioactive decay; Mean life and half-life; types of radioactive decays, Alpha decay; Beta decay, Gamma ray emission, positron emission and electron capture, radioactive series, energy-momentum conservation: electron-positron pair creation by gamma photons in the vicinity of a nucleus, nuclear reactions and Q values.

Nuclear Fission: Energy Released in Fission; Chain Reaction; Atom Bomb; Nuclear Reactors; Nuclear Fusion, Source of Stellar Energy; Thermonuclear reactions

UNIT IV Artificial Transmutation of Elements 15 Lectures

The Discovery of Artificial Transmutation; Bohr's Theory of Nuclear Disintegration, nuclear reactions; The Q-Value Equation for a Nuclear Reaction; Nuclear Reactions; Energy Balance in Nuclear Reactions and the Q-value; Threshold Energy of an Endoergic Reaction; Nuclear Transmutations; The Scattering Cross-Section; Discovery; Preparation of Radioelements; Applications of Radioisotopes; The Discovery of the Neutron; Basic Properties of the Neutron. Classification of Neutrons; Neutron Sources; Neutron Detection; Neutron Collimator.

Learning Experience

The course content will be made experiential and participatory through a blend of interactive lectures, hands-on experiments, and collaborative projects. Students will engage in simulations and lab experiments that replicate key historical experiments, such as Rutherford's scattering experiment and the Davisson-Germer experiment, allowing them to directly observe the principles they learn. Group discussions and problem-solving sessions will encourage collaborative learning, where students can apply theoretical knowledge to real-world scenarios. Additionally, students will participate in peer-led seminars and presentations, fostering a deeper understanding of complex concepts like quantum mechanics and nuclear transformations. By incorporating these experiential activities, the course will not only reinforce theoretical understanding but also develop critical thinking, problem-solving skills, and teamwork, ensuring a well-rounded and engaging learning experience.

Textbooks

- 1. Concepts of Modern Physics, Arthur Beiser, 2002, McGraw-Hill.
- 2. Modern Physics (17th Ed.), 2013, S. Chand & Company Pvt. Ltd.

Suggested Readings

- 1. Introduction to Modern Physics, Rich Meyer, Kennard, Coop, 2002, Tata McGraw Hill
- 2. Introduction to Quantum Mechanics, David J. Griffith, 2005, Pearson Education.
- Physics for scientists and Engineers with Modern Physics, Jewett and Serway, 2010, Cengage Learning.
- Quantum Mechanics: Theory & Applications, A.K.Ghatak & S.Lokanathan, 2004, Macmillan 2002.

Open Educational Resources (OER)

- 1. <u>https://www.youtube.com/live/uFF_ptEDN0o?feature=share</u>
- 2. https://youtu.be/3It49x562b0
- 3. <u>https://youtu.be/UyWXOIK0uSg</u>
- 4. <u>https://youtu.be/QpXIPPln3Ig</u>
- 5. <u>https://youtu.be/Hpn5G1FiuCs</u>
- 6. <u>https://youtu.be/Lhxx2jQmLH4</u>
- 7. <u>https://youtu.be/PNBk5LjweEk</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SCPH202	Quantum mechanics and applications	L	Τ	Р	C
Version1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Mathematical Physics				

Course Perspective: This course is crucial for students as it builds a solid foundation in quantum mechanics, a cornerstone of modern physics. Students will develop analytical and problem-solving skills by understanding wave functions, Schrödinger equations, and quantum systems like atoms in magnetic fields. This knowledge is essential for careers in research, technology, and applied physics, with real-world applications in quantum computing, nanotechnology, and spectroscopy. By mastering these concepts, students will be well-prepared for advanced studies and professional roles in cutting-edge scientific and technological fields.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Understanding key quantum mechanics principles, including the Schrödinger equation, wave functions, and probability densities.

CO2: Applying the knowledge and solve problems involving particle behaviour in potential wells, harmonic oscillators, and hydrogen-like atoms using quantum mechanics.

CO3: Analysing quantum systems, interpreting wave functions, energy levels, and quantum numbers, and predicting outcomes like the Zeeman effect.

CO4: Evaluating quantum principles in scenarios like atoms in magnetic fields, assessing spin, angular momentum, and the Pauli exclusion principle.

Course Content

Unit1:WaveFunctionandSchrodingerequation20 Contact Hours

Time dependent Schrodinger equation and dynamical evolution of a quantum state; Properties of Wave Function. Interpretation of Wave Function Probability and probability current densities in three dimensions; Conditions for Physical Acceptability of Wave Functions. Normalization. Linearity and Superposition Principles. Eigenvalues and Eigenfunctions. Position, momentum and Energy operators; commutator of position and momentum operators; Expectation values of position and momentum. Wave Function of a Free Particle.

Time independent Schrodinger equation-Hamiltonian, stationary states and energy eigenvalues; expansion of an arbitrary wavefunction as a linear combination of energy eigenfunctions; General solution of the time dependent Schrodinger equation in terms of linear combinations of stationary states; Application to spread of Gaussian wave-packet for a free particle in one dimension: wave packets, momentum space wavefunction (qualitative discussion); Position-momentum uncertainty principle.

Unit 2: General discussion of bound states in an arbitrary potential 13 Contact Hours

continuity of wavefunction, boundary condition and emergence of discrete energy levels; application to one-dimensional problem-square well potential; Quantum mechanics of simple harmonic oscillator-energy levels and energy eigenfunctions; Hermite polynomials; ground state, zero-point energy & uncertainty principle.

Unit3:Quantumtheoryofhydrogen-likeatoms12 Contact Hours

time independent Schrodinger equation in spherical polar coordinates; separation of variables for second order partial differential equation; angular momentum operator & quantum numbers; Radial wavefunctions; shapes of the probability densities for ground & first excited states; Orbital angular momentum quantum numbers l and m; s, p, d shells.

Unit4:AtomsinElectric&MagneticFields15 Contact hours

Space quantization- Electron Spin and Spin Angular Momentum. Larmor's Theorem. Spin Magnetic Moment, Stern- Gerlach Experiment. Zeeman Effect, Gyromagnetic Ratio and Bohr Magneton. (8 Lectures)

Atoms in External Magnetic Fields: - Normal Zeeman Effect. Paschen Back (Qualitative Discussion only).

Many electron atoms: Pauli's Exclusion Principle. Symmetric & Antisymmetric Wave Functions, Fine structure. Spin orbit coupling-LS and JJ coupling. Spectral Notations for Atomic State- Term symbols, Total angular momentum, Vector Model, Hund's Rule.

Learning Experience

The course will be conducted through a mix of interactive lectures, hands-on laboratory experiments, and collaborative problem-solving sessions. Lectures will introduce theoretical concepts, while lab sessions will allow students to explore quantum phenomena through practical experiments, such as measuring energy levels and analyzing atomic spectra. Group discussions and workshops will encourage students to apply their knowledge to real-world scenarios and complex quantum systems. Interactive simulations and visualizations will help in understanding abstract concepts like wave functions and potential wells. This approach ensures that learning is experiential and participatory, bridging theoretical knowledge with practical applications.

Textbooks

1. Introduction to Quantum Mechanics, D.J Griffith, Prentice Hall (1994).

2. Advanced Quantum Mechanics, Satya Prakesh, Kedarnaath Ramnaath (2016)

Suggested Readings

1. Modern Quantum Mechanics, J.J Sakurai, Revised Edition, 1994, Addision-Wesley.

2. Advanced Quantum Mechanics, B,S, Rajput, Pragati Prakashan (2004)

3. Quantum Mechanics: Theory and Applications, (2019), (Extensively revised 6th Edition), Ajoy Ghatak and S. Lokanathan, Laxmi Publications, New Delhi.

4. Quantum Mechanics, Eugene Merzbacher, 2004, John Wiley and Sons, Inc.

Open Educational Resources (OER)

- 1. https://ocw.mit.edu/courses/physics/8-04-quantum-physics-i-spring-2016/
- 2. https://online.stanford.edu/courses/soe-ycphysics0002-quantum-mechanics
- 3. https://www.khanacademy.org/science/physics/quantum-physics/quantum-numbers-

and-orbitals/a/quantum-mechanics

- 4. https://ocw.mit.edu/courses/physics/8-05-quantum-physics-ii-fall-2013/
- 5. https://ocw.uci.edu/courses/chem_131a_quantum_principles.html
- 6. https://openstax.org/details/books/university-physics-volume-3

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. Internal Marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

SEMESTER IV					
SCPH204	Waves and Optics	L	Τ	Р	С
Version 1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co-	Electromagnetic Theory				
Requisites					

Course Perspective - This course is essential for providing a deep understanding of wave phenomena and optics. Students will develop strong analytical skills through topics like wave superposition, interference, and diffraction. These concepts are crucial for careers in optics, telecommunications, and engineering. Real-world applications of this course include designing optical instruments, acoustic engineering, and holography for data storage. By mastering these topics, students gain the knowledge and skills needed for advanced studies and diverse careers in science and technology, making this course foundational for both academic and professional development.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Understanding wave superposition, interference, and diffraction and demonstrate their mathematical and physical principles.

CO2: Applying the knowledge gained in solving problems related to wave motion and optics, including wave velocities and standing waves.

CO3: Analyzing wave patterns like Lissajous figures and various interference and diffraction patterns using graphical and analytical methods.

CO4: Evaluating optical instruments, assessing their resolving power and accuracy in applications.

Course Content

UNIT-I Superposition of Collinear Harmonic oscillations 15 Lecture Hours

Linearity and Superposition Principle. Superposition of two collinear oscillations having (1) equal frequencies and (2) different frequencies (Beats). Superposition of N collinear Harmonic Oscillations with (1) equal phase differences and (2) equal frequency differences.

Superposition of two perpendicular Harmonic Oscillations: Graphical and Analytical Methods. Lissajous Figures with equal an unequal frequency and their uses.

Wave Motion: Plane and Spherical Waves. Longitudinal and Transverse Waves. Plane Progressive (Travelling) Waves. Wave Equation. Particle and Wave Velocities. Differential Equation. Pressure of a Longitudinal Wave. Energy Transport. Intensity of Wave. Water Waves: Ripple and Gravity Waves.

UNIT-II Superposition of Waves

15 Lecture Hours

Velocity of Waves: Velocity of Transverse Vibrations of Stretched Strings. Velocity of Longitudinal Waves in a Fluid in a Pipe. Newton's Formula for Velocity of Sound. Laplace's Correction.

Superposition of Two Harmonic Waves: Standing (Stationary) Waves in a String: Fixed and Free Ends. Analytical Treatment. Phase and Group Velocities. Changes with respect to Position and Time. Energy of Vibrating String. Transfer of Energy. Normal Modes of Stretched Strings. Plucked and Struck Strings. Melde's Experiment. Longitudinal Standing Waves and Normal Modes. Open and Closed Pipes. Superposition of N Harmonic Waves.

UNIT-III WAVE OPTICS

15 Lecture Hours

Wave Optics: Electromagnetic nature of light. Definition and properties of wave front. Huygens Principle. Temporal and Spatial Coherence.

Interference: Division of amplitude and wavefront. Young's double slit experiment. Lloyd's Mirror and Fresnel's Biprism. Phase change on reflection: Stokes' treatment. Interference in Thin Films: parallel and wedge-shaped films. Fringes of equal inclination (Haidinger Fringes); Fringes of equal thickness (Fizeau Fringes). Newton's Rings: Measurement of wavelength and refractive index.

Interferometer: Michelson Interferometer-(1) Idea of form of fringes (No theory required), (2) Determination of Wavelength, (3) Wavelength Difference, (4) Refractive Index, and (5) Visibility of Fringes. Fabry-Perot interferometer.

UNIT-IV Diffraction

15 Lecture Hours

Diffraction: Kirchhoff's Integral Theorem, Fresnel-Kirchhoff's Integral formula. (Qualitative discussion only)

Fraunhofer diffraction: Single slit. Circular aperture, Resolving Power of a telescope. Double slit. Multiple slits. Diffraction grating. Resolving power of grating.

Fresnel Diffraction: Fresnel's Assumptions. Fresnel's Half-Period Zones for Plane Wave. Explanation of Rectilinear Propagation of Light. Theory of a Zone Plate: Multiple Foci of a Zone Plate. Fresnel's Integral, Fresnel diffraction pattern of a straight edge, a slit and a wire. **Holography:** Principle of Holography. Recording and Reconstruction Method. Theory of Holography as Interference between two Plane Waves. Point source holograms.

Learning Experience:

The course will be conducted through a blend of lectures, hands-on experiments, and interactive problem-solving sessions. Lectures will provide theoretical foundations, while lab sessions will allow students to conduct experiments on wave behavior, interference, and diffraction. Group discussions and collaborative projects will make learning participatory, encouraging students to explore real-world applications of concepts. Demonstrations using optical instruments and simulations will further enhance experimential learning. By engaging in active experimentation and analysis, students will directly apply theoretical knowledge, making the course both practical and immersive.

Textbooks

- 1. The Physics of Waves and Oscillations, N.K. Bajaj, 1998, Tata McGraw Hill.
- 2. Optics, Ajoy Ghatak, 2008, Tata McGraw Hill

Suggested Readings

- 1. Waves: Berkeley Physics Course, vol. 3, Francis Crawford, 2007, Tata McGraw-Hill.
- 2. Fundamentals of Optics, F.A. Jenkins and H.E. White, 1981, McGraw-Hill
- 3. Principles of Optics, Max Born and Emil Wolf, 7th Edn., 1999, Pergamon Press.
- 4. The Physics of Vibrations and Waves, H. J. Pain, 2013, John Wiley and Sons.

Open Educational Resources (OER)

- 1. <u>https://youtu.be/JWfYzyPvYNU</u>
- 2. <u>https://youtu.be/xjiS1lblx_c</u>
- 3. https://youtu.be/iWSNa8BCgaI
- 4. https://youtu.be/420N11Vuf g
- 5. https://youtu.be/nzwHI6UUnuA
- 6. https://youtu.be/mL63s4QqVPc
- 7. <u>https://youtu.be/hpLADfPpXwg</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

SEMESTER IV					
SCPH252	Waves and Optics Lab	L	T	Р	C
Version 1.0		0	0	4	2
Category of Course	Major				
Total Contact Hours	30				
Pre-Requisites/ Co- Requisites	Basics of Waves and optics theory				

Course Perspective This course enhances student learning by providing practical experience in wave optics and interference phenomena, crucial for understanding advanced physics concepts. Academically, students deepen their knowledge of optical measurements and techniques. Professionally, they gain hands-on skills in using instruments like interferometers and diffraction gratings, preparing them for careers in research, engineering, and technology. The course's real-world applications include designing optical systems and analysing materials in industries like telecommunications and material science. By mastering these techniques, students are equipped to tackle complex problems and innovate in scientific and technological fields.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Observing wave interference and diffraction through experiments like Melde's and Lissajous figures.

CO2: Imitating techniques for measuring wavelengths and refractive indices using tools like Michelson's interferometer.

CO3: Practicing determining optical properties, such as refractive index and dispersive power, through hands-on experiments.

Course Content

1. To determine the frequency of an electric tuning fork by Melde's experiment and verify λ^2 –T law.

2. To investigate the motion of coupled oscillators.

3. To study Lissajous Figures.

4. Familiarization with: Schuster's focusing; determination of angle of prism.

5. To determine refractive index of the Material of a prism using sodium source.

6. To determine the dispersive power and Cauchy constants of the material of a prism using mercury source.

7. To determine the wavelength of sodium source using Michelson's interferometer.

8. To determine wavelength of sodium light using Fresnel Biprism.

9. To determine wavelength of sodium light using Newton's Rings.

10. To determine the thickness of a thin paper by measuring the width of the interference fringes produced by a wedge-shaped Film.

11. To determine wavelength of (1) Na source and (2) spectral lines of Hg source using plane diffraction grating.

12. To determine dispersive power and resolving power of a plane diffraction grating.

Learning Experience

The course will combine theoretical instruction with hands-on laboratory work to ensure an experiential and participatory learning experience. Lectures will introduce concepts such as wave interference and diffraction, while lab sessions will involve direct experimentation, including Melde's experiment, Lissajous figures, and various optical measurements. Students will actively set up experiments, collect and analyse data, and interpret results. Group discussions and collaborative projects will enhance engagement and problem-solving skills.

Real-time troubleshooting and feedback will support practical understanding, making learning interactive and directly applicable to real-world scenarios in optics and materials science.

Textbooks

- 1. B. Sc. Practical Physics by Geeta Sanon
- 2. Advanced Practical Physics for students, B.L. Flint and H.T. Worsnop, 1971, Asia Publishing House

Suggested Readings

- 1. A Textbook of Practical Physics, I. Prakash & Ramakrishna, 11th Ed., 2011, Kitab Mahal
- 2. Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers
- A Laboratory Manual of Physics for undergraduate classes, D.P. Khandelwal, 1985, Vani Pub.

Open Educational Resources (OER)

- 1. <u>https://youtu.be/hwWPDqHFxOg</u>
- 2. <u>https://youtu.be/tnQn994Eqm8</u>
- 3. https://youtu.be/e4poHiwoTH8
- 4. <u>https://youtu.be/S3PiJ0PQghc</u>
- 5. <u>https://youtu.be/JvVOaqej1II</u>
- 6. <u>https://youtu.be/tQj5v74Q8p0</u>
- 7. <u>https://youtu.be/1Qc-HIml-U4</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Practical's)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks

II. External Marks (Practical's):	50 Marks
End Term Examination	

SEMESTER IV					
SEC075	Basic IT Tools	L	Т	Р	C
Version 1.0		3	0	0	3
Category of Course	Skill Enhancement Course / MOOC				
Total Contact Hours	45				
Pre-Requisites/ Co	-				
Requisites					

Course Perspective

The "Basic IT Tools" course equips students with essential skills in information technology, focusing on fundamental concepts and practical applications. It covers the basics of computer hardware, software, operating systems, and internet usage, emphasizing the importance of digital literacy in today's world. Students learn to navigate common productivity software like word processors, spreadsheets, and presentation tools, enabling them to efficiently manage and present information. Additionally, the course introduces basic networking concepts and cybersecurity awareness, preparing students to safely and effectively use IT resources in both academic and professional settings.

Course Outcomes

Upon completion of the course, the learner will be:

CO1. Applying word-processing skills to generate documents with appropriate formatting, layout, review, and referencing.

CO2. Utilizing spreadsheet functions and formulas to manage and analyzing data in worksheets and workbooks.

CO3. Evaluating data using spreadsheets to make informed decisions.

CO4. Creating meaningful representations of data through charts and pivot tables.

CO5. Demonstrating proficiency in managing data in database tables and utilizing it to generate queries, forms, and reports.

Course Content

Unit 1: Introduction to Spreadsheets

Lecture: 15 hours

Spreadsheets: Concept of worksheets and workbooks, creating, opening, closing and saving workbooks, moving, copying, inserting, deleting and renaming worksheets, working with multiple worksheets and multiple workbooks, controlling worksheet views, naming cells using name box, name create and name define; Exchanging data using clipboard, object linking and embedding; Printing and Protecting worksheets: Adjusting margins, creating headers and footers, setting page breaks, changing orientation, creating portable documents and printing data and formulae; Implementing file level security and protecting data within the worksheet; Understanding absolute, relative and mixed referencing in formulas, referencing cells in other worksheets and workbooks, correcting common formula errors, working with inbuilt function categories like mathematical, statistical, text, lookup, information, logical, database, date and time and basic financial functions.

Unit 2: Data Analysis in Spreadsheets

Consolidating worksheets and workbooks using formulae and data consolidate command; Choosing a chart type, understanding data points and data series, editing and formatting chart elements, and creating sparkline graphics, analysing data using pivot tables: Creating, formatting and modifying a pivot table, sorting, filtering and grouping items, creating calculated field and calculated item, creating pivot table charts, producing a report with pivot tables. Introduction to recording and execution of macros.

Unit 3: Word Processing hours

Introduction: Creating and saving your document, displaying different views, working with styles and character formatting, working with paragraph formatting techniques using indents, tabs, alignment, spacing, bullets and numbering and creating borders; Page setup and sections: Setting page margins, orientation, headers and footers, end notes and foot notes, creating

Lecture: 15

Lecture: 15 hours

section breaks and page borders; Working with tables: Creating tables, modifying table layout and design, sorting, inserting graphics in a table, table math, converting text to table and vice versa; Create newspaper columns, indexes and table of contents, Spell check your document using inbuilt and custom dictionaries, checking grammar and style, using thesaurus and finding and replacing text; Create bookmarks, captions and cross referencing, adding hyperlinks, adding sources and compiling and bibliography; Mail merge: Creating and editing your main document and data source, sorting and filtering merged documents and using merge instructions like ask, fill-in and if-then-else; Linking and embedding to keep things together.

his course will integrate lectures, interactive sessions, and hands-on projects to build a foundational understanding of essential IT tools and their applications.

Instruction Methods:

- Lectures: Core concepts of IT, including computer basics, software applications, and internet usage, will be taught using multimedia presentations and real-life scenarios.
- Interactive Sessions: Q&A sessions, live demonstrations, and group discussions will encourage active learning and reinforce practical skills.

Technology Use:

• Online Platforms: A Learning Management System (LMS) will host course materials, recorded lectures, assignments, and discussion forums, promoting extended learning and collaboration.

Assessments:

- Formative: Regular quizzes, practical assignments, and online discussions will provide ongoing feedback and help students monitor their progress.
- **Summative:** Exams, project submissions, and peer reviews will evaluate students' understanding and practical application of IT tools.

Support: The instructor will provide additional help through office hours, and peer collaboration will be fostered through group projects and review sessions, ensuring continuous feedback and improvement in achieving course objectives.

Textbooks

- Swinford, E., Dodge, M., Couch, A., Melton, B. A. (2013). Microsoft Office Professional 2013. United States: O'Reilly Media.
- Wang, W. (2018). Office 2019 For Dummies. United States: Wiley. Microsoft Lambert, J. (2019). Microsoft Word 2019 Step by Step. United States: Pearson Education.
- 3. Jelen, B. (2013). Excel 2013 Charts and Graphs. United Kingdom: Que.

Suggested Readings

- 1. Alexander, M., Jelen, B. (2013). Excel 2013 Pivot Table Data Crunching. United Kingdom: Pearson Education.
- 2. Alexander, M., Kusleika, R. (2018). Access 2019 Bible. United Kingdom: Wiley

Open Educational Resources (OER)

- 1. <u>http://ocw.mit.edu/courses/electrical-engineering-and-computer-science/6-00sc-introduction-to-computer-science-and-programming-spring-2011</u>
- 2. http://learn.saylor.org/course/cs101
- 3. http://openstax.org/books/introduction-to-computer-applications-for-business

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessments (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

AEC002	Communication & Personality Development	L	Τ	P	C
Version 1.0		3	0	0	3
Category of Course	Ability Enhancement Course				
Total Contact Hours	45				
Pre-Requisites/ Co-					
Requisites					

Course Perspective:

The course enhances public speaking and presentation skills, helps students confidently convey ideas, information & build self-reliance and competence needed for career advancement. Personality assessments like the Johari Window and Myers & Briggs Type Indicator (MBTI) provide frameworks to enhance self-understanding, helps people increase their self-awareness, understand and appreciate differences in others and apply personality insights to improve their personal and professional effectiveness. Interpersonal skills included in the course deal with important topics like communication, teamwork and leadership, vital for professional success.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Improve public speaking and presentation abilities to confidently convey ideas and information.

CO 2: Understand the framework of Communication to augment oratory skills and written English

CO 3: Cultivate essential soft skills required at different workplaces.

Course Content

Unit 1: Developing self and others

Content Summary: Self Awareness, Personality Concepts (Personality Assessments -Johari Window, Myers & Brigg), Self-Management, Self-Esteem, Self-Efficacy, Interpersonal skills, mindset, grit and working in teams.

Unit 2: Enhancing Reading and Writing Skills

Content Summary: Speed reading and its importance in competitive examinations, techniques for speed reading, note-taking, and critical analysis. Paragraph Writing, Essay and Summary writing, Business Letter, Email writing

Unit 3: Effective Communication and Public Speaking 7 hours

Content Summary: Communication Framework, barriers & overcoming these barriers, Group Discussions, Extempore & Public Speaking drills, to manage stage fright and anxiety. Structuring and organizing a presentation (Oral & PPT), Etiquettes, Grooming, Body Language and Conversation starters, TMAY.

Unit 4: Career Guide and readiness

Cover Letter, ATS friendly resume, Elevator Pitch, Video Resume (Visume), Networking, Group Discussion, Mock Interviews. Capstone Project

Learning Experience:

The course will be conducted using experiential and participatory methods such as interactive workshops, group discussions, and practical exercises. Students will engage in self-assessment tools, collaborative writing, and public speaking drills to develop critical skills. Hands-on activities like mock interviews, resume writing, and capstone projects will provide real-world career preparation. Continuous peer feedback and role-playing will enhance communication, teamwork, and problem-solving abilities.

Textbooks:

8 hours

15 hours

6 hours

- 1. Talking to Strangers Malcom Gladwell
- 2. Fierce Conversation Scot Susan
- 3. Public Speaking William S. Pfeiffer, Pearson
- 4. Soft Skills for Everyone Jeff Butterfield
- 5. Business Communication Rajendra Pal, J S Korlahalli
- 6. The power of Positive Attitude -Roger Fritz
- 7. Believe in Yourself Dr. Joseph Murphy

Suggested Readings:

- 1. Websites & MOOCs
 - i. <u>www.16personalities.com</u>
 - ii. www.tonyrobbins.com
- 2. Specific Research Papers
 - i. GALLUP PRESS RESEARCH
 - ii. FRANKLIN COVEY LEADERSHIP CENTRE
- 3. Videos
- i. The 7 Habits of Highly Effective People, Dr. Stephen R. Covey
- ii. I Am Not Your Guru, Tony Robbins

4. Podcast

- i. The Tim Ferriss Show
- 5. Magazines
 - i. SUCCESS Magazine
- 6. Journals
- i. The IUP Journal of Soft Skills

Evaluation Scheme:

Evaluation components	Weight age
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	

III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40% marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER V					
SCPH301	Thermodynamics and Statistical Mechanics	L	Т	P	С
Version1.0		4	0	0	4
Category of Course	Major	I	1		1
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Classical Mechanics				

Course Perspective This course offers a comprehensive understanding of both thermodynamics and statistical mechanics, bridging classical and quantum perspectives. It equips students with the skills to apply thermodynamic and statistical principles to solve real-world problems, including those involving energy conversion, radiation, and quantum gases. Through this course, students will gain critical thinking skills, enabling them to analyze complex physical systems and apply theoretical concepts to practical scenarios, which are crucial for careers in physics and engineering.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding laws of thermodynamics and Bose-Einstein and Fermi-Dirac statistics to understand quantum gases and related phenomena like Bose-Einstein condensation and electron behaviour in metals.

CO2: Applying laws of thermodynamics and Statistical mechanics to solve problems related to involving entropy, energy conversion, heat engines, macrostate, microstate, and partition functions.

CO3: Analyzing the limitations of classical physics and the necessity for quantum statistical approaches in describing physical systems

CO4: Evaluating the behavior of complex physical systems, such as photon gases and electron gases, using Bose-Einstein and Fermi-Dirac statistics, and **assess** their real-world applications.

Course Content

UNIT-I Zeroth and First Law of Thermodynamics

15 Lecture Hours

Extensive and intensive Thermodynamic Variables, Thermodynamic Equilibrium, Zeroth Law of Thermodynamics & Concept of Temperature, Concept of Work & Heat, State Functions, First Law of Thermodynamics and its differential form, Internal Energy, First Law & various processes, Applications of First Law: General Relation between C_P and C_V, Work Done during Isothermal and Adiabatic Processes, Compressibility and Expansion Co-efficient.

UNIT-II Second Law of Thermodynamics

15 Lecture Hours

Reversible and Irreversible process with examples. Conversion of Work into Heat and Heat into Work, Heat Engines, Carnot's Cycle, Carnot engine & efficiency. Refrigerator & coefficient of performance, 2nd Law of Thermodynamics: Kelvin- Planck and Clausius Statements and their Equivalence. Carnot's Theorem, Applications of Second Law of Thermodynamics: Thermodynamic Scale of Temperature and its Equivalence to Perfect Gas Scale.

Entropy: Concept of Entropy, Clausius Theorem. Clausius Inequality, Second Law of Thermodynamics in terms of Entropy, Entropy of a perfect gas, Entropy Changes in Reversible and Irreversible processes with examples, Entropy of the Universe, Temperature–Entropy diagrams for Carnot's Cycle, Third Law of Thermodynamics, Unattainability of Absolute Zero.

UNIT-III Classical Statistics

Macrostate & Microstate, Elementary Concept of Ensemble, PhaseSpace, Entropy and Thermodynamic Probability, Maxwell-Boltzmann Distribution Law, Partition Function, Thermodynamic Functions of an Ideal Gas, Classical Entropy Expression, Gibbs Paradox, Sackur Tetrode equation, Law of Equipartition of Energy – Applications to Specific Heat and its Limitations, Thermodynamic Functions of a Two-Energy Levels System.

UNIT-IV Bose-Einstein & Fermi-Dirac Statistics 15 Lecture Hours

Bose-Einstein Statistics: B-E distribution law, Thermodynamic functions of a strongly Degenerate Bose Gas, Bose Einstein condensation, properties of liquid He (qualitative description), Radiation as a photon gas and Thermodynamic functions of photon gas. Fermi-Dirac Statistics: Fermi-Dirac Distribution Law, Thermodynamic functions of a Completely and strongly Degenerate Fermi Gas, Fermi Energy, Electron gas in a Metal, Specific Heat of Metals, Relativistic Fermi gas.

Learning Experience This course will employ lectures, problem-solving sessions, and interactive group activities. Students will engage in hands-on learning through problem sets and group projects. The course will be supported by online resources and active discussions. Assessments will include midterms, a final exam, and graded problem sets.

Textbook:

- 1. Thermal Physics, Agarwal and Prakash, Pragati Prakashan Educational Publishers.
- 2. Statistical Physics, Berkeley Physics Course, F. Reif, 2008, Tata McGraw-Hill
- 3. Modern Thermodynamics with Statistical Mechanics, Carl S. Helrich, 2009, Springer

Reference Books:

- 1. Statistical Mechanics, R.K. Pathria, Butterworth Heinemann: 2nd Ed., 1996, Oxford University Press.
- 2. Statistical Physics, Berkeley Physics Course, F. Reif, 2008, Tata McGraw-Hill
- 3. Statistical and Thermal Physics, S. Lokanathan and R.S. Gambhir. 1991, Prentice Hall
- 4. Thermodynamics, Kinetic Theory and Statistical Thermodynamics, Francis W. Sears and
- 5. Gerhard L. Salinger, 1986, Narosa.
- An Introduction to Statistical Mechanics & Thermodynamics, R.H. Swendsen, 2012, Oxford Univ. Press

15 Lecture Hours

Open Educational Resources (OER)

- 1. <u>https://byjus.com/physics/thermodynamics/</u>
- 2. <u>https://www.youtube.com/watch?v=8N1BxHgsoOw</u>
- 3. <u>https://www.youtube.com/watch?v=NyOYW07-L5g</u>
- 4. <u>https://www.youtube.com/watch?v=8xRFqrNyJCg&list=PLyqSpQzTE6M9iXvWVCopr6</u> <u>7kKt61ntzII</u>
- 5. <u>https://www.youtube.com/watch?v=Qz5D2bUQk4c&list=PLQCYtYck4nKwkiEyJ_mzuz</u> <u>z9aQla1Kvdh</u>
- 6. <u>https://www.youtube.com/watch?v=o1EinUCgFsw&list=PL74Pz7AXMAnOSPWBIzmZ</u> <u>OpSs2KgchH3dP</u>

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER V					
SCPH303	Digital Systems and Applications	L	Τ	Р	С
Version 1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co	Digital Electronics				

Requisites			

Course Perspective

This course is intended to cover most of the basic topics of digital electronics including Number systems, Logic gates and logic families,Boolean Algebra and Simplification, Arithmetic circuits, Data Processing Circuits, various Flip - flops, Clocks And Timers and Shift registers & counters. This course gives the circuit knowledge to students and students will be even able to start their start-ups. Course will also be beneficial for students in day-to-day life. The Digital Systems and Applications Theory course is crucial for B.Sc (H) Physics students as it integrates digital logic and circuit design with physics concepts, preparing them for careers in technology. Students will gain skills in designing and analyzing digital systems, which are essential for developing advanced technologies and solving real-world problems. This course provides a strong foundation for careers in electronics, computing, and research, applying theoretical knowledge to practical applications like embedded systems and signal processing.

Course Outcomes

Upon completion of the solid-state physics course, the learner will be:

CO1: Understanding the basic concepts of digital systems and logic design.

CO2: Applying digital logic principles to solve practical problems and design digital systems.

CO3: Analyzing and break down digital circuits and systems to understand their operation and performance.

CO4: Evaluating different digital design methods to choose the most effective solutions.

Course Content

Unit No I: Number Systems

Decimal, binary, octal and hexadecimal systems - conversion from one number system to another. Codes - BCD code - Excess 3 code, Gray code ,Binary arithmetic - Binary addition subtraction, 1's and 2's complement - Binary multiplication and division.

Unit No II: Boolean Algebra Simplification of Logic circuits No. of Hours 15

No. of Hours 15

Laws and theorems of Boolean algebra - De Morgan's theorems and their circuit implications - Duality theorem, simplification of Boolean equations – sum of products method (SOP), product of sums methods (POS) Karnaugh map(K Map) - pairs, quads, octets - 2,3 and 4 variables ,Reduction of POS using K Map.

Unit No III: Combinational Logic Circuits

Arithmetic building blocks - Half adder - Full adder - parallel binary adder - Half subtractor - Full subtractor - The adder-subtractor - digital comparator - parity checker / generator , Multiplexers – Demultiplexers, Decoders.

Unit No IV: Sequential Logic Circuits

Flip - flops - RS Flip Flop - Clocked RS Flip-flop - D flip-flop - JK flip-flop - JK master slave flip-flop - T type flip-flop registers and counters: Types of registers - serial in serial out - serial in parallel out - parallel in serial out - parallel in parallel out - ring counter.

Learning Experience

The Digital Systems and Applications Theory course will be conducted through interactive lectures, hands-on circuit design activities, and simulations using digital design software. Students will engage in group projects, real-world case studies, and assignments to apply and reinforce their knowledge. Assessments will include quizzes, projects, and exams, with regular feedback provided by the instructor. Students will have opportunities for additional support and collaboration through peer reviews and group work, enhancing their learning experience.

Textbooks

 Malvino and Leech, Digital Principles and Application, 4th edition, Tata McGraw Hill, New Delhi.

Suggested Readings

1. Millman and Halkias, Integrated Electronics, International edition, McGraw Hill, New Delhi.

2. Thomas L. Floyd, Digital Fundamentals (Universal Book Stall, India).

Open Educational Resources (OER)

1. <u>https://cnx.org/contents/85c4b2c4-2b3b-4b73-b36a-3e76a9b8c506@1</u>

No. of Hours 15

No. of Hours 15

- 2. https://www.mathsisfun.com/binary-decimal-hexadecimal.html
- 3. https://www.geeksforgeeks.org/digital-electronics-binary-coded-decimal/
- 4. <u>https://www.khanacademy.org/computing/computer-science/algorithms/boolean-algebra</u>
- 5. https://www.youtube.com/playlist?list=PLBlnK6fEyqRhX6r2uhhlubuF5QextdCSM
- https://www.allaboutcircuits.com/textbook/digital/chpt-9/combinational-logicfunctions/
- 7. <u>https://www.youtube.com/playlist?list=PLBlnK6fEyqRjMH3mWf6kwqiTbT798eAO</u> <u>m</u>
- 8. https://www.youtube.com/playlist?list=PLBlnK6fEyqRhFUZX8d6GwI5H5frkZT2lk

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. Internal Marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

SEMESTER V					
SCPH351	Digital Systems and Applications Lab	L	T	Р	C
Version 1.0		0	0	4	2
Category of Course	Major Lab				
Total Contact Hours	30				
Pre-Requisites/ Co	Digital Electronics				

Requisites			

Course Perspective

This course is intended to cover most of the basic topics of digital electronics including Number systems, Logic gates and logic families, Boolean Algebra and Simplification, Arithmetic circuits, Data Processing Circuits, various Flip - flops, Clocks And Timers and Shift registers & counters. This course gives the experimental and circuit knowledge to students which will be beneficial for students in day-to-day life.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Observing and describing phenomena related to digital systems, demonstrating the ability to accurately record and interpret data from various experiments and applications.

CO2: Imitating standard procedures and techniques used in digital systems and applications, replicating established methods with attention to accuracy and detail.

CO3: Practicing utilizing digital tools and techniques, refining their skills through hands-on experience with digital circuits, systems, and applications.

Course Content

List of experiments

- 1. Verification of the truth tables of TTL gates.
- 2. Verify the NAND and NOR gates as universal logic gates.
- 3. Design and verification of the truth tables of Half and Full adder circuits.
- 4. Design and verification of the truth tables of Half and Full subtractor circuits.
- 5. Verification of the truth table of the Multiplexer 74150.
- 6. Verification of the truth table of the De-Multiplexer 74154.
- 7. Design and test of an S-R flip-flop using NOR/NAND gates.
- 8. Verify the truth table of a J-K flip-flop (7476).
- 9. Verify the truth table of a D flip-flop (7474).
- **10.** Operate the counters 7490, 7493.
- 11. Design of 4-bit shift register (shift right).

12. Design of modulo-4 counter using J K flip flop.

Learning Experience

The Digital Systems and Applications Laboratory course will be delivered through a hands-on approach, incorporating the use of advanced digital tools, simulation software, and hardware components to enhance practical learning. Students will actively participate in group projects, case studies, and real-world scenarios to apply theoretical concepts to practical problems. Assessments will include lab reports, technical presentations, and practical examinations. The course will provide continuous support and feedback from the instructor, along with opportunities for additional assistance and collaboration with peers. This approach ensures an engaging and interactive learning environment, fostering both individual and collaborative skills in digital systems and applications.

Textbooks

 Malvino and Leech, Digital Principles and Application, 4th edition, Tata McGraw Hill, New Delhi.

Suggested Readings

- 1. Modern Digital Electronics, R.P. Jain, 4th Edition, 2010, Tata McGraw Hill.
- 2. Basic Electronics: A text lab manual, P.B. Zbar, A.P. Malvino, M.A. Miller, 1994, Mc-Graw Hill.
- 3. Microprocessor Architecture Programming and applications with 8085, R.S. Goankar, 2002, Prentice Hall.
- 4. Microprocessor 8085: Architecture, Programming and interfacing, A. Wadhwa, 2010, PHI

Open Educational Resources (OER)

- 1. <u>https://www.youtube.com/watch?v=47u7b2yh7s8&pp=ygUsdmVyaWZpY2F0aW9uIG9</u> <u>mIHRydXRoIHRhYmxlcyBvZiBsb2dpYyBnYXRlcyA%3D</u>
- https://www.youtube.com/watch?v=lqN8xLTtdaA&pp=ygVeMy4JRGVzaWduIGFuZCB
 2ZXJpZmljYXRpb24gb2YgdGhlIHRydXRoIHRhYmxlcyBvZiBIYWxmIGFuZCBGdW
 xsIGFkZGVyIGFuZCBzdWJ0cmFjdG9yIGNpcmN1aXRzLg%3D%3D
- 3. <u>https://www.youtube.com/watch?v=SYTDxdACf2E&pp=ygVeMy4JRGVzaWduIGFuZC</u> <u>B2ZXJpZmljYXRpb24gb2YgdGhlIHRydXRoIHRhYmxlcyBvZiBIYWxmIGFuZCBGd</u>

WxsIGFkZGVyIGFuZCBzdWJ0cmFjdG9yIGNpcmN1aXRzLg%3D%3D

- 4. <u>https://www.youtube.com/watch?v=ap0RMkqHWHQ&pp=ygU1My4JMTIuCURlc2lnbi</u> <u>BvZiBtb2R1bG8tNCBjb3VudGVyIHVzaW5nIEogSyBmbGlwIGZsb3A%3D</u>
- 5. <u>https://www.youtube.com/watch?v=Ub1VixA-</u> <u>uSE&pp=ygU1My4JMTIuCURlc2lnbiBvZiBtb2R1bG8tNCBjb3VudGVyIHVzaW5nIEo</u> <u>gSyBmbGlwIGZsb3A%3D</u>
- 6. <u>https://www.youtube.com/watch?v=q_W_qBnOZvw&pp=ygUxMTEuCURlc2lnbiBvZiA</u> <u>0LWJpdCBzaGlmdCByZWdpc3RlciAoc2hpZnQgcmlnaHQpLg%3D%3D</u>
- 7. <u>https://www.youtube.com/watch?v=GJ8xxaIoIJ0&pp=ygUfT3BlcmF0ZSB0aGUgY291b</u> <u>nRlcnMgNzQ5MCwgNzQ5Mw%3D%3D</u>
- 8. <u>https://www.youtube.com/watch?v=p6yPvw88BJk&pp=ygUhSW50cm9kdWN0aW9uIH</u> <u>RvIE11bHRpcGxleGVyIDc0MTUw</u>
- 9. <u>https://www.youtube.com/watch?v=QKLWSs3z0C4&pp=ygUiUy1SIGZsaXAtZmxvcCB</u> <u>1c2luZyBOT1IvTkFORCBnYXRlcw%3D%3D</u>

Evaluation Scheme

Evaluation components	W	Weightage					
Internal marks (practic	al)						
I. Conduct of experiment	10	10 Marks					
II. Lab Record	10 Marks						
III. Lab Participation	10 Marks						
II. External Marks (pra	ctical)	50	Marl	KS			
End Term Examination							
SEMESTER V							
SCPH305	Classical Dynamics	L	Т	Р	C		
Version1.0		4	0	0	4		
	Major						
Category of Course	1414j01						
Category of Course Total Contact Hours	60						

Requisites		

Course Perspective

This course offers essential insights into advanced mechanics, including classical mechanics, Hamiltonian dynamics, special relativity, and fluid dynamics. By mastering these topics, students will gain critical skills in solving complex problems, from particle motion in electromagnetic fields to relativistic effects and fluid behaviour. These concepts are foundational for further studies in physics and engineering, as well as for careers in research, technology, and applied sciences. Understanding these principles is crucial for designing systems and technologies across various industries, making this course vital for academic and professional development.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and articulating fundamental principles and theories of advanced mechanics, explaining their applications and relevance to real-world phenomena.

CO2: Applying theoretical models and equations to solve practical problems, demonstrating proficiency in using physics principles in various contexts.

CO3: Analyzing experimental data and physical systems, interpreting results to identify patterns, relationships, and anomalies, and making informed conclusions based on their analysis.

CO4: Evaluating different methodologies and approaches, assessing their effectiveness and accuracy in solving complex problems, and making necessary adjustments to enhance problem-solving strategies.

Course Contents

UNIT I Classical Mechanics of Point Particles No. of Hours: 20

Review of Newtonian Mechanics; Application to the motion of a charge particle in external electric and magnetic fields- motion in uniform electric field, magnetic field- gyro radius and gyrofrequency, motion in crossed electric and magnetic fields. Generalized

coordinates and velocities, Hamilton's principle, Lagrangian and the Euler-Lagrange equations, one-dimensional examples of the Euler-Lagrange equations- one dimensional Simple Harmonic Oscillations and falling body in uniform gravity; applications to simple systems such as coupled oscillators

UNIT II Hamiltonian formulation

No. of Hours: 10

Canonical momenta & Hamiltonian. Hamilton's equations of motion. Hamiltonian for a harmonic oscillator, solution of Hamilton's equation for Simple Harmonic Oscillations; particle in a central force field- conservation of angular momentum and energy.

Small Amplitude Oscillations: Minima of potential energy and points of stable equilibrium, expansion of the potential energy around a minimum, small amplitude oscillations about the minimum, normal modes of oscillations example of N-identical masses connected in a linear fashion to (N -1) - identical springs.

UNIT III Special Theory of Relativity

Postulates of Special Theory of Relativity. Lorentz Transformations. Minkowski space. The invariant interval, light cone and world lines. Spacetime diagrams. Time -dilation, length contraction and twin paradox. Four vectors: space-like, time-like and light-like. Four-velocity and acceleration. Metric and alternating tensors. Four momentum and energy-momentum relations. Doppler effect from a four-vector perspective. Concept of four-force. Conservation of four-momentum. Relativistic kinematics. Application to two-body decay of an unstable particle. (33 Lectures)

UNIT IV Fluid Dynamics

Density and pressure P in a fluid, an element of fluid and its velocity, continuity equation and mass conservation, stream-lined motion, laminar flow, Poiseuille's equation for flow of a liquid through a pipe, Navier-Stokes equation, qualitative description of turbulence, Reynolds number.

Learning Experience

This course will be conducted through a dynamic blend of lectures, discussions, and integrating technology to enhance learning. Assignments and presentations will challenge students to apply and communicate their understanding creatively. Regular quizzes, exams, and timely feedback will assess and guide their progress. Support will be readily available from the course

No. of Hours: 10

No. of Hours: 20

instructor, and peer collaboration will be encouraged to enrich the learning experience, ensuring students receive comprehensive support and opportunities for growth.

Textbooks

1. Classical Mechanics, H.Goldstein, C.P. Poole, J.L. Safko, 3rd Edn. 2002, Pearson Education.

2. Classical Mechanics, J.C. Upadhyaya, 2nd Edn. 2005, Himalaya Publishing House

Suggested Readings

- 1. Mechanics, L. D. Landau and E. M. Lifshitz, 1976, Pergamon.
- 2. Classical Electrodynamics, J.D. Jackson, 3rd Edn., 1998, Wiley.
- 3. Classical Mechanics, P.S. Joag, N.C. Rana, 1st Edn., McGraw Hall.
- 4. Classical Mechanics, R. Douglas Gregory, 2015, Cambridge University Press.

Open Educational Resources (OER)

- 1. https://en.wikipedia.org/wiki/Spacetime_diagram
- 2. <u>https://en.wikipedia.org/wiki/Four-force</u>
- 3. <u>https://www.researchgate.net/profile/Mohamed_Mourad_Lafifi/post/In_a_robust_control_system_design_using_mixed-sensitivity_approach_the_sensitivity_is_constant_at_0_db_isnt_is_supposed_to_be_high-pass_filter/attachment/59d63ab079197b8077997c75/AS:407064284286977@147406309 6225/download/NotesCh12+Coupled+Oscillators+and+Normal+Modes.pdf</u>
- 4. https://mathsci.kaist.ac.kr/~nipl/am621/lecturenotes/Euler-Lagrange_equation.pdf

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	

Project/quizzes/assignment and essays/presentation/ participation/case studies/reflective journals (minimum of five components to be	
evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER V					
SCPH307	Laser Fundamentals	L	Т	Р	С
Version 1.0		4	2	0	6
Category of Course	Major- discipline specific Elective / N	1000			
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Optics				

Course Perspective

This course covers the fundamentals and industrial applications of laser technology. It begins with the basic principles of laser operation, including population inversion and the interaction of electromagnetic radiation with matter. The course then explores various types of lasers, such as He-Ne, CO₂, and Nd, along with their specific media. It further delves into laser electronics, beam characteristics, and material processing applications like machining and welding. Finally, the course examines the industrial uses of lasers in measurement, material processing, and other critical functions.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and describe the fundamental principles of laser operation, including population inversion, electromagnetic radiation, and the interaction of radiation with matter.

CO2: Applying knowledge of laser principles and systems to practical applications, including laser communication, material processing, and measuring physical quantities such as distance and velocity.

CO3: Analyzing the characteristics of advanced laser systems, including optical parametric oscillators and laser electronics, and their processing techniques and applications in various fields.

CO4: Evaluating and implementing the industrial applications of lasers in measurement and material processing, implementing appropriate laser technologies for specific tasks such as distance measurement, welding, and vaporization.

Course Contents

UNIT I Laser Fundamentals

Phenomenon of population inversion. The Nature of Electromagnetic Radiation. Interaction of Electromagnetic Radiation with Matter. Absorption and Emission of Radiation by atoms, ions and molecules.

UNIT II Types of lasers

Laser medium (solid state medium: crystals, glass, semiconductor, gaseous medium). Types of lasers: He-Ne laser, CO2 laser, Argon laser, Nd:YAG, Excimer laser, Diode laser, Fiber laser etc.

UNIT III Advanced Laser Systems:

Characteristics, Processing Techniques, and Applications: Optical parametric oscillator (OPO), Laser Electronics. Laser Beam Characteristics, Wavelength, Coherence, Mode and Beam Diameter, Polarisation. Laser Material Processing e.g. machining, welding, sintering, forming etc. Laser Fracture & Damage. Laser Communication.

UNIT IV Industrial Application of Lasers

No. of Hours: 15

Laser for measurement of distance, length, velocity, acceleration, current, voltage and Atmospheric effect – Material processing – Laser heating, welding, melting and trimming of material – Removal and vaporization.

Learning Experience

The learning experience for this course will be dynamic and integrative, combining theoretical foundations with practical insights. Students will begin by exploring the fundamental principles of laser technology, including the nature of electromagnetic radiation and its interaction with matter. They will then delve into the various types of lasers, gaining hands-on understanding of different laser mediums and their applications. As the course progresses, students will engage with advanced topics such as laser beam characteristics and electronics, culminating in the study of industrial applications where lasers are used for precise measurements and material processing. Throughout the course, a blend of interactive lectures and practical exercises will ensure a deep and applied understanding of laser technology.

Textbooks

- 1. "Lasers: Principles and Applications" by K. R. Nambiar. It provides a comprehensive understanding of the fundamentals of lasers, types of lasers, and their applications.
- 2. "Laser Electronics" by Joseph T. Verdeyen. It covers a wide range of topics including laser fundamentals, laser electronics, and laser beam characteristics.
- 3. "Laser Systems and Applications" by B.B. Laud. It gives a detailed view of laser applications in various fields, particularly in the industrial context.

Suggested Readings

- 1. "Fundamentals of Photonics" by Bahaa E. A. Saleh and Malvin Carl Teich. A comprehensive guide that touches on many aspects of photonics, including lasers.
- 2. "Laser Fundamentals" by William T. Silfvast. This book provides an in-depth understanding of laser fundamentals.
- 3. "Introduction to Laser Technology" by C. Breck Hitz, J. J. Ewing, and Jeff Hecht. A book focused on laser technology and its diverse applications.
- 4. "Handbook of Laser Technology and Applications" by Colin Webb and Julian Jones. This handbook provides a comprehensive overview of lasers and their broad applications.

Open Educational Resources (OER)

1. MIT OpenCourseWare - Introduction to Lasers

- 2. Introduction to Lasers Stanford University
- 3. Laser Basics Khan Academy
- 4. **RP Photonics Encyclopedia**
- 5. Photonics Media
- 6. Optical Society (OSA) Webinar series on Lasers
- 7. Physics of Lasers University of Southampton
- 8. Nobel Prize in Physics 1964 Presentation Speech
- 9. Coursera Introduction to Lasers
- 10. Laser Classroom Lesson Plans

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/ participation/case studies/reflective journals (minimum of five components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40% marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER V					
AEC003	Arithmetic Reasoning-III	L	Τ	Р	С

Version 1.0		3	0	0	3
Category of Course	Ability Enhancement Course				
Total Contact Hours	45				
Pre-Requisites/ Co-					
Requisites					

Course Perspective:

The course aims to improve basic arithmetic skills, speed, and accuracy in mental calculations, and logical reasoning. These abilities are essential for a strong math foundation, helping students succeed in academics and various practical fields.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Understanding arithmetic algorithms required for solving mathematical problems.

CO2: Applying arithmetic algorithms to improve proficiency in calculations.

CO3: Analyzing cases, scenarios, contexts and variables, and understanding their interconnections in each problem.

CO4: Evaluating & deciding approaches and algorithms to solve mathematical & reasoning problems.

Course Content

Unit I: Mathematical Essentials12Hours

Traditional Indian Calculation methods, Number types and divisibility principles, Practical uses of Percentage in calculating changes and discount, understanding Ratio and Proportion in everyday context.

Unit II: Fundamentals of Logical Reasoning Blood Relations, Direction Sense, Coding Decoding

Unit III: Elementary Quantitative Skills

Simple and Compound Interest in everyday situations like loans, investment, Practical problems involving Averages, Real life examples and scenarios involving Partnership

Unit IV: Reasoning Skills

Introduction to reasoning, logical reasoning, Analytical reasoning, deductive reasoning, Inductive reasoning, Abductive reasoning, Reasoning in Communication, reasoning in decision making, Reasoning in Research and analysis

Learning Experience:

The course will be conducted using experiential and participatory methods such as practical problem-solving sessions, group discussions, and real-life applications. Students will practice traditional Indian calculation techniques and apply mathematical concepts like percentages, ratios, and interest to everyday situations. Logical and analytical reasoning skills will be developed through exercises in blood relations, direction sense, and reasoning techniques. Case studies and scenarios will help students apply reasoning to communication, decision-making, and research contexts, ensuring hands-on learning and active participation.

Textbooks:

- i. Guha Abhijit: Quantitative Aptitude for Competitive Examinations, Tata McGraw Hill Publication
- ii. Quantitative Aptitude by R.S. Aggarwal
- iii. Verbal & Non-Verbal Reasoning by R.S. Aggarwal

Suggested Readings:

- i. https://www.indiabix.com/online-test/aptitude-test/
- ii. https://www.geeksforgeeks.org/aptitude-questions-and-answers/
- iii. https://www.hitbullseye.com/

Evaluation Scheme:

9 hours

13 hours Practical

11hours

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER V					
SIPH002	EvaluationofSummerInternship/Project	L	T	Р	C
Version 1.0		2	0	0	2
Category of Course	Internship				
Total Contact Hours					
Pre-Requisites/ Co- Requisites	Practical Exposure				

Course Perspective:

In the end of Semester IV, students will be asked to join research/academic organizations or industries to get hands on knowledge on the selected topics. The student will work on the assigned topic for 3-4 weeks in regular consultation with his/her assigned expert/guide. The student will write a report based on the work carried out during internship and prepare two copies to be submitted to the office of the Head of the Department duly signed by the student and the expert. The student will make a power point presentation based on the work carried out and mentioned in the report to the board of examiners appointed by the University in the fifth semester. The student will be evaluated based on a report and presentation.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Carrying out the extensive literature survey on the topic assigned by academicians and industry experts.

CO 2: Applying various methods and techniques applicable to their research topic to study and contributing to domain knowledge.

CO 3: Analyzing the result of the experiment carried out and presenting the results using data visualization methods.

CO 4: Evaluating the effectiveness of methods used and the significance of research findings.

CO 5: Writing and presenting technical reports/articles.

Learning Experience:

The internship course will be experiential through hands-on lab work, real-world research projects, and active participation in ongoing studies. Students will collaborate with faculty and researchers, applying theoretical knowledge to experimental tasks and data analysis. Regular group discussions, progress presentations, and peer feedback will enhance collaborative learning. The course will also include reflective journaling to encourage self-assessment and growth throughout the internship.

Evaluation Scheme:

Particular	Weightage
Internal Marks: - (Punctuality, Performance, Work Ethics, Efforts and Research Output)	50 Marks
External Marks (Practical): -	50 Marks
Presentation	20
Report Writing/Dissertation	10
Viva Voce	20

SEMESTER VI					
SCPH302	Analog Systems and Applications	L	Т	Р	С
Version 1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Basics of Electronics				

Course Perspective

This course provides a focused study of semiconductor devices and their applications in electronic circuits. Starting with the fundamentals of PN junction diodes, students will explore their characteristics and applications in rectification and voltage regulation. The course then moves to bipolar junction transistors, covering their configurations, amplifier design, and analysis techniques. Emphasis is placed on feedback mechanisms and oscillator design, essential for understanding stability and signal generation in circuits. Finally, students will study operational amplifiers, learning their characteristics and diverse applications, including amplification and analog-to-digital conversion, equipping them with the skills to analyze and design basic electronic circuits.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding the experimental procedures and theoretical concepts by studying the key variables, equipment, and principles involved in each experiment.

CO2: Applying their theoretical knowledge by performing experiments, using laboratory equipment, and executing procedures accurately to obtain and interpret experimental results.

CO3: Analyzing experimental data and observations to identify trends, correlations, and anomalies, applying critical thinking to interpret results and derive meaningful conclusions.

CO4: Evaluating the effectiveness and accuracy of experimental methods and results, making informed decisions to refine techniques and improve overall experimental outcomes.

Course Contents

UNIT I Semiconductor Diodes

No. of Hours: 15

P and N type semiconductors. Energy Level Diagram. Conductivity and Mobility, Concept of Drift Velocity. PN Junction Fabrication (Simple Idea). Barrier Formation in PN Junction Diode. Static and Dynamic Resistance. Current Flow Mechanism in Forward and Reverse Biased Diode. Drift Velocity. Derivation for Barrier Potential, Barrier Width and Current for Step Junction. Current Flow Mechanism in Forward and Reverse Biased Diode.

Two-terminal Devices and their Applications: (1) Rectifier Diode: Half-wave Rectifiers. Centre-tapped and Bridge Full-wave Rectifiers, Calculation of Ripple Factor and Rectification Efficiency, C-filter (2) Zener Diode and Voltage Regulation. Principle and structure of (1) LEDs, (2) Photodiode and (3) Solar Cell.

UNIT II Transistors & Amplifiers

No. of Hours: 15

No. of Hours: 15

Bipolar Junction transistors: n-p-n and p-n-p Transistors. Characteristics of CB, CE and CC Configurations. Current gains α and β Relations between α and β . Load Line analysis of Transistors. DC Load line and Q-point. Physical Mechanism of Current Flow. Active, Cutoff and Saturation Regions.

Amplifiers: Transistor Biasing and Stabilization Circuits. Fixed Bias and Voltage Divider Bias. Transistor as 2-port Network. h-parameter Equivalent Circuit. Analysis of a single-stage CE amplifier using Hybrid Model. Input and Output Impedance. Current, Voltage and Power Gains. Classification of Class A, B & C Amplifiers.

Coupled Amplifier: Two stage RC-coupled amplifier and its frequency response.

UNIT III Feedback In Amplifier

Feedback in Amplifiers: Effects of Positive and Negative Feedback on Input Impedance, Output Impedance, Gain, Stability, Distortion and Noise.

Sinusoidal Oscillators: Barkhausen's Criterion for self-sustained oscillations. RC Phase shift oscillator, determination of Frequency. Hartley & Colpitts oscillators.

UNIT IV Operational Amplifiers

No. of Hours: 15

Operational Amplifiers (Black Box approach): Characteristics of an Ideal and Practical OpAmp. (IC 741) Open-loop and Closed-loop Gain. Frequency Response. CMRR. Slew Rate and concept of Virtual ground.

Applications of Op-Amps: (1) Inverting and non-inverting amplifiers, (2) Adder, (3) Subtractor, (4) Differentiator, (5) Integrator, (6) Log amplifier, (7) Zero crossing detector (8) Wein bridge oscillator.

Conversion: Resistive network (Weighted and R-2R Ladder). Accuracy and Resolution. A/D Conversion (successive approximation).

Learning Experience

The learning experience in this course will be both theoretical and hands-on, providing students with a robust understanding of semiconductor devices and their applications. Through detailed lectures and practical labs, students will explore the operation of diodes and transistors, analyze amplifier circuits, and design feedback systems and oscillators. Interactive exercises and real-world problem-solving will enhance their ability to apply theoretical concepts to practical circuit design and analysis. By working with operational amplifiers and various electronic components, students will develop the skills needed to tackle complex electronic challenges and gain a solid foundation for advanced studies in electronics.

Textbooks

- Robert Boylestad, Louis Nashelsky, Electronic Devices and Circuit Theory, 8Th Edition, Pearson Education, India.
- 2. Albert P. Malvino, David J. Bates. Electronic Principles, Eighth Edition, McGraw-Hill Education, United States.

Suggested Readings

- 1. Electronic Communication, Rudy and Cohlen (Prentice Hall).
- 2. Semiconductor Devices Physics & Technology by S. M. Sze (John Wiley).

Open Educational Resources (OER)

- 1. <u>https://www.electronics-tutorials.ws/</u>
- 2. https://nptel.ac.in/courses/108102112
- 3. https://nptel.ac.in/courses/108105158

- 4. <u>https://www.learningelectronics.net/</u>
- 5. <u>https://www.youtube.com/watch?v=XG3cVoUh7wc&list=PLs5_Rtf2P2r674CTMNJ3ode</u> <u>Hk9Wtb-WWl</u>
- 6. <u>https://www.youtube.com/watch?v=EdUAecpYVWQ&list=PLwjK_iyK4LLBVM18VZ7</u> <u>JKW-q88FAtnr8_</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/ participation/case studies/reflective journals (minimum of five components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40% marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VI					
SCPH352	Analog Systems and Applications Lab	L	T	Р	С
Version1.0		0	0	4	2
Category of Course	Major Lab	<u> </u>	<u> </u>	<u> </u>	<u> </u>

Total Contact Hours	30
Pre-Requisites/ Co- Requisites	Basics of Electronics

Course Perspective

This course offers a comprehensive exploration of electronic circuit design and analysis through practical experimentation. Students will study and analyze the *V-I* characteristics of Zener diodes and their application in voltage regulation and evaluate the performance of solar cells by examining their *V-I* and power curves to determine efficiency and maximum power points. They will gain hands-on experience with Bipolar Junction Transistors (BJTs) in various configurations, including class A operation, and design CE amplifiers with specific gain requirements. The course also covers the design and analysis of oscillators, digital-to-analog converters (DACs), and precision differential amplifiers using operational amplifiers. Students will further investigate the use of op-amps as integrators and differentiators and apply these concepts to solve simultaneous and differential equations. Through these practical projects, students will develop a deep understanding of electronic components and circuit design, preparing them for advanced work in electronics and electrical engineering.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Observing and interpreting the *V*-*I* characteristics of various electronic components, such as Zener diodes and solar cells, to understand their behaviour and applications in circuits.

CO2: Imitating established design methodologies for constructing and analyzing electronic circuits, including amplifiers, oscillators, and digital-to-analog converters, following precise experimental procedures.

CO3: Practicing, designing and testing electronic circuits, including BJT amplifiers and precision differential amplifiers, to achieve specific performance criteria and gain hands-on experience with circuit analysis tools.

Course Contents

At least 8 experiments from the following:

- 1. To study the V-I characteristics of a Zener diode and its use as voltage regulator.
- 2. Study of V-I & power curves of solar cells and find maximum power point & efficiency.
- 3. To study the characteristics of a Bipolar Junction Transistor in CE configuration.
- 4. To study the various biasing configurations of BJT for normal class A operation.
- 5. To design a CE transistor amplifier of a given gain (mid-gain) using voltage divider bias.
- 6. To study the frequency response of voltage, gain of a two stage RC-coupled transistor amplifier.
- 7. To design a Wien bridge oscillator for given frequency using an op-amp.
- 8. To design a phase shift oscillator of given specifications using BJT.
- 9. To design a digital to analog converter (DAC) of given specifications.
- 10. To design a precision Differential amplifier of given I/O specification using Op-amp.
- 11. To investigate the use of an op-amp as an Integrator.
- 12. To investigate the use of an op-amp as a Differentiator.
- To design a circuit to simulate the solution of simultaneous equation and 1st/2nd order differential equation.

Learning Experience

The learning experience in this course will be hands-on and interactive, combining theoretical knowledge with practical experiments. Students will engage in a range of laboratory activities, including analyzing the *V-I* characteristics of Zener diodes and solar cells, designing and testing BJT amplifiers, and building various oscillators and converters. They will gain practical skills in circuit design and performance evaluation, using tools such as oscilloscopes and function generators to measure and analyze circuit behaviour. By working on diverse projects like designing precision amplifiers and solving differential equations with op-amps, students will develop a robust understanding of electronic circuits and enhance their problem-solving abilities in real-world applications.

Textbooks

 Basic Electronics: A text lab manual, P.B. Zbar, A.P. Malvino, M.A. Miller, 1994, Mc-Graw Hill. OP-Amps

Open Educational Resources (OER)

- 1. <u>https://www.youtube.com/watch?v=BxbXnYFqygU&pp=ygVTdG8gc3R1ZHkgdGhlIHY</u> <u>taSBjaGFyYWN0ZXJpc3RpY3Mgb2YgYSB6ZW5lciBkaW9kZSBhbmQgaXRzIHVzZS</u> <u>BhcyB2b2x0YWdlIHJlZ3VsYXRvciA%3D</u>
- 2. <u>https://www.youtube.com/watch?v=9GuB1SgHXEc&pp=ygVXLiBTdHVkeSBvZiBWLUkgJiBwb3dlciBjdXJ2ZXMgb2Ygc29sYXIgY2VsbHMsIGFuZCBmaW5kIG1heGltdW0gcG93ZXIgcG9pbnQgJiBlZmZpY2llbmN5</u>
- 3. <u>https://www.youtube.com/watch?v=BsbqB07Mwmo&pp=ygVSLlRvIHN0dWR5IHRoZS</u> <u>BjaGFyYWN0ZXJpc3RpY3Mgb2YgYSBCaXBvbGFyIEp1bmN0aW9uIFRyYW5zaXN</u> <u>0b3IgaW4gQ0UgY29uZmlndXJhdGlvbg%3D%3D</u>
- 4. <u>https://www.youtube.com/watch?v=E58rxqJB5kA&list=PLC1WV1OE4n9gVU3bQfBW</u> <u>sQhnmICRKYM</u>
- 5. <u>https://www.youtube.com/watch?v=ZZomhoZL18&pp=ygVbVG8gZGVzaWduIGEgQ0</u> <u>UgdHJhbnNpc3RvciBhbXBsaWZpZXIgb2YgYSBnaXZlbiBnYWluIChtaWQtZ2Fpbikg</u> <u>dXNpbmcgdm9sdGFnZSBkaXZpZGVyIGJpYXMuIA%3D%3D</u>
- 6. <u>https://www.youtube.com/watch?v=QAHsqK8La1c&pp=ygVhVG8gc3R1ZHkgdGhlIGZyZXF1ZW5jeSByZXNwb25zZSBvZiB2b2x0YWdlIGdhaW4gb2YgYSB0d28gc3RhZ2UgUkMtY291cGxlZCB0cmFuc2lzdG9yICBhbXBsaWZpZXIuIA%3D%3D</u>
- 7. <u>https://www.youtube.com/watch?v=eg884iSHGok&pp=ygVGVG8gZGVzaWduIGEgcG</u> <u>hhc2Ugc2hpZnQgb3NjaWxsYXRvciBvZiBnaXZlbiBzcGVjaWZpY2F0aW9ucyB1c2luZ</u> <u>yBCSlQuIA%3D%3D</u>
- https://www.youtube.com/watch?v=_TD0xG5FbCs&pp=ygVIVG8gZGVzaWduIGEgV21 lbiBicmlkZ2Ugb3NjaWxsYXRvciBmb3IgZ2l2ZW4gZnJlcXVlbmN5IHVzaW5nIGFuIG 9wLWFtcC4g
- 9. <u>https://www.youtube.com/watch?v=upvw2uLJDeI&pp=ygVFVG8gZGVzaWduIGEgZG1</u> <u>naXRhbCB0byBhbmFsb2cgY29udmVydGVyIChEQUMpIG9mIGdpdmVuIHNwZWNpZ</u> <u>mljYXRpb25z</u>
- 10. <u>https://www.youtube.com/watch?v=2Gb16xH6g7g&pp=ygVXIFRvIGRlc2lnbiBhIHByZ</u> <u>WNpc2lvbiBEaWZmZXJlbnRpYWwgYW1wbGlmaWVyIG9mIGdpdmVuIEkvTyBzcG</u> <u>VjaWZpY2F0aW9uIHVzaW5nIE9wLWFtcC4g</u>
- 11. <u>https://www.youtube.com/watch?v=PRPkvGjUFO4&pp=ygVKIFRvIGludmVzdGlnYXR</u> <u>lIHRoZSB1c2Ugb2YgYW4gb3AtYW1wIGFzIGFuIEludGVncmF0b3IgYW5kIERpZmZ</u> <u>lcmVudGlhdG9yLiA%3D</u>
- 12. https://www.youtube.com/shorts/nxThhTE1fVg

Evaluation Scheme

Evaluation components	Weightage
Internal marks (practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical)	50 Marks
End Term Examination	

It is compulsory for a student to secure 40% marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VI					
SCPH304	Nuclear Physics	L	Т	Р	С
		4	0	0	4
Version1.0					
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co- Requisites	Basic Physics				

Course Perspective This course offers a comprehensive study of nuclear physics, focusing on the fundamental properties and behaviours of atomic nuclei, radioactivity, and nuclear reactions. Students will delve into the intrinsic properties of nuclei, including mass, charge density, and binding energy, and explore various nuclear models to understand nuclear stability. The course also covers the principles of radioactive decay processes—alpha, beta, and gamma decay—providing students with theoretical and practical insights into nuclear transformations. Additionally, students will examine the interaction of nuclear radiation with matter, including ionization and radiation detection methods. Emphasis is placed on both theoretical concepts and practical applications, equipping students with the knowledge and skills to analyze and measure nuclear phenomena effectively and prepare them for advanced studies and careers in nuclear science and related fields.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Understanding general properties of nuclei, including constituents, mass, radii, charge density, binding energy, and the main features of nuclear models like the liquid drop model and semi-empirical mass formula.

CO2: Applying their understanding of radioactivity to analyze α -decay, β -decay, and γ -decay processes, including using decay theories and kinematics to solve related problems.

CO3: Analyzing nuclear reactions by applying conservation laws, calculating Q-values, and evaluating reaction rates and cross sections, distinguishing between compound and direct reactions as well as resonance reactions.

CO4: Evaluating the performance and efficiency of different nuclear radiation detectors, such as ionization chambers, GM counters, and scintillation detectors, based on their principles of operation and practical applications.

Course Content

UnitI:NucleiProperties&NuclearModel15 Contact Hours

General Properties of Nuclei: Constituents of nucleus and their Intrinsic properties, quantitative facts about mass, radii, charge density (matter density), binding energy, average binding energy and its variation with mass number, main features of binding energy versus mass number curve, N/A plot, angular momentum, parity, magnetic moment, electric moments, nuclear excites states.

Nuclear Models: Liquid drop model approach, semi empirical mass formula and significance of its various terms, condition of nuclear stability.

Unit II: Radioactivity decay

Alpha decay: basics of α -decay processes, theory of α -emission, Gamow factor, Geiger Nuttall law, α -decay spectroscopy. (b) β -decay: energy kinematics for β - decay, positron emission,

15 Contact Hours

electron capture, neutrino hypothesis. (c) Gamma decay: Gamma rays emission & kinematics, internal conversion.

UnitIII:InteractionofNuclearRadiationwithmatter15 Contact Hours

Nuclear Reactions: Types of Reactions, Conservation Laws, kinematics of reactions, Q-value, reaction rate, reaction cross section, Concept of compound and direct Reaction, resonance reaction,

Interaction of Nuclear Radiation with matter: Energy loss due to ionization (Bethe-Block formula), energy loss of electrons, Cerenkov radiation. Gamma ray interaction through matter, photoelectric effect, Compton scattering, pair production, neutron interaction with matter.

UnitIV:DetectorforNuclearRadiations15 Contact Hours

Gas detectors: estimation of electric field, mobility of particle, for ionization chamber and GM Counter. Basic principle of Scintillation Detectors and construction of photo-multiplier tube (PMT). Semiconductor Detectors (Si and Ge) for charge particle and photon detection (concept of charge carrier and mobility), neutron detector

Learning Experience

The course will be delivered through a combination of lectures, and interactive demonstration sessions to ensure a comprehensive understanding of nuclear physics concepts and techniques. Group discussions, problem-solving exercises, and project-based assessments will facilitate collaborative learning and the application of concepts to real-world scenarios. Regular quizzes, class test and assignments will assess students' grasp of the material and their ability to apply theoretical principles. Additional support will be available through office hours, study groups, and online resources, promoting an interactive and supportive learning environment.

Textbooks:

1. Nuclear Physics by S N Ghoshal, First edition, S. Chand Publication, 2010.

Suggested Readings:

- 1. Introductory nuclear Physics by Kenneth S. Krane (Wiley India Pvt. Ltd., 2008).
- 2. Introduction to Elementary Particles, D. Griffith, John Wiley & Sons

Open Educational Resources (OER)

- 1. <u>https://www.youtube.com/watch?v=MnPJgXyXHW8</u>
- 2. <u>https://www.youtube.com/watch?v=RTIThUySwUE</u>
- 3. <u>https://www.youtube.com/watch?v=Rd0CJje59bE</u>
- 4. https://www2.lbl.gov/abc/wallchart/chapters/03/2.html

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VI					
SCPH306	Electromagnetic Theory	L	Т	Р	С
Version1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co-	Basic Electrodynamics, Calculus				
Requisites					

Course Perspective This course provides a comprehensive introduction to the fundamental principles of electromagnetism, crucial for any physics student. Understanding electromagnetic

theory is essential for pursuing advanced studies in physics, engineering, and related fields. This course will help students develop problem-solving skills and a deep understanding of physical laws governing electric and magnetic fields. It is applicable in various real-world contexts, including electrical engineering, telecommunications, and material science. Students will learn to apply theoretical concepts to practical situations, such as designing electrical circuits and understanding electromagnetic waves.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and listing Maxwell's equations and fundamental concepts of electromagnetism.

CO2: Solving practical problems related to electromagnetic wave propagation in different media using appropriate mathematical techniques.

CO5: Analyzing the laws of electromagnetic induction, including Faraday's and Lenz's laws, and the role of Maxwell's equations in the description of electromagnetic waves and energy flow.

CO4: Evaluating and dissecting complex boundary value problems involving electric and magnetic fields to identify underlying principles and patterns.

Course Content

Unit 1: Electrostatics

(12 Hours)

Electric Field and Potential: Definition of electric field and electric potential. Calculation of electric field due to point charges, dipoles, and continuous charge distributions. Electric field lines, equipotential surfaces. Work done by electric forces, electric potential energy. Gauss's Law: Statement of Gauss's law, its application to various symmetrical charge distributions (plane, spherical, and cylindrical symmetries). Differential and integral forms of Gauss's law. Electric flux, Gauss's law in dielectric media. Capacitance and Dielectrics: Capacitors and capacitance, calculation of capacitance for simple geometries (parallel plate, spherical, cylindrical). Energy stored in a capacitor. Dielectrics and their properties, electric susceptibility, dielectric constant, energy in the presence of dielectrics.

Unit 2: Magnetostatics

(12 Hours)

(12 Hours)

Magnetic Field and Magnetic Forces: Biot-Savart law, calculation of magnetic field due to currents (straight wire, circular loop, solenoid). Lorentz force law, force on a moving charge in a magnetic field. Magnetic dipole moment, torque on a magnetic dipole. Ampere's Law: Ampere's circuital law and its applications to various geometries (infinite wire, solenoid, toroid). Differential and integral forms of Ampere's law. Magnetic vector potential. Magnetic Materials: Magnetization, magnetic susceptibility, and permeability. Diamagnetic, paramagnetic, and ferromagnetic materials. Hysteresis and magnetic domains.

Unit 3: Electrodynamics

Faraday's Law of Induction: Faraday's law and Lenz's law. Induced EMF and electric fields. Self-inductance and mutual inductance. Energy stored in magnetic fields, inductors in circuits. **Maxwell's Equations**: Derivation and physical interpretation of Maxwell's equations in free space and in matter. Displacement current, continuity equation, and the modification of Ampere's law. Poynting vector and the flow of electromagnetic energy. **Electromagnetic Waves**: Wave equation in free space, plane electromagnetic waves. Propagation of electromagnetic waves in vacuum and in dielectrics. Energy and momentum of electromagnetic waves, radiation pressure.

Unit 4: Applications of Electromagnetic Theory (12 Hours)

Radiation from Accelerated Charges: Larmor formula for power radiated by an acceleratedcharge. Radiation from a dipole antenna, Hertzian dipole. Electromagnetic radiation and itsspectrum.Relativistic Electrodynamics: Introduction to special relativity, Lorentztransformations.Relativistic transformation of electric and magnetic fields. Covariantformulation of Maxwell's equations.Applications of relativistic electrodynamics, relativisticmotioninelectromagneticfields.fields.

Learning Experience

The course will be conducted through a mix of lectures, interactive sessions, and problemsolving workshops. Technology such as online resources will be used to enhance learning. Students will engage in group work, case studies, and practical assignments. The course will also include laboratory sessions to provide hands-on experience with electromagnetic measurements. Regular feedback and support will be available through office hours, and peer reviews will foster collaborative learning.

Textbooks

- 1. "Introduction to Electrodynamics" by David J. Griffiths, 4th Edition, Pearson (2017)
- 2. "Classical Electrodynamics" by John David Jackson, 3rd Edition, Wiley (1998)

Suggested Readings

- "Electricity and Magnetism" by Edward Purcell and David Morin, 3rd Edition, Cambridge University Press (2013)
- 2. "Principles of Electrodynamics" by Melvin Schwartz, Dover Publications (2013)

Open Educational Resources (OER)

- MIT Open Course Ware Electromagnetic Theory: https://ocw.mit.edu/courses/physics/8-02-electromagnetism-spring-2016/Khan Academy - Electricity and Magnetism:
- 2. https://www.khanacademy.org/science/physics/electricity-and-magnetism

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER VI					
SCPH308	Atomic and Molecular Physics	L	Т	Р	C
Version1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	60				
Pre-Requisites/ Co-	Basics of Atomic Structure				
Requisites					

Course Perspective

The course "Atomic and Molecular Physics" is integral to the Physics curriculum, serving as a foundational pillar in understanding the behaviour of atoms and molecules, which are essential in various fields of physics, chemistry, and material science. Students will gain a deep understanding of atomic models, quantum mechanics, and molecular structures, which are critical for careers in research, technology development, and advanced studies in physical sciences. This course enables students to apply theoretical knowledge to real-world problems such as spectroscopy, laser technology, and quantum computing. For instance, knowledge of atomic spectra is essential in fields like astrophysics and the development of new materials. By mastering the concepts taught in this course, students will be well-prepared to engage in cutting-edge research and innovation in both academic and industrial settings.

Course Outcomes

Upon completion of the course the learner will be:

CO1: Understanding the fundamental models of the atom, including Rutherford, Bohr, Sommerfeld, and quantum models, and their applications in explaining atomic spectra.

CO2: Applying quantum mechanical principles to analyze the effects of magnetic fields on atomic spectra and understand phenomena such as the Zeeman effect.

CO3: Analyzing the production, characteristics, and applications of X-rays, including Laue's experiment, Bragg's law, and X-ray absorption spectra.

CO4: Evaluating the quantum statistical approaches for understanding molecular bonding, rotational and vibrational energy levels, and their applications in spectroscopy.

Course Content

Unit I: Atom Models

15 contact hours

15 contact

Rutherford scattering experiment and the nuclear model of the atom, size of the nucleus, atomic spectra and spectral series. Bohr model of the atom: energy levels and spectral series, line spectra, discovery of deuterium, correspondence principle, nuclear (reduced) mass and its effect of the atomic spectra: discovery of deuterium, positronium and muonic atom energy levels compared to hydrogen energy levels, critical potentials, atomic excitation, Franck-Hertz experiments. Sommerfeld relativistic model and fine structure of hydrogen.Quantum (Vector) model of the hydrogen atom (no derivation) and quantum numbers, principal quantum number, orbital quantum number, magnetic quantum number, probabilistic electronic orbits (radial and angular), radiative transitions, selection rules.

Unit II: Many Electron Atoms hours

Effect of Magnetic Fields and Many Electron Atoms: Normal Zeeman effect, gyro-magnetic ratio, Bohr magneton, spin of the electron, spin angular momentum, magnetic dipole moments due to orbital motion and spin of the electron, exclusion principle, Stern-Gerlach experiment. Symmetric and anti-symmetric wave functions, bosons and fermions, atomic shells, subshells and periodic table Spin-orbit coupling, anomalous Zeeman effect, Paschen-Back effect, Stark effect, total angular momentum, LS coupling, j-j coupling, singlet, doublet, triplet, term symbols. Atomic spectra of hydrogen and sodium.

Unit III: X-Rays & Laser

X-Ray Spectra: X-rays: production, Laue's experiment, Bragg's law, X-ray spectra: continuous and characteristic spectra, Mosley's law and X-ray series, Auger effect, X-ray absorption spectra, absorption edges.

Lasers

Einstein's A and B coefficients, Metastable states, Spontaneous and Stimulated emissions, Optical Pumping and Population Inversion, Three-Level and Four-Level Lasers, Ruby Laser and He-Ne Laser,

Unit IV: Molecular Physics

15 contact hours

15 contact hours

Molecular Physics: Molecular bond, covalent bond, H2+ molecular ion, Hydrogen molecule, complex molecules, hybrid orbitals: ethylene, benzene, Rotational Energy levels, Selection Rules and Pure Rotational Spectra of a Molecule, Vibrational Energy Levels, Selection Rules and Vibration Spectra, Rotation-Vibration Energy Levels, Selection Rules and Rotation-Vibration Spectra, Determination of Internuclear Distance, electronic spectra: fluorescence, phosphorescence.

Raman Effect: Quantum Theory of Raman Effect, Characteristics of Raman Lines, Stoke's and Anti-Stoke's Lines, Complimentary Character of Raman and infrared Spectra;'

Learning Experience This course offers a comprehensive learning experience through a combination of lectures, hands-on labs, and real-world case studies. Students will engage in practical experiments to apply theoretical concepts. Group projects and collaborative activities will enhance teamwork and problem-solving skills. Regular assessments and assignments will ensure students' understanding and application of key topics. Support and feedback will be provided throughout, encouraging independent learning and continuous improvement.

Textbooks

 Modern Physics, R. Murugeshan and Kiruthiga Sivaprasath, 17th Ed., S. Chand & Company Pvt. Ltd.

2. Atomic and Molecular Spectra: Laser, Rajkumar, 2020, Knrn

Suggested Readings

- 1. Concepts of Modern Physics by Arthur Beiser (McGraw-Hill Book Company, 1987)
- 2. Atomic physics by J,B,Rajam& foreword by Louis De Broglie, (S,Chand& Co,, 2007),
- 3. Atomic Physics by J,H,Fewkes& John Yarwood, Vol, II (Oxford Univ, Press, 1991),
- 4. Physics of Atoms and Molecules, Bransden and Joachein,
- 5. Molecular Spectroscopy, Banwell,
- 6. Optoelectronics by Ghatak and Thyagarajan, Principles of Lasers by Svelto

Open Educational Resources (OER)

- 1. https://www.geeksforgeeks.org/rutherfords-alpha-scattering-experiment/
- 2. http://labs.plantbio.cornell.edu/wayne/pdfs/Fine%20structure.pdf

- 3. <u>https://en.wikipedia.org/wiki/Gyromagnetic_ratio#:~:text=In%20physics%2C%20the%20</u> gyromagnetic%20ratio,by%20the%20symbol%20%CE%B3%2C%20gamma.
- 4. <u>https://www.youtube.com/watch?v=_1S6KfMzOH8</u>
- 5. https://www.radiologymasterclass.co.uk/tutorials/physics/x-ray_physics_production
- https://eng.libretexts.org/Bookshelves/Electrical_Engineering/Electro Optics/Direct_Energy_(Mitofsky)/07%3A_Lamps%2C_LEDs%2C_and_Lasers/7.01%3
 A_Absorption%2C_Spontaneous_Emission%2C_Stimulated_Emission
- <u>https://chem.libretexts.org/Bookshelves/Physical_and_Theoretical_Chemistry_Textbook_</u> <u>Maps/Book%3A_Quantum_States_of_Atoms_and_Molecules_(Zielinksi_et_al)/07%3A_</u> <u>Rotational_States</u>

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VII					
SCCH401	Research Methodology	L	Τ	Р	С
Version 1.0		4	0	0	4
Category of Course	Major			1	
Total Contact Hours	60				
Pre-Requisites/ Co	-				
Requisites					

Course Perspective

Research Methodology equips students with essential tools and techniques for conducting scientific research. It covers the systematic process of identifying a problem, formulating a hypothesis, collecting data, and analyzing results. By understanding various research designs, methods of data collection, and statistical analysis, students develop the ability to critically evaluate existing research and contribute original findings. The course emphasizes ethical considerations, ensuring that research is conducted with integrity and responsibility. Ultimately, Research Methodology fosters a deeper understanding of the scientific process, enabling students to approach problems methodically and contribute to academic and professional knowledge.

Course Outcomes

Upon completion of the Research Methodology course, the learner will be:

CO1: Understanding and explaining fundamental concepts of research methodology, including its meaning, objectives, and utility, as well as key research terminology such as concept, construct, definition, and variable.

CO2: Differentiating between empirical and theoretical research and apply deductive and inductive reasoning to various research scenarios.

CO3: Applying various sampling techniques and implement effective data preparation methods, including univariate and bivariate analysis, in research projects.

CO4: Analyzing different research designs, such as exploratory, descriptive, and experimental, and evaluate the importance of qualitative and quantitative research, including concepts like causality, generalization, and replication.

Course Content

Unit I: Introduction Basic Fundamentals

No. of Hours: 10

Meaning, objective, motivation and utility of research. Concept of theory, empiricism, deductive and inductive reasoning. Characteristics of scientific methods – Understanding the language of research-Concept, Construct, Definition, Variable.

Unit II: Research Design: No. of Hours: 18

Concept and Importance of research design. Features of a good research design - Exploratory Research Design - concept, types and uses, Descriptive Research Designs - concept, types and uses. Experimental Research Design: Concept of Independent & Dependent variables. Qualitative research - Quantitative research - Concept of measurement, causality, generalization, replication. Merging the two approaches.

Unit III: Sampling and Data Preparation Sampling: No. of Hours: 12

Concepts of Statistical Population, Sample, Sampling Frame, Sampling Error, Sample Size, Non-Response. Characteristics of a good sample. Probability Sample- Simple Random Sample, Systematic Sample, Stratified Random Sample & Multi-stage sampling. Data Preparation -Univariate analysis (frequency tables, bar charts, pie charts, percentages), Bivariate analysis Cross tabulations and Chi-squares test including testing hypothesis of association. Interpretation of Data.

Unit IV: Paper Writing and Publishing Process Paper Writing- No. of Hours: 10

Layout of a Research Paper, Journals, Impact factor of Journals, Choosing Journals and Conferences to publish the work. Ethical issues related to publishing, Plagiarism and Self Plagiarism. Preparing response to reviewers and editors. Reviewing the manuscript.

Unit V: Use of various resources and software No. of Hours: 10

Use of encyclopaedia, handbooks, research guides and academic databases. Use of tools and techniques for Research, methods to search required information effectively. Reference Management Software like Zotero/ Mendeley. Software for paper formatting like Latex, LyX, MS Office. Software for detection of Plagiarism.

Learning Experience

This course will integrate lectures, interactive sessions, and hands-on research projects to provide students with a comprehensive understanding of research methodology and its application across various disciplines.

Instruction Methods:

- Lectures: Fundamental concepts of research methodology, including research design, data collection, data analysis, and report writing, will be taught using case studies, multimedia presentations, and examples from contemporary research.
- Interactive Sessions: Students will engage in Q&A sessions, group discussions, and peer critiques to explore research questions, methodologies, and ethical considerations, fostering critical thinking and collaborative learning.

Technology Use:

• Online Platforms: A Learning Management System (LMS) will be utilized to provide access to course materials, recorded lectures, assignment submissions, and discussion forums, promoting extended learning and peer-to-peer interaction.

Assessments:

- Formative: Continuous assessments through quizzes, research proposals, and participation in online discussions will provide feedback and help students monitor their understanding and progress.
- **Summative:** Final assessments will include research project presentations, written reports, and peer evaluations, focusing on students' ability to apply research methodologies in practical scenarios.

Support:

• Instructor Guidance: The course instructor will offer additional support through scheduled office hours, and students will be encouraged to collaborate through group work and peer review sessions. Continuous feedback will be provided to ensure students' development in meeting course objectives.

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Textbooks

1. Panneerselvam, R., Research Methodology, Prentice Hall of India, New Delhi, 2004.

2. Kothari CR, Research Methodology-Methods and Techniques, New Wiley Eastern Ltd., Delhi, 2009.

3. Brymam Alan and Bell, Emma, Business Research Methods

Suggested Readings

- "Research Methodology: A Step-by-Step Guide for Beginners" by Ranjit Kumar (ISBN: 9781446269978, 2014).
- "Research Design: Qualitative, Quantitative, and Mixed Methods Approaches" by John W. Creswell and J. David Creswell (ISBN: 9781506386706, 2017).
- "Qualitative Inquiry and Research Design: Choosing Among Five Approaches" by John W. Creswell (ISBN: 9781506330204, 2016).
- "Introduction to the Practice of Statistics" by David S. Moore, George P. McCabe, and Bruce A. Craig (ISBN: 9781319013387, 2017).

Open Educational Resources (OER)

https://oercommons.org/courseware/lesson/25010 https://www.oercommons.org/courses/research-methods-knowledge-base http://www.youtube.com/watch?v=2FYm3GOonhk http://www.youtube.com/watch?v=PtjxtG2FcaQ http://www.youtube.com/watch?v=A1URJE7x7ec

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessments (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

SEMESTER VII

SEC076	Technology in Research	L	Τ	Р	С
Version 1.0		2	0	0	2
Category of Course	Skill Enhancement Course	1		1	
Total Contact Hours	30				
Pre-Requisites/ Co	-				
Requisites					

Course Perspective

Research Methodology equips students with essential tools and techniques for conducting scientific research. It covers the systematic process of identifying a problem, formulating a hypothesis, collecting data, and analyzing results. By understanding various research designs, methods of data collection, and statistical analysis, students develop the ability to critically evaluate existing research and contribute original findings. The course emphasizes ethical considerations, ensuring that research is conducted with integrity and responsibility. Ultimately, Research Methodology fosters a deeper understanding of the scientific process, enabling students to approach problems methodically and contribute to academic and professional knowledge.

Course Outcomes

Upon completion of the Research Methodology course, the learner will be:

CO1: Understanding how various technologies enhance data collection, analysis, and collaboration in research, demonstrating an understanding of their impact on the research process.

CO2: Applying various technological tools and methods to design and conduct research projects, effectively integrating technology into their research workflows.

CO3: Analyzing the effectiveness of different technological tools in optimizing various stages of the research process, evaluating their impact on data quality and research outcomes.

CO4: Critically evaluating the strengths and limitations of different technological approaches in research, assessing their suitability for specific research objectives and contexts.

Course Content

Unit I Introduction to Technology in Research

Overview of Technology's Role in Research: Historical perspective, current trends, and future directions; Types of Research Technologies: Data collection tools (e.g., surveys, sensors), data management systems, and analysis software; Ethical and Practical Considerations: Data privacy, security, and responsible use of technology.

Unit II: Data Collection and Management

Advanced Data Collection Methods: Digital surveys, remote sensing, IoT devices; Data Storage Solutions: Cloud computing, databases, and data warehousing; Data Quality and Integrity: Techniques for ensuring accurate and reliable data collection.

Unit III: Data Analysis and Visualization

Analytical Tools and Techniques: Statistical software, machine learning algorithms, and AI applications; Data Visualization: Tools and methods for creating effective charts, graphs, and interactive visualizations; Interpreting and Presenting Data: Techniques for making data-driven decisions and communicating findings.

Unit IV: Innovation and Application in Research No. of Hours: 10

Emerging Technologies: Blockchain, augmented reality (AR), virtual reality (VR) in research; Developing and Implementing Technological Solutions: Designing research methodologies and tools using advanced technology; Case Studies and Practical Applications: Real-world examples of technology applied in various research fields and hands-on project development.

Learning Experience:

This course on will be conducted through a mix of interactive lectures, hands-on workshops, and real-world case studies to ensure experiential learning. Students will explore the role of technology in research, learning to use advanced tools for data collection, management, and analysis. Practical sessions will involve using statistical software, AI applications, and data visualization techniques, allowing students to apply these tools to real research scenarios. The course will also include group discussions, collaborative projects, and case studies, fostering participatory learning and innovation in applying emerging technologies like blockchain and AR/VR in research.

Textbooks

No. of Hours: 5

No. of Hours: 10

No. of Hours: 5

Suggested Readings Open Educational Resources (OER)

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessments (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

SEMESTER VII					
SEC077	Research Ethics and Report Writing	L	Τ	Р	С
Version 1.0		2	0	0	2
Category of Course	Skill Enhancement Course				
Total Contact Hours	30				
Pre-Requisites/ Co	-				
Requisites					

Course Perspective

Research Methodology equips students with essential tools and techniques for conducting scientific research. It covers the systematic process of identifying a problem, formulating a hypothesis, collecting data, and analyzing results. By understanding various research designs, methods of data collection, and statistical analysis, students develop the ability to critically evaluate existing research and contribute original findings. The course emphasizes ethical considerations, ensuring that research is conducted with integrity and responsibility.

Ultimately, Research Methodology fosters a deeper understanding of the scientific process, enabling students to approach problems methodically and contribute to academic and professional knowledge.

Course Outcomes

Upon completion of this course, the student will be:

CO1: Understanding key concepts in research ethics and recall significant ethical guidelines.

CO2: Applying ethical principles to research design and utilize proper citation techniques.

CO3: Analyzing ethical dilemmas in research case studies and break down the structure of research reports.

CO4: Evaluating research proposals for ethical integrity and assess data management practices.

CO5: Developing ethically sound research proposals and compose well-structured research reports.

Course Content

Unit 1: Introduction to Research Ethics

No. of Hours 5

- Definition and importance of research ethics
- Historical perspectives on research ethics
- Core ethical principles: autonomy, beneficence, non-maleficence, and justice
- Ethical issues in various research contexts

Unit 2: Ethical Issues in Research Design and Data ManagementNo. of Hours: 5

- Informed consent, confidentiality, and privacy
- Ethical considerations in qualitative vs. quantitative research
- Role and function of Institutional Review Boards (IRBs)
- Ethical data management: collection, storage, and sharing
- Avoiding research misconduct: fabrication, falsification, and plagiarism
- Authorship and intellectual property rights

Unit 3: Research Proposal Writing and Report Structuring No. of Hours: 10

- Components of a research proposal
- Ethical considerations in proposal writing

• Structuring a research report: Introduction, Literature Review, Methodology, Results, Discussion, Conclusion

• Language, tone, and style in academic writing

Unit 4: Citation, Peer Review, and Presentation of Research No. of Hours: 10

- Importance of proper citation and common citation styles (APA, MLA, Chicago)
- Avoiding plagiarism through correct attribution
- Understanding the peer review process
- Techniques for self-editing and revising research reports
- Preparing and presenting research findings
- Ethical issues in presenting research: avoiding bias and misrepresentation
- Case studies in research ethics: group discussions and analysis

Learning Experience

This course will be conducted through interactive lectures, discussions, and case studies, making the learning process experiential and participatory. Students will engage in real-world scenarios to understand ethical issues, participate in group discussions to explore research dilemmas, and work on writing exercises to develop research proposals and reports. Hands-on activities like peer reviews and presentations will allow students to apply ethical principles in research design, data management, and academic writing. The course emphasizes active learning through debates, case study analysis, and collaborative projects.

Recommended Textbooks and Readings:

- 1. "Research Ethics: A Reader" by David B. Resnik
- "The Craft of Research" by Wayne C. Booth, Gregory G. Colomb, and Joseph M. Williams
- 3. Selected journal articles and case studies provided by the instructor

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessments (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

SEMESTER VIII ((FOR B.Sc. (honours))				
SCPH402	ADVANCED MECHANICS-II	QUANTUM	L	T	P	C
Version 1.0			4	0	0	4
Category of Course	Major					
Total Contact Hours	55					
Pre-Requisites/ Co- Requisites	Quantum Mechanics/	Mathematical	Physi	cs		

Course Perspective

This course equips students with both theoretical and practical tools needed to solve complex problems in quantum mechanics, preparing them for higher-level courses and research. Students should learn this course to gain a strong foundation in quantum mechanics, which is essential for careers in physics, engineering, and materials science. The course will enhance technical competence, problem-solving and analytical skills of the students. The Knowledge of quantum mechanics is essential for developing algorithms and hardware for quantum computers, medical Imaging and semiconductor industry. By mastering these concepts, students can contribute to advancements in technology and research, making them valuable in both academic and professional settings.

Course Outcomes

Upon completion of the course the learner will be:

CO 1: Understanding the principles of time-independent, time-dependent perturbation theory, scattering phenomenon and principles of relativistic quantum mechanics and apply them to analyze phenomena such as the Zeeman effect, isotopic shifts, and the Stark effect.

CO 2: Applying the variational method to solve quantum mechanical problems involving approximation methods for stationary systems and relativistic quantum mechanics solve the Klein-Gordon and Dirac equations and applying them on physical systems.

CO 3: Analyzing time-dependent perturbation theory to predict system behaviour under various types of perturbations, including constant and harmonic, using Fermi's golden rule.

CO 4: Evaluating scattering processes using wave packet descriptions, born approximation, and partial wave analysis to derive results from scattering experiments.

Course Content

UNIT-I

14 Lectures

14

Approximation methods for stationary systems: Time-independent perturbation theory: (a) non-degenerate and(b) degenerate. Applications to Zeeman effect, isotopic shift and stark effect. Variational methodand its applications.

UNIT-II

Lectures

Approximation methods for time-dependent problems: Interaction picture. Time-dependent perturbation theory. Transition to a continuum of final states – fermi's golden rule. Application to constant and harmonic perturbations. Adiabatic and sudden approximations.

UNIT-III

12 Lectures

Scattering: wave packet description of scattering. Formal treatment of scattering by green function method. Born approximation and applications. Partial wave analysis. Optical theorem.

UNIT-IV

15 Lectures

Relativistic quantum mechanics: Klein-Gorden and Dirac equations. Properties of Dirac matrices. Plane wave solutions of Dirac equation. Spin and magnetic moment of electron. Nonrelativistic reduction of the Dirac equation. Spin-orbit coupling. Energy levels in a coulombs field.

Learning Experience

This course will utilize experiential and participatory learning methods to engage students in advanced quantum mechanics. Traditional lectures will introduce key concepts like perturbation theories and relativistic quantum mechanics, while case studies will link theory to practical applications, such as the Zeeman effect and Fermi's golden rule. Group activities and regular assignments will reinforce learning and foster collaboration. In-class demonstrations will be enhanced with visual aids and complemented by online discussions to promote continuous engagement. The instructor will offer regular office hours for support, and students will be encouraged to collaborate and provide peer feedback to enhance their learning experience.).

Textbooks

1. Advanced Quantum Mechanics, Satya Prakesh, Kedarnaath Ramnaath (2016).

2. Introduction to Quantum Mechanics, D.J. Griffith, Prentice Hall (1994).

Suggested Readings

1. Modern Quantum Mechanics, J.J Sakurai, Revised Edition, 1994, Addision-Wesley.

2. Advanced Quantum Mechanics, B. S. Rajput, Pragati Prakashan (2004).

3. Quantum Mechanics: Theory and Applications, (2019), (Extensively revised 6th Edition), Ajoy Ghatak and S. Lokanathan, Laxmi Publications, New Delhi.

Open Educational Resources (OER)

1. https://archive.nptel.ac.in/courses/115/103/115103104/

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VIII	III (FOR B.Sc. (hono		urs))		
SCPH406	ADVANCED SOLID-STATE PHYSICS-II	L	Τ	Р	С
Version 1.0		4	0	0	4
Category of Course	Major				
Total Contact Hours	54				
Pre-Requisites/ Co- Requisites	Solid State Physics				

Course Perspective

This course is essential for physics students as it provides a comprehensive understanding of solid-state physics, helping them build a strong foundation for academic and professional development in high-tech and research fields. By bridging theoretical physics with practical applications, it prepares students for advanced study in condensed matter physics and materials science, crucial for developing new materials, electronic devices, and energy technologies. Learning this course allows students to understand material behaviour at the atomic and

molecular levels, which is key for designing and enhancing electronic devices, sensors, and energy storage systems. The course fosters analytical skills, technical knowledge, and problemsolving abilities, making it vital for careers in semiconductor technology, electronics, telecommunications, nanotechnology, and renewable energy.

Course Outcomes

Upon completion of the course, the learner will be:

CO 1: Understanding the principles of band theory, including the Bloch theorem and Kronig-Penney model, to explain the behaviour of electrons in conductors, semiconductors, and insulators.

CO 2: Applying the concepts of Fermi levels and carrier concentrations to analyze the electrical properties of intrinsic and extrinsic semiconductors.

CO 3: Analyzing the dielectric properties of materials by exploring the microscopic concepts of polarization and the Clausius-Mosotti relation.

CO 4: Evaluating the effects of magnetic fields and thermal properties on superconductors, distinguishing between Type-I and Type-II superconductors.

Course Content

UNIT-I: Band theory

Bloch theorem, the Kronig-Penny Model, constriction of Brillouin zones, symmetry properties of energy function, extended, reduced and periodic zones schemes, effective mass of an electron, the nearly free electron model, conductors, semiconductors and insulators.

UNIT-II: Semiconducting properties of Materials

Semiconductors, carrier concentrations in intrinsic semiconductors, Fermi level; Statics of Extrinsic semiconductors: Fermi level, general equation for impurity semiconductor, Fermi level in an n-type semiconductor at very low temperature, Theory of n-type semiconductor, Theory of p-type semiconductor, mobility of charge carriers, effect of temperature on mobility, electrical conductivity of semiconductors, Hall Effect, junction properties.

UNIT-III: Dielectrics

Lectures

13 Lectures

14 Lectures

14

Dielectric constant, permittivity, microscopic concept of polarization, Langevin's theory of polarization in polar dielectrics, Local fields in solids and liquids, Classius-Mosotti relation, Static Dielectric constant in solids and liquids, Ferroelectricity, Piezoelectricity, dielectrics in AC fields, Ionic polarizability as a function of frequency, Complex dielectric constant of non-polar solids, Dipolar relaxation, dielectric losses,

UNIT-IV: Superconductivity

13 Lectures

Mechanism of superconductors, Effect of magnetic field, AC resistivity, Critical currents, Meissner Effect, Thermal properties, energy gap, Isotope effect, Mechanical effects, penetration depth, Type-I and Type-II superconductors, London equations, Superconductors in AC fields, Thermodynamics of superconductors, BCS theory, Quantum tunnelling, Josephson's tunnelling, DC Josephson's effect, Applications of superconductors

Learning Experience

This course in solid-state physics will employ a mix of experiential and participatory learning methods to engage students and enhance their understanding of core concepts. Traditional lectures will introduce foundational topics, while interactive sessions will encourage active participation and discussion. Experiential learning will be a key focus, with case studies linking theory to practical applications, hands-on laboratory sessions allowing for experimentation with materials, and collaborative group projects fostering peer learning. Assignments and problem sets will reinforce lecture material and encourage deeper exploration of topics. Assessments will include frequent quizzes, a midterm exam, and a final project, each designed to test and apply students' understanding in real-world contexts. Support will be provided through regular office hours for personalized guidance, peer review sessions to enhance critical thinking, and group activities that promote a collaborative.

Textbooks

1. Solid State Physics, S.O.Pillai, New Age Publication

Suggested Readings

- 1. Introduction to Solid State Physics, Charles Kittel, 8th Ed., 2004, Wiley India Pvt. Ltd.
- 2. Elements of Solid-State Physics, J.P. Srivastava, 2nd Ed., 2006, Prentice-Hall of India.
- 3. Introduction to Solids, Leonid V. Azaroff, 2004, Tata Mc-Graw Hill.
- 4. Solid State Physics, N.W. Ashcroft and N.D. Mermin, 1976, Cengage Learning.

Open Educational Resources (OER)

- 1. <u>https://archive.nptel.ac.in/courses/115/104/115104109/</u>
- 2. https://archive.nptel.ac.in/courses/115/105/115105099/

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VIII	(FOR]	B.Sc.	(hono	urs))
SCPH452	ADVANCED SOLID-STATE PHYSICS-II LAB	L	T	P	C
Version1.0		0	0	4	2
Category of Course	Major Lab				
Total Contact Hours	30				
Pre-Requisites/ Co- Requisites	Solid State Physics				

Course Perspective

This course offers a vital exploration of solid-state physics through a hands-on experimental approach, essential for physics students aiming to excel in high-tech and research-oriented careers. Emphasizing the practical application of theoretical concepts, the course allows

students to engage deeply with the material properties and behaviors at atomic and molecular levels. Through a series of carefully designed experiments, such as determining the band gap of semiconductors or measuring dielectric constants, students gain firsthand experience in essential techniques and methodologies. The experimental focus bridges the gap between theory and real-world applications, equipping students with the skills to design and improve electronic devices, sensors, and energy storage systems. By fostering analytical skills, technical knowledge, and problem-solving abilities, this course prepares students for advanced study and professional opportunities in semiconductor technology, electronics, nanotechnology, and renewable energy.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Observing and describing experimental phenomena in solid state physics, demonstrating an ability to accurately record and interpret data.

CO2: Imitating standard experimental procedures and techniques, replicating established methods with attention to accuracy and detail (Applying).

CO3: Practising using laboratory instruments and techniques, refining their skills through repeated use and procedural application.

Course Content

- 1. To find the band gap of intrinsic semi-conductor by using four probe method.
- 2. To determine the hysteresis loss of ferromagnetic material by using CRO.
- 3. To measure the dielectric constant of the material.
- 4. Determine the value of Hall coefficient for a given sample.
- 5. To study the temperature dependence of Hall coefficient of a given semiconductor.
- 6. Determine the Curie Temperature of a given ferroelectric material
- 7. Determine the specific heat of a given sample at room temperature.
- 8. Preparation of single crystals of copper sulfate using slow evaporation technique and

identification of the c-axis of the newly grown triclinic crystal.

9. Determination of magnetoresistance of a given semiconductor for different magnetic

field.

Learning Experience

The practical component of this solid-state physics course emphasizes hands-on experimentation and active engagement with material properties to enhance learning. Students will participate in laboratory sessions to apply theoretical knowledge through experiments, such as measuring semiconductor band gaps and exploring the Hall effect. These activities build practical skills and understanding of experimental techniques. Case studies and collaborative projects will connect theory to real-world applications and promote peer learning. Assignments and problem sets will reinforce concepts and develop analytical skills. Regular assessments, including quizzes and a final project, will evaluate students' ability to integrate theoretical and practical knowledge. Continuous support from instructors and peer reviews will foster a collaborative and supportive learning environment.

Textbooks

1. Solid State Physics, S.O.Pillai, New Age Publication

Suggested Readings

- 1. Introduction to Solid State Physics, Charles Kittel, 8th Ed., 2004, Wiley India Pvt. Ltd.
- 2. Elements of Solid-State Physics, J.P. Srivastava, 2nd Ed., 2006, Prentice-Hall of India.
- 3. Introduction to Solids, Leonid V. Azaroff, 2004, Tata Mc-Graw Hill.
- 4. Solid State Physics, N.W. Ashcroft and N.D. Mermin, 1976, Cengage Learning.

Open Educational Resources (OER)

- 1. https://archive.nptel.ac.in/courses/115/104/115104109/
- 2. https://archive.nptel.ac.in/courses/115/105/115105099/

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VIII	(FOR B.Sc. (B.Sc. (honours))		
SCPH454	Latex Lab	L	Т	Р	С
Version1.0		0	0	4	2
Category of Course	Skill Enhancement Course				
Total Contact Hours	30				
Pre-Requisites/ Co- Requisites					

Course Perspective: This course imparts the basic concepts of Latex Lab. It enables them to type mathematical documents and research paper. This course helps students in variety of ways to upload their own figures and tables in document. The course introduces the basic concepts about packages, commands and formatting of documents.

Course Outcomes:

Upon completion of the course, the learner will be able:

CO1: Understanding the basic commands and environments in TeX and LaTeX, including how to structure a simple document and add basic information like footnotes and sections.

CO 2: Explaining how different environments, accents, symbols, and mathematical expressions are used in LaTeX, demonstrating a clear understanding of typesetting rules and conventions.

CO 3: Utilizing LaTeX to create and typeset documents that include complex mathematical expressions, such as fractions, arrays, and multiline formulas, as well as incorporate graphics and plots.

CO 4: Breaking down and comparing the use of different LaTeX packages and commands for mathematical typesetting, graphics creation, and Beamer presentations, determining the most efficient approaches for specific tasks.

CO 5: Critically assessing the formatting and presentation quality of documents and Beamer slides created using LaTeX, identifying areas for improvement in terms of readability and visual appeal.

CO6: Designing and producing professional-quality documents and presentations using advanced LaTeX features, including graphics, plots, and Beamer slides, showcasing creativity and technical proficiency.

Course Content:

Unit 1:

No. of Hours: 10

No. of Hours: 10

Introduction to TeX and LaTeX, Type-setting a simple document, Adding basic information to a document, Environments, Footnotes, Sectioning and displayed material

Unit 2:

Accents and symbols, Mathematical typesetting (elementary and advanced): Subscript/ Superscript, Fractions, Roots, Ellipsis, Mathematical Symbols, Arrays, Delimiters, Multiline formulas, Spacing and changing style in math mode.

Unit 3:

No. of Hours: 10

Graphics in LaTeX, Simple pictures using PSTricks, Plotting of functions, Beamer presentation.

Learning Experience:

This course will be conducted through hands-on sessions where students will actively engage in typesetting documents using TeX and LaTeX. Each unit will involve real-time practice, where participants create and modify documents, mathematical expressions, and graphics. Group activities and peer reviews will make the learning process experiential and participatory, encouraging collaboration and continuous feedback.

Textbooks:

 Bindner, Donald & Erickson, Martin. (2011). A Student's Guide to the Study, Practice, and Tools of Modern Mathematics. CRC Press, Taylor & Francis Group, LLC.

Suggested Readings:

 Lamport, Leslie (1994). LaTeX: A Document Preparation System, User's Guide and Reference Manual (2nd ed.). Pearson Education. Indian Reprint.

Open Educational Resources (OER)

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VIII	(FOR B.Sc. (ho	nours	with	Resea	rch))
SCPH404	Research project	L	Τ	Р	C
Version1.0					12
Category of Course	Major	1	1	1	1
Total Contact Hours					
Pre-Requisites/ Co-	Practical exposure				
Requisites					

Course Perspective: Students will be divided among faculty members of the Department for the supervision of the research work. In the first week of Semester VII, each faculty member will assign a suitable research topic to the students from the selected topics in the areas of chemical sciences. The student will work on the assigned research topic during semesters VII and VIII in regular consultation with his/her assigned teacher. The student will write a dissertation based on the research work carried out during Semesters VII and VIII and prepare two copies to be submitted to the office of the Head of the Department duly signed by the student and the supervisor in the sixth week of VIII semester or a date decided by the HOD of the department. Before preparing power point presentation and submission of dissertation, each student has to deliver a seminar talk on his/ her research project work on a date fixed by HOD, necessary suggestions have to be incorporated in the final draft of dissertation. The student will make a power point presentation based on the work carried out and mentioned in the dissertation to the board of examiners appointed by the University.

Course Outcomes

On completion of this course, the students will be:

- CO1. Carrying out the extensive literature survey.
- C02. Applying the learning techniques to write and present technical reports/articles.
- **CO3.** Analyzing various methods and techniques applicable to the topic to study and contribute to domain knowledge.
- **CO4.** Evaluating the result of the experiment carried out and presenting the results using data visualization methods.

Learning Experience:

The course will be conducted through hands-on research activities, where students choose a topic, design experiments, and collect data under faculty mentorship. Regular group discussions, peer reviews, and presentations will foster a participatory learning environment. The course will emphasize experiential learning through real-world problem-solving, encouraging students to apply theoretical knowledge in practical research settings.

Evaluation Scheme

Particular	Weightage
Internal Marks: -	
(Punctuality, Performance, Work	50 Marks
Ethics, Efforts and Research Output)	
External Marks (Practical): -	50 Marks
Presentation	20
Report Writing	10
Viva Voce	20

Minor-Nano Science

SEMESTER I					
UNS101	Study of Materials	L	Т	Р	С
Version1.0		4	0	0	4
Category of Course	Minor				
Total Contact Hours	60 hours				
Pre-Requisites/ Co-	Basic concepts of Physics, Chemistry				
Requisites					

Course Perspective: All the modern materials show some unique properties which either are by the virtue of material or may be tailored. Metallurgists and Materials scientists are responsible for designing and producing new materials. The desired properties may be introduced in the materials by altering their microstructures. This course will help students understand the properties of different types of materials and their applications. The course will also be helpful to develop new kind of materials for engineering applications.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Understanding the basic concepts of stress, strain, and deformation, and explain how materials respond to mechanical forces.

CO 2: Applying the principles of dislocation and strengthening mechanisms to improve the mechanical properties of metals.

CO 3: Analyzing solid solutions and phase diagrams to determine the impact of cooling and structural changes on material properties.

CO 4: Evaluating different types of failure such as fracture, fatigue, creep, and corrosion, and recommend strategies to prevent material degradation.

Course Content

UNIT I Mechanical Properties of Metals

No. of Hours: 15

Concepts of Stress and Strain, Elastic Deformation: Stress-Strain Behavior, Anelasticity, Elastic Properties of Materials; Plastic Deformation: Tensile Properties, True Stress and Strain, Elastic Recovery after Plastic Deformation, Compressive, Shear, and Torsional Deformation, Hardness; Property Variability and Design/Safety Factors: Variability of Material Properties, Design/Safety Factors.

UNIT II Dislocations and Strengthening Mechanisms No. of Hours: 15

Characteristics of Dislocations, Slip Systems, Slip in Single Crystals, Plastic Deformation of Polycrystalline Materials, mechanism of plastic deformation, deformation by twinning, Mechanisms of Strengthening In Metals: Strengthening by Grain Size Reduction, 7.9 Solid-Solution Strengthening, Strain Hardening; Recovery, Recrystallization and Grain Growth: Recovery, Recrystallization, Grain Growth.

UNIT III Solid solutions and phase diagram No. of Hours: 15

Introduction to single and multiphase solid solutions and types of solid solutions, importance and objectives of phase diagram, systems, phase and structural constituents, cooling curves, unary & binary phase diagrams, Gibbs's phase rule, Lever rule, eutectic and eutectoid systems, peritectic and peritectoid systems, iron carbon equilibrium diagram and TTT diagram.

UNIT IV Failures of metals

No. of Hours: 15

Failure analysis, fracture, process of fracture, types of fracture, fatigue, characteristics of fatigue, fatigue limit, mechanism of fatigue, factors affecting fatigue. Definition and concept of Creep, creep curve, mechanism of creep, impact of time and temperature on creep, creep fracture, creep testing and prevention against creep. Corrosion: Mechanism and effect of corrosion, prevention of corrosion

Learning Experience:

- 1. Classes will incorporate interactive lectures supported by multimedia presentations, simulations, and virtual labs.
- 2. Group activities such as problem-solving sessions, discussions, and peer reviews will be encouraged.
- 3. Regular assignments will be designed to challenge students to apply concepts learned in class. Quizzes, mid-term exams, and final assessments will focus on evaluating students' understanding, analytical skills, and problem-solving abilities. Students will receive timely feedback on their progress.
- 4. The course instructor will be available for additional support through office hours and one-on-one meetings.

Textbooks"

 Materials Science and Engineering: An Introduction (7th Ed.), William D. Callister, Jr., John Wiley & Sons, Inc.

Suggested Readings"

- 1. Material Science Narula, Narula and Gupta. New Age Publishers
- 2. Material Science & Engineering –V. Raghvan, Prentice Hall of India Pvt. Ltd, New Delhi.
- 3. A Textbook of Material Science & Metallurgy O.P. Khanna, Dhanpat Rai & Sons

Open Educational Resources (OER):

- https://www.govinfo.gov/content/pkg/GOVPUB-C13
 e18ffcc1681da9e902df23acaeb5cc6c/pdf/GOVPUB-C13 e18ffcc1681da9e902df23acaeb5cc6c.pdf
- 2. https://uomustansiriyah.edu.iq/media/lectures/6/6_2018_05_19!12_50_38_AM.pdf

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER II					
UNS102	Elements of Nano sciences and nanomaterial	L	T	Р	C
Version 1.0		4	0	0	4
Category of Course	Minor	1		1	
Total Contact Hours	60 hours				
Pre-Requisites/ Co- Requisites	Basic concepts of Physics, Chemistry				

Course Perspective: The aim of this course is to introduce an emerging class of materials called nanomaterials that consists of a broad spectrum of examples with at least one dimension in the range of 1 to 100 nm. Exceptionally high surface areas can be achieved through the rational design of nanomaterials. It will also explain how nanomaterials can be produced with outstanding magnetic, electrical, optical, mechanical, and catalytic properties that are substantially different from their bulk counterparts. The course will conclude with various types of characterization techniques which can be used for analysing these nanomaterials.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Understanding the basic concepts of nanoscience, including the size effects, crystal structures, and the influence of nanostructuring on material properties.

CO 2: Applying quantum mechanics principles to explain the behavior of particles at the nanoscale, including solutions to the Schrödinger equation for different scenarios.

CO 3: Analyzing different types of nanostructured materials and evaluate how their dimensional properties affect their mechanical, optical, electronic, and chemical characteristics.

CO 4: Evaluating various chemical and biomimetic synthesis techniques for creating nanomaterials and assess their effectiveness in different applications.

Course Content:

Unit I Background to Nano science No. of Hours: 15

Definition of Nano, Scientific revolution-atomic Structure and atomic size, emergence and challenges of nano science and nanotechnology, carbon age-new form of carbon (CNT to Graphene), influence of Nano over micro/macro, size effects and crystals, large surface to volume ratio, surface effects on the properties. Influence of Nano structuring on Mechanical, optical, electronic, magnetic and chemical properties.

Unit-II Introduction to Quantum Mechanics No. of Hours: 15

Schrodinger equation and expectation values, Solutions of the Schrodinger equation for free particle, particle in a box, particle in a finite well, Reflection and transmission by a potential step and by a rectangular barrier. Angular momentum and its operators, Eigen values and Eigen functions of the angular momentum operators, spin, Pauli spin operators and their properties, hydrogen atom, density of states, free electron theory of metals.

Unit III Types of nanostructure and properties of nanomaterial No. of Hours: 15

One dimensional, two dimensional and three-dimensional nanostructured materials, Quantum Dots shell structures, metal oxides, semiconductors, composites, mechanical-physical-chemical properties.

Chemical synthesis of nano material: Self-assembly, self-assembled monolayers (SAMs). Langmuir-Blodgett (LB) films, colloids, zeolites, organic block copolymers, emulsion polymerization, template synthesis, and confined nucleation and/or growth. Biomimetic Approaches: polymer matrix isolation, and surface-template nucleation and/or crystallization. Vapour (or solution) – liquid – solid (VLS or SLS) growth -Electrochemical Approaches: anodic oxidation of alumina films, porous silicon, and pulsed electrochemical deposition

Unit IV Characterization of nanomaterial

No. of Hours: 15

X-ray Diffraction - Thermal Analysis Methods, Differential Thermal Analysis and Differential scanning calorimetry - Spectroscopic techniques, UV-Visible Spectroscopy – IR Spectroscopy – Microwave Spectroscopy - Raman Spectroscopy: Electron Spin Resonance Spectroscopy, NMR Spectroscopy, Particle size characterization: Zeta Potential Measurement, Particle size Analysis: X-ray Photoelectron spectroscopy. Imaging techniques for nanotechnology: Scanning Electron Microscopy, Transmission Electron Microscopy, and Atomic Force Microscopy.

Learning Experience:

- 1. The course will combine traditional lectures with interactive digital tools like virtual labs, 3D simulations, and video tutorials.
- 2. Students will participate in hands-on experiments, focusing on nanomaterial synthesis, characterization, and quantum mechanics applications.
- 3. Group projects will encourage students to work together on designing experiments, analyzing nanomaterials, and solving problems related to nanoscale phenomena.
- 4. Students will complete regular assignments that challenge them to apply course concepts, such as solving quantum mechanics problems or designing synthesis techniques for nanomaterials.
- 5. The course instructor will be available during office hours for additional support and guidance.

Textbooks:

- 1. Nanomaterials Chemistry by Rao C. N., A. Muller, A. K. Cheetham, WileyVCH, 2007.
- Nanomaterials and Nanochemistry by Brechignac C., P. Houdy, M. Lahmani, Springer publication, 2007.
- 3. Quantum Physics A. Ghatak

Suggested Readings:

- 1. Chemistry of nanomaterials: Synthesis, properties and applications by CNR Rao et.al.
- 2. Nanoparticles: From theory to applications G. Schmidt, Wiley Weinheim 2004.
- **3.** Instrument E L Principe, P Gnauck and P Hoffrogge, Microscopy and Microanalysis (2005), 11: 830- 831, Cambridge University Press.
- Processing & properties of structural naonmaterials Leon L. Shaw, Nanochemistry: A Chemical Approach to Nanomaterials, Royal Society of Chemistry, Cambridge UK 2005.

Open Educational Resources (OER):

- 1. <u>https://www.youtube.com/watch?v=0EWCqCIsFOA</u>
- 2. <u>https://www.youtube.com/watch?v=-K7Gs0Nj-</u> <u>50&list=PLQzUXa8lZVq_y0i5dOjW6oEr6h43bJCV</u>
- 3. https://nptel.ac.in/courses/118104008
- 4. https://nptel.ac.in/courses/115101007

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER III					
UNS103	Nanostructured materials	L	Т	Р	C
Version 1.0		4	0	0	4

Minor
60 Hours
Basics of nanomaterial

Course Perspective:

The aim of this course is making students understand the importance of nanostructured materials. Nanostructured materials have gained prominence in technological advancements due to their tunable physicochemical characteristics such as melting point, wettability, electrical and thermal conductivity, catalytic activity, light absorption and scattering resulting in enhanced performance over their bulk counterparts. Knowledge about these emerging materials will further help the students to explore these materials for advanced real-life applications.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Understanding the principles of nanocomposites, their classification, and their applications in fields like nuclear energy, Spintronics, and high-temperature environments.

CO 2: Applying the concepts of quantum confinement and size effects to explain the properties of nanostructures and their influence on material behavior and applying knowledge to optimize film properties for specific uses.

CO 3: Analyzing the synthesis methods and properties of nano ceramics, nano polymers, and conducting polymers, and evaluate their potential in various industrial applications.

CO 4: Evaluating the applications of nanotechnology in fields such as healthcare, consumer products, and energy devices, and assess the impact of these advancements.

Course Content

Unit I: Nano Composites

No. of Hours: 15

Nano Composites and their Applications, Metal-Metal Nano composites for nuclear energy applications, Magnetic Nano composites for Spintronics application, Ceramic Nano composites for high temperature applications. Length, energy, and time scales - Quantum confinement of electrons in semiconductor nanostructures: Quantum confinement in 3D, 2D, 1D and zero dimensional structures -Size effect and properties of nanostructures, Top down and Bottom-up approach.

Unit II: Nano Ceramics

Nano ceramics: Dielectrics, ferroelectrics and magneto ceramics, Nano polymers: Preparation and characterization of d-block Copolymer based Nano composites, Nanoparticles polymer ensembles; Applications of Nano polymers in Catalysis.

Unit III: Polymers

Classification of conducting polymers: Intrinsic and extrinsic conducting polymers - Chemical and electrochemical methods of the synthesis of conducting polymers – Applications of conducting polymers in corrosion protection, sensors, electronic and electrochemical energy devices.

Unit IV: Thin Films

Miscellaneous applications of nanotechnology: dental implants, consumer products, biomimetic nanomaterial for tissue engineering, biopolymer tagging, semiconductor quantum dots.

Thin Film Formation Methods- Physical methods: thermal evaporation - vapour sources -Wire, crucible and electron beam gun - sputtering mechanism and methods - epitaxy - MBE. Chemical methods: chemical vapour deposition and chemical solution deposition techniques – spray pyrolysis - laser ablation

Learning Experience:

- 1. The course will use multimedia tools, including animations and simulations, to explain complex concepts such as quantum confinement, thin film deposition, and nanocomposites.
- 2. Recorded lectures and online resources will also be available for self-paced learning.
- 3. Students will work in groups to analyze specific nanotechnology applications, conduct experiments, and develop innovative solutions.

No. of Hours: 15

No. of Hours: 15

No. of Hours: 15

- 4. Regular assignments will be given to reinforce key concepts, such as the classification of conducting polymers and thin film deposition methods.
- **5.** The course instructor will be available for additional support during office hours and will provide timely feedback on student progress.

Textbooks:

- 1. Materials Science and Engineering An Introduction, William D Callister, 12th Edition, John Wiley
- Nanomaterials An introduction to synthesis, properties and applications, D. Vollath, Wiley-VCH, Second Edition 2013.

Suggested Readings:

- 1. Novel Nanocrystalline Alloys and Magnetic Nanomaterials- Brian Cantor
- 2. Nanoscale materials -Liz Marzan and Kamat.

Open Educational Resources (OER):

- 1. https://youtu.be/6TprsnrvKIk
- 2. https://youtu.be/j wQgy97Pi4
- 3. <u>https://youtu.be/CJn2gXp3pyo</u>
- 4. <u>https://youtu.be/TgwpVGWL6dQ</u>
- 5. <u>https://youtu.be/nSAvyQajVzE</u>
- 6. https://youtu.be/mbOQYIBp0VQ
- 7. https://youtu.be/ev1EiLWgDIs

Evaluation Scheme:

Evaluation components			Weightage
Internal marks (Theory)			30 Marks
I. continuous assessment (30	marks)		
All the components to be evenl	y spaced		
Project/quizzes/assignment	and	essays/presentation/	
II. internal marks (Theory):			20 Marks
Mid Term Examination			

III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER IV					
UNS104	Crystallography	L	Т	Р	C
Version 1.0		4	0	0	4
Category of Course	Minor				
Total Contact Hours	60				
Pre-Requisites/ Co Requisites	Basic knowledge in Materials	Science			

Course Perspective

This course is an introduction to the principles of structure of materials, and theory and applications of diffraction and imaging techniques for materials characterization using X-ray diffraction and transmission electron microscopy (TEM). This course enhances your understanding of material properties and is crucial for careers in materials science, solid-state physics. You'll gain skills in analyzing and interpreting crystal structures, which are essential for research and development in various scientific and industrial fields. For example, knowledge of crystallography helps in the design of new materials for electronics or pharmaceuticals, directly impacting technology and industry.

Course Outcomes

Upon completion of the solid-state physics course, the learner will be:

CO1: Identifying and describing fundamental concepts and principles related to crystallography.

CO2: Analyzing and interpreting data using standard crystallographic techniques and tools.

CO3: Applying theoretical knowledge to solve practical problems and conducting experiments in crystallography.

CO4: Evaluating and synthesizing information from various sources to make informed decisions and recommendations related to crystallographic research.

Course Content

Unit No I: Geometric crystallography

Amorphous and crystalline materials, lattice translation vectors, lattice with a basis –unit cell, types of lattices symmetry elements, inter planer spacing, packing fraction, Miller Indices, Bonding in solids- ionic bond. covalent bond, metallic bonds, hydrogen bonding, van Der Waals bond, crystal defects, point defects, line defects, Burger's vector, surface imperfections.

Unit No II: Structural crystallography and crystal chemistry No. of Hours 15

The symmetry of the unit cell., Space groups, atomic positions and structural positions, Crystal structures, Principles that govern the formation of crystalline structures, Variations in the chemical composition of the crystals. Isomorphism, solid solutions and stoichiometry, X-ray diffraction by crystals, Diffraction methods: fundamentals and information they provide.

Unit No III: Physical properties of crystals

Introduction to the physical properties of crystals and their relation to crystalline symmetry. Optical properties, Nature of light, and other basic concepts, Optical properties, Isotropy and optical anisotropy. The optical surfaces, Optical properties, transmitted light polarization microscope, Optical properties, Optical observations with parallel light and without analyzer. Optical determinations with parallel light and analyser, Optical determinations with convergent light.

Unit No IV: Crystal Dynamics

The real crystal, Crystal defects and crystalline dynamics, Influence of defects on the physical properties of crystals, Crystal defects: punctual, linear, two-dimensional and three-dimensional. Crystal formation and growth, Morphology of the real crystal, Add and twins, Polymorphism.

Learning Experience

No. of Hours 15

No. of Hours 15

No. of Hours 15

The Crystallography course will be delivered through interactive lectures, practical lab work, and collaborative projects. Students will analyze crystal structures using X-ray diffraction, engage in group case studies, and complete individual assignments. Technology will be used for data analysis, and the course will include opportunities for presentations and peer feedback. Support and feedback will be available from the instructor, and students are encouraged to collaborate and seek help as needed.

Textbooks

1. David B. Williams and C. Barry Carter, Transmission Electron Microscopy: A Textbook for Materials Science, Plenum Press, NY (2007).

Suggested Readings

- 1. Introduction to Solid State Physics C. Kittel.
- 2. Principles of Solid-State Physics R. A. Levy Solid State Physics- S.O. Pillai.
- 3. Elements of X-Ray diffraction B.D. Cullity.
- 4. Elementary Solid-State Physics Ali Omar.
- 5. Elements of Solid-State Physics J.P. Srivastava.
- 6. Nano: The Essentials by T. Pradeep (Tata McGraw Hill Publ).

Open Educational Resources (OER)

- 1. <u>Introduction to Crystallography and Mineral Crystal Systems</u> A comprehensive overview of geometric crystallography.
- 2. <u>Crystallography Open Database</u> A database of crystal structures.
- 3. <u>MIT Open Courseware Crystal Structure Reading Collection</u> Reading materials on crystal structures.
- 4. <u>Fundamentals of Crystallography</u> An article on the principles that govern the formation of crystalline structures.
- 5. <u>Introduction to Crystal Physics</u> A detailed course on the physical properties of crystals.
- 6. <u>Crystalline Materials</u> Explains the optical properties of crystals.
- 7. <u>Solid State Physics</u> A chapter on crystal defects and dynamics from a course on solid state physics.

8. <u>Crystal Growth & Design</u> - A journal with open access articles on crystal formation and growth.

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER IV					
UNS105	Crystallography Lab	L	Т	Р	C
Version 1.0		0	0	4	2
Category of Course	Minor Lab				
Total Contact Hours	30				
Pre-Requisites/ Co	Basic knowledge in Materials Science	è			
Requisites					

Course Perspective

This course is an introduction to study of structure of materials, coordination number, and bond lengths etc. using Diamond software. The Crystallography Lab is crucial for students as it provides hands-on experience with X-ray diffraction and other techniques to analyze crystal structures. This lab enhances academic understanding of material properties, supports career development in materials science and research, and builds practical skills in data analysis and problem-solving. Learning these techniques is vital for real-world applications like designing new materials or studying pharmaceuticals, where precise crystal structure knowledge is essential.

Course Outcomes

Upon completion of the solid-state physics course, the learner will be:

CO1: Observing and recording experimental phenomena accurately, demonstrating an understanding of lab processes.

CO2: Imitating established experimental techniques, applying standard procedures with precision.

CO3: Practicing using lab equipment and methods, enhancing technical skills through handson experience.

Course Content

List of experiments

- 1. To Study the structure of Simple cubic crystal system.
- 2. To Study the structure of Body centred cubic crystal system.
- 3. To Study the structure of Face centred cubic crystal system.
- 4. To Study the structure of tetragonal crystal system.
- 5. To Study the structure of Orthorhombic crystal system.
- 6. To Study the structure of Rhombohedral crystal system.
- 7. To Study the structure of hexagonal crystal system.
- 8. To Study the structure of monoclinic Crystal system.
- 9. To Study the structure of Triclinic crystal system.
- 10. To Study the structure of Perovskites.

Learning Experience

The Crystallography Lab course combines interactive lectures with hands-on experiments to provide practical experience in crystallographic techniques. Students will work in groups on lab projects, utilize software for data analysis, and engage in case studies to understand real-world applications. Assessments include practical exams, detailed lab reports, and

presentations. The instructor will offer regular feedback and be available for additional support, while peer collaboration and additional resources will further enhance learning.

Textbooks

1. David B. Williams and C. Barry Carter, Transmission Electron Microscopy: A Textbook for Materials Science, Plenum Press, NY (2007).

Suggested Readings

- 1. Introduction to Solid State Physics C. Kittel.
- 2. Principles of Solid-State Physics R. A. Levy Solid State Physics- S.O. Pillai.
- 3. Elements of X-Ray diffraction B.D. Cullity.
- 4. Elementary Solid-State Physics Ali Omar.
- 5. Elements of Solid-State Physics J.P. Srivastava.
- 6. Nano: The Essentials by T. Pradeep (Tata McGraw Hill Publ).

Open Educational Resources (OER)

- 1. <u>https://youtu.be/HCWwRh5CXYU</u>
- 2. https://youtu.be/_9RnbGqtkd4
- 3. https://youtu.be/GSPVC34ijIA
- 4. <u>https://youtu.be/JS9ysbgr0BE</u>
- 5. <u>https://youtu.be/07iZ7-IEyYE</u>

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Practical's)	
I. Conduct of experiment	10 Marks
II . Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical's):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER V						
UNS106	SYNTHESIS O	F	L	Т	Р	С
	NANOMATERIALS-I					
Version 1.0			4	0	0	4
Category of Course	Minor					
Total Contact Hours	60 hours					
Pre-Requisites/ Co-	Basics of Nanomaterials					
Requisites						

Course Perspective:

This course introduces the fundamentals of nanostructures, including their synthesis, characterization, and applications. Topics covered include thin films, carbon nanotubes, mesoporous structures, and quantum devices. A background in physics, chemistry, or materials science is recommended. Upon completion of this course, students will be able to: Understand the basic principles of nanostructure synthesis and characterization, apply these principles to the design and fabrication of nanostructures

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Describing the fundamental concepts and methods involved in nanotechnology, including top-down and bottom-up processes, the properties of atoms and solids, and the different classification of nanostructures.

CO2: Applying principles of nanotechnology to understand the synthesis and growth of nanoparticles through both homogeneous and heterogeneous nucleation.

CO3: Analyzing various fabrication methods and their impact on the properties of nanoparticles and nanostructures.

CO4: Evaluating the effectiveness of various methods for creating one-dimensional nanostructures, such as nanowires and nanorods.

Course Content:

UNIT-I Generic methodologies for nanotechnology: classification and fabrication No. of Hours: 15

Introduction and classification: definition nanotechnology, Classification of nanostructures, Nanoscale architecture, Summary of the electronic properties of atoms and solids: The isolated atom Bonding between atoms, Giant molecular solids, The free electron model and energy bands, Crystalline solids, Periodicity of crystal lattices, Electronic conduction; Effects of the nanometre length scale: Changes to the system total energy, Changes to the system structure, How nanoscale dimensions affect properties, Fabrication methods: Top-down processes, Bottom-up processes, Methods for templating the growth of nanomaterials, Ordering of nano systems, Preparation, safety and storage issues.

UNIT-II Physical Chemistry of Solid Surface

No. of Hours: 15

Introduction, Surface Energy, Chemical Potential as a Function of Surface Curvature, Electrostatic Stabilization: Surface charge density, Electric potential at the proximity of solid surface, Van der Waals attraction potential, Interactions between two particles: DLVO theory, Solvent and polymer, Interactions between polymer layers, Mixed steric and electric interactions.

UNIT-III Zero-Dimensional Nanostructures: Nanoparticles No. of Hours: 15

Introduction, Nanoparticles through Homogeneous Nucleation: Fundamentals of homogeneous nucleation, Subsequent growth of nuclei (Growth controlled by diffusion, Growth controlled by surface process), Synthesis of metallic nanoparticles); Epitaxial of reduction reagents, Influences by other factors, Influences of polymer stabilizer), Synthesis of semiconductor nanoparticles, Synthesis of oxide nanoparticles (Introduction to sol-gel processing, Forced hydrolysis, Controlled release of ions), Vapor phase reactions, Solid state phase segregation; Nanoparticles through Heterogeneous Nucleation: (Fundamentals of heterogeneous nucleation, Synthesis of nanoparticles); Kinetically Confined Synthesis of Nanoparticles: (Synthesis inside micelles or using microemulsions, Aerosol synthesis, Growth termination, Spray pyrolysis, Template-based synthesis); Epitaxial Core-Shell Nanoparticles.

UNIT IV One-Dimensional Nanostructures: Nanowires and Nanorods No. of Hours: 15

Introduction, Spontaneous Growth: Evaporation (dissolution)-condensation growth:(Fundamentals of evaporation (dissolution)-condensation growth, Evaporationcondensation growth, Dissolution-condensation growth); Vapor (or solution)-liquid-solid (VLS or SLS) growth: (Fundamental aspects of VLS and SLS growth, VLS growth of various nanowires, Control of the size of nanowires, Precursors and catalysts, SLS growth); Stressinduced recrystallization: Template-Based Synthesis: Electrochemical deposition, Electrophoretic deposition, Template filling (Colloidal dispersion filling, Melt and solution filling, Chemical vapor deposition, Deposition by centrifugation), Converting through chemical reactions; Electrospinning; Lithography.

Learning Experience:

- 1. The course will combine traditional lectures with interactive digital tools like virtual labs, 3D simulations, and video tutorials.
- 2. Students will participate in hands-on experiments, focusing on nanomaterial synthesis, characterization, and quantum mechanics applications.
- 3. Group projects will encourage students to work together on designing experiments, analyzing nanomaterials, and solving problems related to nanoscale phenomena.
- 4. Students will complete regular assignments that challenge them to apply course concepts, such as solving quantum mechanics problems or designing synthesis techniques for nanomaterials.
- 5. The course instructor will be available during office hours for additional support and guidance.

Textbooks:

- Introduction to Nanoscience and Nanotechnology" by Gabor L. Hornyak, H.F. Tibbals, Joydeep Dutta, John J. Moore
- 2. Nanoscale Science and Technology" by Robert Kelsall, Ian Hamley, Mark Geoghegan

Suggested Readings

- 1. Nanotechnology: An Introduction" by Jeremy Ramsden
- Nanoscale: Visualizing an Invisible World" by Kenneth S. Deffeyes, Stephen E. Deffeyes
- 3. Nanotechnology: Principles and Practices" by Sulabha K. Kulkarni

Open Educational Resources (OER):

- 1. <u>https://byjus.com/jee/surface-chemistry/</u>
- 2. <u>https://onlinecourses.nptel.ac.in/noc21_cy45/preview</u>
- 3. https://www.youtube.com/watch?v=O2So0xcdDiA
- 4. <u>https://www.nobelprize.org/prizes/chemistry/2007/ertl/lecture/</u>
- 5. <u>https://www.doubtnut.com/question-answer-chemistry/if-physical-adsorption-</u> <u>the-gas-molecules-are-held-on-solid-surface-by-46827508</u>

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER VI					
UNS107	Synthesis of Nanomaterials-II	L	T	P	C
Version 1.0		4	0	0	4
Category of Course	Minor	I			
Total Contact Hours	60 Hours				
Pre-Requisites/ Co- Requisites	Basics of Nanomaterial				

Course Perspective:

This course introduces the fundamentals of nanostructures, including their synthesis, characterization, and applications. Topics covered include thin films, carbon nanotubes, mesoporous structures, and quantum devices. A background in physics, chemistry, or materials science is recommended. Upon completion of this course, students will be able to: Understand the basic principles of nanostructure synthesis and characterization, apply these principles to the design and fabrication of nanostructures.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Understanding the fundamental principles of various film growth techniques and deposition methods, including PVD, CVD, ALD, and self-assembly, and explain their applications in nanotechnology.

CO2: Applying techniques to fabricate and characterize special nanomaterials, such as carbon nanotubes, metal-oxide structures, and nanocomposites, demonstrating their properties and potential uses.

CO3: Analyzing different nanostructure fabrication methods, including lithography, nanomanipulation, and soft lithography, and evaluate their effectiveness in producing nanostructures with desired features.

CO4: Evaluating the applications of nanomaterials in fields like molecular electronics, biological applications, and catalysis, assessing their impact and potential for innovation in various technologies.

Course Content

UNIT-I Two-Dimensional Nanostructures: Thin Films No. of Hours: 15

Introduction; Fundamentals of Film Growth; Vacuum Science; Physical Vapor Deposition (PVD): (Evaporation, Molecular beam epitaxy (MBE), Sputtering, Comparison of evaporation and sputtering); Chemical Vapor Deposition (CVD):(Typical chemical reactions, Reaction kinetics, Transport phenomena,5CVD methods, Diamond films by CVD).

Atomic Layer Deposition (ALD); Superlattices; Self-Assembly:(Monolayers of organosilicon or alkyl silane derivatives, Monolayers of alkanethiols and sulfides, Monolayers of carboxylic acids, amines alkyl silane derivatives and alcohols); Langmuir-Blodgett Films; Electrochemical Deposition; Sol-Gel Films

UNIT-II Special Nanomaterials

Introduction; Carbon Fullerenes and Nanotubes (Carbon fullerenes, Fullerene-derived crystals, Carbon nanotubes); Ordered mesoporous structures; Random mesoporous structures; Crystalline microporous materials: zeolites; Metal-oxide structures; Metal-polymer structures; Oxide-polymer structures; Organic-Inorganic Hybrids :(Class I hybrids, Class II hybrids); Intercalation Compounds; Nanocomposites and Nanograined Materials.

UNIT-III Nanostructures Fabricated by Physical Techniques No. of Hours: 15

Introduction; Lithography:(Photolithography, Phase-shifting photolithography, Electron beam lithography, X-ray lithography, Focused ion beam (FIB) lithography, Neutral atomic beam lithography); Nanomanipulation and Nanolithography : (Scanning tunnelling microscopy (STM), Atomic force microscopy (AFM) ,Near-field scanning optical microscopy (NSOM) ,Nanomanipulation, Nanolithography); Soft Lithography: (Microcontact printing, Moulding, Nanoimprint, Dip-pen nanolithography); Assembly of Nanoparticles and Nanowires: (Capillary forces, Dispersion interactions, Shear force assisted assembly, Electric-field assisted assembly, Covalently linked assembly, Gravitational field assisted assembly, Template-assisted assembly); Other Methods for Microfabrication

UNIT IV Applications of Nanomaterials

No. of Hours: 15

Introduction; Molecular Electronics and Nanoelectronics; Nanobots; Biological Applications of Nanoparticles; Catalysis by Gold Nanoparticles; Band Gap Engineered Quantum Devices: (Quantum well devices, Quantum dot devices); Nanomechanics; Carbon Nanotube Emitters; Photoelectrochemical Cells; Photonic Crystals and Plasmon Waveguides: (Photonic crystals, Plasmon waveguides)

Learning Experience

1. The course will combine traditional lectures with interactive digital tools like virtual labs, 3D simulations, and video tutorials.

No. of Hours: 15

- 2. Students will participate in hands-on experiments, focusing on nanomaterial synthesis, characterization, and quantum mechanics applications.
- 3. Group projects will encourage students to work together on designing experiments, analyzing nanomaterials, and solving problems related to nanoscale phenomena.
- 4. Students will complete regular assignments that challenge them to apply course concepts, such as solving quantum mechanics problems or designing synthesis techniques for nanomaterials.
- 5. The course instructor will be available during office hours for additional support and guidance.

Textbooks:

1. Nanostructures and Nanomaterials: Synthesis, Properties and Applications, G, Cao, Imperial College Press (2003).

Suggested Readings:

- 1. Nanoscale Science and Technology, Robert W. Kelsall, Ian W. Hamley and Mark Geoghegan, John Wiley & Sons Ltd (2005).
- Nanomaterials and Nanochemistry, C. Brechignac, P. Houdy, M. Lahmani, Springer-Verlag Berlin Heidelberg (2007).
- 3. Introduction to Nanoscale Science and Technology, Massimiliano Di Ventra, Stephane Evoy and James R. Heflin, Jr., Kluwer Academic Publishers (2004)
- Springer handbook of nanotechnology, Bharat Bhushan (ed.) Spinger-Verlag Berlin Heidelberg New York (2004)

Open Educational Resources (OER):

- 1. https://news.mit.edu/2015/explained-chemical-vapor-deposition-0619
- 2. <u>https://www.nanowerk.com/nanotechnology/introduction/introduction_to_nanotechno</u> <u>logy_22.php</u>
- 3. <u>https://www.youtube.com/watch?v=aOVU2aqgqe8</u>
- 4. <u>https://www.youtube.com/watch?v=dw9IvpilfUo</u>
- 5. <u>https://www.youtube.com/watch?v=1WGEMYDLsNs</u>
- 6. <u>https://en.wikipedia.org/wiki/Nanocomposite#:~:text=Nanocomposite%20is%20a%2</u> <u>0multiphase%20solid,that%20make%20up%20the%20material</u>.

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

SEMESTER VI					
UNS108	Synthesis of Nanomaterials Lab	L	Τ	Р	С
Version 1.0		0	0	4	2
Category of Course	Minor Lab				
Total Contact Hours	30 hours				
Pre-Requisites/ Co-	Synthesis of Nanomaterials				
Requisites					

Course Perspective:

The course focuses on practical techniques for synthesizing and characterizing various materials, including Polystyrene films, magnetite particles, and ferrofluids. Students will gain hands-on experience in preparing and analyzing materials, studying their stability and optical properties, and using methods like solid-state reactions and Archimedes' principle for density measurement. The course combines theoretical knowledge with laboratory skills to provide a thorough understanding of material preparation and characterization.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Observing various material preparation techniques such as the solvent evaporation method for polystyrene film and the coprecipitation method for magnetite particles to understand the steps involved in synthesizing materials.

CO 2: Imitating the procedures for preparing ferrofluid, synthesizing calcium titanate, and forming pellets, by following the guided steps for solid-state reactions, sintering, and polishing.

CO 3: Practising techniques for measuring the optical band gap, density of pellets using Archimedes' principle, and testing the stability of magnetite particles in various media, enhancing hands-on experimental skills.

Course Content

- 1. Preparation of Polystyrene film using solvent evaporation technique.
- 2. Preparation of magnetite particles using coprecipitation method.
- 3. Preparation of ferrofluid using water and magnetite particles,
- 4. 4 Study the stability of magnetite particles in water and vegetable oil.
- 5. Prepare calcium titanate using solid state reaction method.
- 6. 6 Find the optical band gap of magnetite particles.
- 7. Make pallet of calcium titanate powder, sinter and polish.
- 8. Find the density of pallet using Archimedes principle.

Learning Experience:

- 1. Students will engage in practical lab sessions to synthesize and analyze nanomaterials.
- 2. Group activities will include collaborative projects where students prepare nanomaterials and conduct joint analysis.
- 3. Peer reviews and group discussions will encourage teamwork, knowledge sharing, and problem-solving.
- 4. The instructor will provide regular feedback on assignments and lab work, with office hours available for additional support.
- **5.** Students will be encouraged to seek help and collaborate with peers, fostering a supportive learning environment.

Textbooks:

 Nanostructures and Nanomaterials: Synthesis, Properties and Applications, G, Cao, Imperial College Press (2003).

Suggested Readings:

- 1. Nanoscale Science and Technology, Robert W. Kelsall, Ian W. Hamley and Mark Geoghegan, John Wiley & Sons Ltd (2005).
- Nanomaterials and Nano chemistry, C. Brechignac, P. Houdy, M. Lahmani, Springer-Verlag Berlin Heidelberg (2007).
- 3. Introduction to Nanoscale Science and Technology, Massimiliano Di Ventra, Stephane Evoy and James R. Heflin, Jr., Kluwer Academic Publishers (2004)
- 4. Springer handbook of nanotechnology, Bharat Bhushan (ed.) Spinger-Verlag Berlin Heidelberg New York (2004).

Open Educational Resources (OER):

- 1. <u>https://doi.org/10.1021/ma001440d</u>
- 2. https://pubs.acs.org/doi/abs/10.1021/ma000094x
- 3. https://www.sciencedirect.com/science/article/abs/pii/S0927775708000721
- 4. https://www.sciencedirect.com/science/article/abs/pii/S0167577X08005740
- 5. https://pubs.acs.org/doi/abs/10.1021/ed076p943
- 6. https://www.sciencedirect.com/science/article/abs/pii/S0304885305011406
- 7. https://www.sciencedirect.com/science/article/abs/pii/S0021979705004935
- 8. https://www.sciencedirect.com/science/article/abs/pii/S0021979705005515
- 9. https://doi.org/10.1063/1.108974
- 10. <u>https://link.springer.com/article/10.1007/s10853-006-0103-y</u>
- 11. <u>https://www.youtube.com/watch?v=4q9Bh48RTxg</u>
- 12. https://www.youtube.com/watch?v=YpbNyDzpB3A

Evaluation Scheme:

Evaluation components	Weightage

Internal marks (practical)	
I. Conduct of experiment	10 Marks
II. Lab Record	10 Marks
III. Lab Participation	10 Marks
II. External Marks (practical):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER VII					
UNS109	Characterization Techniques of Nanomaterials	L	T	Р	С
Version 1.0		4	0	0	4
Category of Course	Minor				
Total Contact Hours	60 hours				
Pre-Requisites/ Co- Requisites	Basic knowledge in Materials Scienc and Nanotechnology.	e, Phy	ysics,	Chem	nistry,

Course Perspective:

This course is an introduction to the principles of instrumental techniques for characterization of nanomaterials. This course aims to teach the students the underlying principles of analytical techniques that are commonly used for the evaluation of structural, morphological, optical, thermal, mechanical and electrical properties of nanomaterials.

Course Outcomes:

Upon completion of the course, the learner will be:

CO1: Understanding the basic principles of different characterization techniques to study material's properties.

CO2: Applying the knowledge learned to determine the appropriate characterization technique for a given material or situation.

CO3: Analyzing the experimental data obtained from different characterization techniques to determine material properties.

CO4: Evaluating mechanical, magnetic, electrical, and thermal properties of different materials using different characterization techniques.

Course Content:

Unit 1 Basic of Characterization Techniques

Types of characterization techniques, Basics, Importance. Structural and compositional characterization tools, Difference between Microscopy and Spectroscopy, Optical Microscopy, Atomic Force Microscopy, Scanning Electron Microscopy, Transmission electron Microscopy, Scanning Tunnelling Microscopy.

Unit 2 Spectroscopy

UV visible spectroscopy, Infrared Spectroscopy and Fourier Transform Infrared Spectroscopy, Raman Spectroscopy, Photoluminescence (PL), Photoelectron Spectroscopy (X-Ray Photoelectron Spectroscopy, Auger Electron Spectroscopy & Ultraviolet Photoelectron Spectroscopy).

Unit 3 X-ray techniques

X-ray diffraction (XRD) technique, particle size determination using XRD, Applications of XRD, Electron diffraction and its application, neutron diffraction and its applications, X-Ray Photoelectron Spectroscopy, Auger Electron Spectroscopy.

Unit 4 Mechanical, Magnetic, electrical and Thermal properties measurement No. of Hours: 15

Nanoindentation principles- elastic and plastic deformation -mechanical properties of materials in small dimensions, Hardness testing of thin films and coatings, Vibration Sample Magnetometer, Impedance Spectroscopy- PPMS, - Measurement of Magnetic and electrical properties of Nanomaterials, Dielectric constant measurement, Differential Thermal Analysis (DTA), Differential scanning calorimetry (DSC).

Learning Experience:

No. of Hours: 15

No. of Hours: 15

No. of Hours: 15

- 1. The course will use multimedia tools, including animations and simulations, to explain complex concepts such as quantum confinement, thin film deposition, and nanocomposites.
- 2. Recorded lectures and online resources will also be available for self-paced learning.
- 3. Students will work in groups to analyze specific nanotechnology applications, conduct experiments, and develop innovative solutions.
- 4. Regular assignments will be given to reinforce key concepts, such as the classification of conducting polymers and thin film deposition methods.
- **5.** The course instructor will be available for additional support during office hours and will provide timely feedback on student progress.

Textbooks:

 Nanomaterials, Nanotechnologies and Design: An Introduction for Engineers, Daniel L, Schodek, Paulo Ferreira, Michael F. Ashby, Elsevier, 2009

Suggested Readings:

- 1. Elements of X-ray Diffraction B. D. Cullity, Addison Wesley, 1977
- 2. Transmission Electron Microscopy: A Textbook for Materials Science
- 3. David B Williams, C Barry Carter, (1996) Plenum Press, New York
- 4. Impedance Spectroscopy: Theory, Experiment, and Applications,
- 5. E. Barsoukov and J. Ross Macdonald (Editors) (2000) John Wiley & Sons (P)Ltd.
- Fundamentals of Fourier Transform Infrared Spectroscopy, Brian C Smith, (1995) CRC Press

Open Educational Resources (OER)"

- 1. <u>https://ocw.mit.edu/courses/materials-science-and-engineering/3-14-materials-</u> laboratory-for-engineers-spring-2009/
- 2. https://nanohub.org/
- <u>https://phys.libretexts.org/Bookshelves/Physical_and_Theoretical_Chemistry_Textbook_Maps/Supplemental_Modules_(Physical_and_Theoretical_Chemistry)/Spectroscopy</u>
- 4. https://www.msm.cam.ac.uk/teaching/part-ii-courses/characterisation-materials
- 5. https://www.merlot.org/merlot/viewMaterial.htm?id=637562

- 6. https://www.khanacademy.org/science/physics/light-waves
- 7. http://demonstrations.wolfram.com/XRayDiffractionOfCrystals/
- 8. https://ncl.cancer.gov/resources/assay-cascade-protocols
- 9. https://openstax.org/details/books/college-physics-ap-courses
- 10. https://www.coursera.org/learn/material-behavior

Evaluation Scheme:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade.

MINOR- ENVIRONMENTAL SCIENCE

COURSE PERSPECTIVE

POOL OF ENVIRONME	ENTAL SCIENCE AS MINOR				
UEV101	EARTH AND EARTH SURFACE PROCESSES	L	Τ	Р	С
Version 1.0		4	0	0	4
Total Contact Hours	60				
Pre-requisites/Exposure	Earth and its processes				
Co-requisites					

This course provides an in-depth understanding of Earth's history, processes, and materials, equipping students with essential knowledge for careers in geology, environmental science,

and related fields. It covers the formation of Earth, plate tectonics, rock cycles, and surface processes, enabling students to analyze and interpret geological phenomena. The skills gained, such as mineral identification, understanding seismic activities, and assessing erosion impacts, are crucial for addressing real-world challenges like resource management, disaster preparedness, and environmental conservation. By linking theoretical concepts with practical applications, the course prepares students for professional roles that require expertise in Earth sciences and supports informed decision-making in various environmental contexts.

COURSE OUTCOMES (CO)

On completion of this course, the student will be:

CO1: Understanding the Earth's history, including how the solar system formed and the layers of the Earth.

CO2: Explaining how Earth processes work, such as plate tectonics, earthquakes, and volcanic activity, with examples like the Himalayas.

CO3: Applying knowledge of minerals and rocks, including how they form and change through the rock cycle, weathering, and erosion.

CO4: Analyzing processes on the Earth's surface, including atmospheric changes, land interactions, and the effects of rivers and glaciers.

CO5: Evaluating the importance of mountains in India, including their role in river systems, monsoon patterns, and the development of human civilization.

15

COURSE CONTENT

Unit 1: History of Earth

lectures

Solar system formation and planetary differentiation; formation of the Earth: formation and composition of core, mantle, crust, atmosphere and hydrosphere; chemical composition of Earth; geological time scale and major changes on the Earth's surface; Holocene and the emergence of humans, role of humans in shaping landscapes; development of cultural landscapes.

Unit2: Earth system processes

15lectures

Movement of lithosphere plates; mantle convection and plate tectonics, major plates and hot spots, plate boundaries; sea floor spread; earthquakes; volcanic activities; orogeny; isostasy; gravitational and magnetic fields of the earth; origin of the main geomagnetic field; continental drift, Pangaea and present-day continents, paleontological evidence of plate tectonics; continental collision and mountain formation with specific example of the Himalaya.

Unit3: Minerals and rocks

lectures

Minerals and important rock forming minerals; rock cycle: lithification and metamorphism; Three rock laws; rock structure, igneous, sedimentary and metamorphic rocks; weathering: physical, biogeochemical processes; erosion: physical processes of erosion, factors affecting erosion; agents of erosion: rivers and streams, glacial and aeolian transportation and deposition of sediments by running water, wind and glaciers.

Unit4: Earth surface processes

lectures

Atmosphere: evolution of earth's atmosphere, composition of atmosphere, physical and optical properties, circulation; interfaces: atmosphere–ocean interface, atmosphere–land interface, ocean–land interface; land surface processes: fluvial and glacial processes, rivers and geomorphology; types of glaciers, glacier dynamics, erosional and depositional processes and glaciated landscapes; coastal processes.

Unit5: Importance of being a mountain

Formation of Peninsular Indian Mountain systems - Western and Eastern Ghats, Vindhyas, Aravallis, etc. Formation of the Himalaya; development of glaciers, perennial river systems and evolution of monsoon in Indian subcontinent; formation of Indo-Gangetic Plains, arrival of humans; evolution of Indus Valley civilization; progression of agriculture in the Indian subcontinent in Holocene; withdrawing monsoon and lessons to draw.

Learning Experience

This course integrates lectures, interactive sessions, and practical exercises to explore Earth's history, processes, and landscapes.

Instruction Methods:

• Lectures: Cover topics like Earth's formation, plate tectonics, minerals, rock cycles, and surface processes.

• Interactive Sessions: Engage through Q&A, quizzes, and discussions.

Technology Use:

• Online Platforms: LMS for lecture resources, recorded content, and discussions. Assessments:

• Formative: Quizzes and discussions for continuous feedback.

• Summative: Exams and presentations to assess overall understanding. Support:

10 lectures

10

• Instructor support and peer collaboration, with regular feedback to help students achieve learning outcomes.

Textbooks

1. Bridge, J., & Demicco, R. 2008. *Earth Surface Processes, Landforms and Sediment deposits*. Cambridge University Press.

Duff, P. M. D., & Duff, D. (Eds.). 1993. *Holmes' Principles of Physical Geology*. Taylor & Francis.

Suggested Readings

3. Gupta, A. K., Anderson, D. M., Pandey, D. N., & Singhvi, A. K. 2006. Adaptation and human migration, and evidence of agriculture coincident with changes in the Indian summer monsoon during the Holocene. *CurrentScience*90: 1082-1090.

4. Keller, E.A. 2011. *Introduction to Environmental Geology* (5thedition). Pearson Prentice Hall.

5. Krishnan, M. S. 1982. *Geology of India and Burma*.CBS Publishers & Distributors.

6. Leeder, M., Arlucea, M. P. 2005. *Physical Processes in Earth and Environmental Sciences*. Blackwell Publishing.

7. Pelletier, J. D. 2008. *Quantitative Modeling of Earth Surface Processes* (Vol. 304). Cambridge: Cambridge University Press. Chicago

Open Educational Resources (OER)

- <u>CrashCourse Earth Science</u>
- <u>PBS Eons</u>
- <u>Khan Academy Earth Science</u>
- <u>NOVA PBS Geology Playlist</u>
- <u>Geology Kitchen</u>
- <u>Rocks and Minerals Education</u>
- <u>MinuteEarth</u>
- <u>TED-Ed Earth and Space Science</u>
- <u>National Geographic</u>
- <u>The Science Channel</u>

Evaluation

Evaluation components

Weightage

Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

POOL OF ENVIRONMEN	NTAL SCIENCE AS MINOR				
UEV102	Hydrology and Hydrogeology	L	T	Р	C
Version 1.0		4	0	0	4
Total Contact Hours	60				
Pre-requisites/Exposure	Basics of Chemistry				
Co-requisites					

Course Perspective

This course offers a comprehensive understanding of hydrology and hydrogeology, essential for students pursuing careers in water resources management, environmental engineering, and related fields. By covering the fundamentals of the hydrological cycle, groundwater flow, and water resource management, students gain the knowledge needed to analyze and address water-related challenges. The course also equips students with practical skills in hydrological data collection, analysis, and modeling, including the use of Geographic Information Systems (GIS). Through the study of advanced topics like climate change impacts, flood management, and groundwater exploration, students are prepared to apply their knowledge in real-world scenarios, contributing to sustainable water resource management and informed decision-making in environmental and governmental roles.

Course Outcomes

On completion of this course, the student will be:

CO1: Understanding key concepts in hydrology and hydrogeology, including the hydrological cycle, precipitation, and groundwater flow.

CO2: Applying techniques to collect and analyze hydrological data, such as measuring precipitation, streamflow, and soil moisture.

CO3: Analyzing water resource management strategies, including planning, conservation, and quality monitoring, to address issues like groundwater pollution.

CO4: Evaluating the effectiveness of hydrological models and flood management practices and assessing the impact of climate change on hydrological processes.

CO5: Creating and applying advanced hydrological models, GIS tools, and policy frameworks to address complex problems in water resource management.

CO6: Reviewing case studies to identify best practices and innovative solutions in hydrology and hydrogeology.

COURSE CONTENT

Unit 1: Fundamentals of Hydrology and Hydrogeology	15 lectures
Introduction to Hydrology and Hydrogeology	
Basic concepts and definitions in hydrology and hydrogeology	
The hydrological cycle and its components	
Precipitation and evapotranspiration processes	
Infiltration and soil water storage	
• Surface water hydrology: runoff generation and streamflow analysis	
• Groundwater hydrology: aquifers, groundwater flow, and well hydraulics	
Unit 2: Hydrological Data Collection and Analysis	15 lectures
Collection and analysis of hydrological data	
Measurement and analysis of precipitation	
Soil moisture measurement techniques	
Streamflow measurement and hydrograph analysis	
Groundwater level measurement and well hydraulics	
Statistical methods for data analysis in hydrology	
Introduction to hydrological modeling techniques	
Unit 3: Water Resources and Management	15 lectures
• Water resources planning and allocation	
Integrated water resources management principles	

• Water conservation and demand management

- Water quality parameters and standards
- Groundwater pollution and remediation techniques
- Water quality monitoring and assessment
- Introduction to Geographic Information Systems (GIS) in hydrology
- Unit 4: Advanced Topics in Hydrology and Hydrogeology
- Hydrological modeling: model calibration, validation, and applications
- Climate change impacts on hydrological processes
- Flood frequency analysis and floodplain management
- Groundwater exploration techniques
- Spatial analysis and modeling of hydrological data using GIS
- Water policy, governance, and legal frameworks
- Case studies and applications in hydrology and hydrogeology

Learning Experience

This course combines lectures, interactive sessions, and practical exercises to cover hydrology and hydrogeology fundamentals, data analysis, and advanced topics.

Instruction Methods:

• Lectures: Cover basics of hydrology, data collection, water management, and advanced topics.

• **Interactive Sessions:** Engage through Q&A, quizzes, and discussions.

Technology Use:

• **Online Platforms:** LMS for resources, recorded lectures, and discussions.

Assessments:

- Formative: Quizzes and discussions for ongoing feedback.
- **Summative:** Exams, peer reviews, and presentations for overall evaluation.

Support:

• Instructor guidance and peer collaboration with regular feedback to achieve learning outcomes.

Text Books

1. K Subramanya, Engineering Hydrology, Mc-Graw Hill. New Delhi.

2. K N Muthreja, Applied Hydrology, Tata Mc-Graw Hill.

Reference Books/Materials

3. K Subramanya, Water Resources Engineering through Objective Questions, Tata McGraw Hill.

4. G L Asawa, Irrigation Engineering, Wiley Eastern

15lectures

- 1. L W Mays, Water Resources Engineering, Wiley.
- 2. J D Zimmerman, Irrigation, John Wiley & Sons
- 3. C S P Ojha, R Berndtsson and P Bhunya, Engineering Hydrology, Oxford.

4. R.K. Sharma and T.K. Sharma, Hydrology and Water Resources Engineering, Prentice Hall of India, New Delhi.

Open Educational Resources (OER)

- <u>https://www.usgs.gov/special-topic/water-science-school</u>
- <u>https://www.coursera.org/learn/hydrology-hydrogeology</u>
- <u>https://ocw.mit.edu/courses/civil-and-environmental-engineering/1-72-</u>

groundwater-hydrology-spring-2003/

- <u>https://www.epa.gov/water-research/hydrology-research</u>
- <u>http://hydrogeo.uky.edu/</u>
- <u>https://www.bgs.ac.uk/research/groundwater/hydrology.html</u>
- <u>https://ocw.un-ihe.org/courses</u>
- <u>https://oer2go.org/mods/en-boundless/www.boundless.com/environmental-</u>

science/textbooks/boundless-environmental-science-textbook/water-resources-

5/hydrology-and-water-resources-50/hydrology-245-10941/index.html

- <u>https://hydrology.berkeley.edu/ce-170.html</u>
- <u>https://www.indiawaterportal.org/articles/groundwater-and-hydrogeology-</u>

<u>introduction</u>

Modes of Evaluation

Evaluation components		Weightage
Internal marks (Theory)		30 Marks
I. Continuous assessment (30 marks)		
All the components to be evenly spaced		
Project/quizzes/assignment and	essays/presentation/	
II. Internal marks(Theory):		20 Marks
Mid Term Examination		
III. External Marks (Theory):		50 Marks
End Term Examination		

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

POOL OF ENVIRON	MENTAL SCIENCE AS MINOR				
UEV103	NATURAL RESOURCES MANAGEMENT	4	0	0	4
	AND SUSTAINABILITY				
Version 1.0					<u> </u>
Total Contact Hours	60				
Pre-	Environmental studies				
requisites/Exposure					
Co-requisites					

COURSE PERSPECTIVE

This course explores the classification, availability, and conservation of resources, both renewable and non-renewable. Investigate mineral resources, their identification, extraction methods, and the global consumption patterns that shape our world. Gain insights into energy resources, from oil and gas to renewable sources like solar, wind, and nuclear power, assessing their environmental impacts and potential. Understand resource management approaches, integrated strategies, and sustainability science principles to address real-world challenges. Through a balanced mix of theory, case studies, and discussions, develop a holistic understanding of how our actions impact the planet and how responsible resource management can shape a sustainable future.

COURSE OUTCOMES (CO)

On completion of this course, the student will be:

CO1: Understanding the types of natural resources, including renewable and non-renewable resources, and how they are affected by human activities.

CO2: Applying knowledge of forest, water, soil, and food resources, including their importance and how to conserve them.

CO3: Analyzing mineral resources, their extraction methods, and the environmental impacts of mining.

CO4: Evaluating non-renewable energy sources like oil, natural gas, and coal, including their formation, extraction, and environmental effects.

Course Content

Unit1: Introduction

10 lectures

Resource and reserves; classification of natural resources; renewable and non-renewable resources; resource degradation; resource conservation; resource availability and factors

influencing its availability; land resources; water resources; fisheries and other marine resources; energy resources; mineral resources; human impact on natural resources; ecological, social and economic dimension of resource management.

Unit2: Natural resources and conservation

lectures

Forest resources: economic and ecological importance of forests, forest management strategies, sustainable forestry; water resources: supply, renewal, and use of water resources, fresh water shortages, strategies of water conservation; soil resources: importance of soil, soil conservation strategies; food resources: world food problem, techniques to increase world food production, green revolution.

Unit3: Mineral resources

lectures

Mineral resources and the rock cycle; identified resources; undiscovered resources; reserves; types of mining: surface, subsurface, open-pit, dredging, strip; reserve-to-production ratio; global consumption patterns of mineral resources techniques to increase mineral resource supplies; ocean mining for mineral resources; environmental effects of extracting and using mineral resources.

Unit4: Non-renewable energy resources

Oil: formation, exploration, extraction and processing, oil shale, tar sands; natural gas: exploration, liquefied petroleum gas, liquefied natural gas; coal: reserves, classification, formation, extraction, processing, coal gasification; environmental impacts of non renewable energy consumption; impact of energy consumption on global economy; application of green technology; future energy options and challenges.

Unit 5: Renewable energy resources

lectures

Energy efficiency; life cycle cost; cogeneration; solar energy: technology, advantages, passive and active solar heating system, solar thermal systems, solar cells, J N N solar mission; hydropower: technology, potential, operational costs, benefits of hydropower development; nuclear power: nuclear fission, fusion, reactors, pros and cons of nuclear power, storage of radioactive waste, radioactive contamination; tidal energy; wave energy; ocean thermal energy conversion (OTEC); geothermal energy; energy from biomass; bio-diesel.

Unit6: Resource management

Approaches in resource management: ecological approach; economic approach; ethnological approach; implications of the approaches; integrated resource management strategies; concept

10 lectures

10 lectures

10

10

10

of sustainability science: different approach towards sustainable development and its different constituents; sustainability of society, resources and framework; sustainable energy strategy; principles of energy conservation; Indian renewable energy programme.

Learning Experience

This course features lectures, interactive discussions, and hands-on activities on resource management and conservation.

- Instruction Methods: Lectures and discussions on core concepts.
- **Technology Use:** LMS for resources and forums.
- Assessments: Quizzes, exams, and projects.
- **Support:** Instructor guidance and peer collaboration.

Suggested Text Books

1. Craig, J. R., Vaughan. D. J. & Skinner. B. J. 1996. *Resources of the Earth: Origin, Use, and Environmental Impacts* (2nd edition). Prentice Hall, New Jersey.

2. Freeman, A. M. 2001. *Measures of value and Resources: Resources for the Future*. Washington DC.

Reference Books/Materials

3. Freeman, A. M. 2003. *Millennium Ecosystem Assessment: Conceptual Framework*. Island Press.

4. Ginley, D. S. & Cahen, D. 2011. Fundamentals of Materials for Energy and Environmental Sustainability. CambridgeUniversity Press.

5. Klee, G. A. 1991. *Conservation of Natural Resources*. Prentice Hall Publication.

6. Miller, T. G. 2012. *Environmental Science*. Wadsworth Publishing Co.

7. Owen, O. S, Chiras, D. D, & Reganold, J. P. 1998. *Natural Resource Conservation– Management for Sustainable Future* (7thedition). Prentice Hall.

8. Ramade, F. 1984. *Ecology of Natural Resources*. John Wiley & Sons Ltd.

9. Tiwari, G. N. & Ghosal. M. K. 2005. *Renewable Energy Resources: Basic Principles and Application*. Narosa Publishing House.

Open Educational Resources (OER)\

https://www.k-state.edu/nrm/webinars/intro/index.html)

https://open.oregonstate.education/woodproducts/)

https://www.sc.edu/study/colleges_schools/artsandsciences/environment_and_sustainabi

lity/academics/courses/introduction to geology.php

https://www.energy.gov/eere/education/energy-education-resources

https://www.umass.edu/windenergy/education/renewable-energy-and-environmentalsustainability

https://ocw.uci.edu/collections/7ba20ee3-0e56-46a5-b6c2-334eb7cb8c10

Assessment & Evaluation

Modes of Evaluation

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

POOL OF ENVIRONME	NTAL SCIENCE AS MINOR				
UEV104	NATURAL AND ANTHROPOGENIC HAZARDS	L	Т	Р	C
Version 3.0		4	0	0	4
Total Contact Hours	60				<u></u>
Pre-requisites/Exposure	Baics of Environment				
Co-requisites					

COURSE PERSPECTIVE

This course explores the causes and impacts of natural and anthropogenic hazards, including earthquakes, tsunamis, volcanic eruptions, and pollution. Through case studies and scientific analysis, students will learn about disaster prediction, mitigation strategies, and the influence of human activities on environmental stability. The course aims to equip students with the knowledge to understand and reduce the risks associated with these hazards.

COURSE OUTCOMES (CO)

On completion of this course, the student will be:

CO1: Understanding natural hazards and disasters, including their causes, human impacts, and methods for predicting and mitigating these events.

CO2: Applying knowledge of volcanic hazards, landslides, floods, and storms, and assessing their effects on communities and the environment.

CO3: Analyzing anthropogenic disasters related to pollution, population growth, and soil degradation, and understanding their impacts on ecosystems.

CO4: Evaluating water and atmospheric pollution through legislation like the Air Act and Water Act and studying case examples to understand their implications.

CO5: Reviewing case studies of natural and anthropogenic disasters to identify effective mitigation strategies and improve disaster preparedness.

COURSE CONTENT

Unit A NATURAL AND ANTHROPOGENIC HAZARDS 15 lectures

Natural Hazards and Disasters, Human Impact on Natural Disaster, Predicting Catastrophe, Mitigating Hazards; Plate Tectonics and related Hazards, Earthquakes and their causes, Ground Motion and Failures, Case study of Nepal earthquake and Bhuj earthquake; Tsunami: Giant Tsunamis, Generation and Movement, Tsunami Hazard Assessment, Tsunami – 2004, Fukushima disaster

Unit B Volcanic Hazard

Eruption-Type of Volcanoes and Tectonic environment; Landslide and their causes, Type of downslope movement, associated hazard, Land Subsidence and associated hazard; Floods and Human Interaction, Flood Frequency and Recurrence Interval, Human intervention and mitigation; Storms: Tropical Cyclone, Hurricane, Tornado, Storm damage and safety; Wildfires: Fire Process and Secondary effects; Case studies of devastating natural hazards

Unit C Anthropogenic Disasters I

Pollution: Role of natural and anthropogenic factors; Population growth and Environmental Impact; Carrying capacity of ecosystem; Soil and soil degradation, desertification, Ways to improve soil and case studies related to soil degradation

Unit D Anthropogenic Disasters II

Fundamental concepts of water and atmospheric pollution, Air Act, Water Act, Ambient Air quality, case studies related to water and atmospheric pollution; Waste and Hazardous

15 lectures

15 lectures

15 lectures

Learning Experience

The course integrates lectures, case studies, and interactive discussions to explore natural and anthropogenic hazards.

• **Instruction Methods:** Lectures on various hazards, their causes, and mitigation strategies; case studies for real-world understanding.

• **Technology Use:** LMS for resources, recorded lectures, and discussion forums.

• Assessments: Quizzes, exams, and case study analyses to assess comprehension.

• **Support:** Instructor support and peer collaboration encouraged for deeper understanding.

Suggested Textbooks

1. Natural Hazards and Disasters" by D. Hyndman & D. Hyndman:

2. "Introduction to Environmental Geology" by E. A. Keller

Reference Books/Materials

3. "Environmental Hazards: Assessing Risk and Reducing Disaster" by K. Smith

4. "Introduction to Environmental Engineering and Science" by G. M. Masters & W. P. Ela

5. "Natural Hazards: Earth's Processes as Hazards, Disasters, and Catastrophes" by E. A. Keller & D. E. DeVecchio

6. "Environmental Science: Earth as a Living Planet" by D. B. Botkin& E. A. Keller Open Educational Resources (OER)

- <u>https://openstax.org/details/books/earth-science</u>
- <u>https://www.nssl.noaa.gov/education</u>
- <u>https://www.usgs.gov/natural-hazards/earthquake-hazards</u>
- <u>https://volcano.si.edu</u>
- <u>https://www.fema.gov/</u>
- <u>http://www.unesco.org/new/en/education/themes/education-building-blocks/disaster-</u>

risk-reduction/resources/

- <u>https://open.umn.edu/opentextbooks/textbooks?subject_area_id=28</u>
- <u>https://www.merlot.org/merlot/materials.htm?category=2665</u>

Evaluation

Evaluation components	Weightage

Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

POOL OF ENVIRONMENTAL SCIENCE AS MINOR					
UEV105	ENVIRONMENTLEGISLATIONPOLICIESAND ESG'S	L	Т	Р	С
Version 3.0		4	0	0	4
Total Contact Hours	60			•	
Pre- requisites/Exposur e					
Co-requisites					

COURSE PERSPECTIVE

This course provides an in-depth study of environmental legislation, policies, and ESG principles. Students will explore the constitutional basis for environmental protection, the evolution of key laws, and the significance of legal definitions related to pollution, biodiversity, and sustainability. By analyzing major acts like the Indian Forest Act and the Environmental Protection Act, along with recent laws like the Biological Diversity Act, students will gain the knowledge needed to navigate environmental legal frameworks and contribute to sustainable practices.

COURSE OUTCOMES

On completion of this course, the student will be:

CO1: Understanding the Constitution of India and its provisions related to fundamental rights, duties, and the structure of government, including the roles of the legislature and judiciary.

CO2: Exploring the historical evolution of environmental legislation and policies in India, from ancient practices to contemporary laws, highlighting key periods and acts.

CO3: Analyzing legal definitions related to environmental concepts, including pollution, natural resources, and sustainable development, as outlined in the Indian Constitution.

CO4: Evaluating major legislative instruments that govern environmental protection in India, including the Forest Act, Wildlife Protection Act, and the Environment (Protection) Act, among others.

COURSE CONTENT

UnitA: Introduction

Constitution of India; fundamental rights; fundamental duties; Union of India; union list, state list, concurrent list; legislature; state assemblies; judiciary; panchayats and municipal bodies; National Green Tribunal.

UnitB: History of environmental legislation and policy

of water, Ancient period: worship air, trees; Mauryan period: Kautilya's Arthashastra, Yajnavalkya smriti and Charaksamhita; Medieval period: forests as woodland and hunting resources during Mughal reign; British India: Indian Penal Code 1860, Forest Act 1865, Fisheries Act1897; Independent India: Van Mahotsava 1950, National Forest Policy 1952, Orissa River pollution and prevention Act 1953.

UnitC: Environmental legislation

Legal definitions (environmental pollution, natural resource, biodiversity, forest, sustainable development); Article 48A (The protection and improvement of environment and safeguarding of forests and wildlife); Article51 A (Fundamental duties).

UnitD: Legislative Instruments

The Indian Forest Act 1927; The Wildlife (Protection) Act 1972; The Water (Prevention and Control of Pollution) Act 1974; The Forests (Conservation) Act 1980; The Air (Prevention and Control of Pollution) Act 1981; The Environment (Protection) Act 1986; Motor Vehicle Act 1988; The Public Liability Insurance Act 1991; Noise Pollution (Regulation and Control) Rules 2000; The BiologicalDiversity Act 2002; The Schedule Tribes and other Traditional Dwellers (Recognition of Forests Rights) Act 2006; The National Green Tribunal Act 2010; scheme and labeling of environment friendly products, Ecomarks.

Learning Experience

20lectures

10lectures

10 lectures

20 lectures

The course features lectures, interactive sessions, and practical exercises to explore environmental legislation and policies.

Instruction Methods:

- Lectures: Multimedia presentations and problem-solving.
- Interactive Sessions: Q&A, quizzes, and discussions.

Technology Use:

• **Online Platforms:** LMS for resources and discussions.

Assessments:

- **Formative:** Quizzes and discussions.
- **Summative:** Exams, peer reviews, and presentations.

Support: Instructor guidance and peer collaboration with regular feedback.

Suggested Textbooks

- 1. "Environmental Law in India" by Shyam Diwan and Armin Rosencranz
- 2. "Environmental Law" by Bimal N. Patel

Reference Books/Materials

- 3. "Environmental Management: Text and Cases" by Rajagopalan Raman
- 4. "Environmental Laws in India: An Introduction" by Gurdip Singh
- 5. "Environmental Legislation and Policy: Selected Statutes" by Gitanjali Nain Gill
- 6. "Environmental Governance in India: Problems and Perspectives" by N.C. Saxena and

Kanchi Kohli

7. "Environmental Law: Pollution and Management" by Suresh P. Harsha and PallaviBedi Open Educational Resources (OER)

- <u>https://openstax.org/details/books/earth-science</u>
- <u>https://www.nssl.noaa.gov/education</u>
- <u>https://www.usgs.gov/natural-hazards/earthquake-hazards</u>
- <u>https://volcano.si.edu</u>
- <u>https://www.fema.gov/</u>
- http://www.unesco.org/new/en/education/themes/education-building-blocks/disaster-

risk-reduction/resources/

- <u>https://open.umn.edu/opentextbooks/textbooks?subject_area_id=28</u>
- <u>https://www.merlot.org/merlot/materials.htm?category=2665</u>

Assessment & Evaluation

Evaluation components		Weightage
Internal marks (Theory)		30 Marks
I. Continuous assessment (30 marks)		
All the components to be evenly spaced		
Project/quizzes/assignment and	essays/presentation/	
II. Internal marks (Theory):		20 Marks
Mid Term Examination		
III. External Marks (Theory):		50 Marks
End Term Examination		

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination

separately to secure minimum passing grade

POOL OF ENVIRON	MENTAL SCIENCE AS MINOR				
UEV106	Waste Management	L	Т	Р	C
Version 3.0		4	0	0	4
Total Contact Hours	60	•			
Pre-	Basics of Environment				
Co-requisites					

COURSE PERSPECTIVE: This course provides a detailed understanding of solid waste management and resource recovery, covering waste sources, environmental impacts, and management techniques. Students will explore methods for handling municipal, hazardous, and biomedical waste, including collection, landfilling, thermal treatment, and recycling. The course also addresses industrial waste management, resource recovery practices, waste-to-energy processes, and integrated waste management strategies, alongside lifecycle assessment and relevant policies. This knowledge equips students with essential skills for effective waste management and sustainability.

COURSE OUTCOMES (CO)

On completion of this course, the student-teacher will be able to:

CO1: Understanding the sources and types of solid waste, including municipal, hazardous, and biomedical waste, and their impact on health and the environment.

CO2: Applying techniques for collecting, storing, transporting, and disposing of solid waste, including landfill design and thermal treatment methods.

CO3: Analyzing the effects of industrial waste on air, water, and soil, and understanding the importance of effective waste management.

CO4: Evaluating resource recovery methods, including the 4Rs (reducing, reusing, recycling, recovering) and biological processing techniques.

CO5: Creating energy from waste through processes like combustion, pyrolysis, and anaerobic digestion.

CO6: Reviewing policies for solid waste management and their role in promoting eco-friendly practices.

COURSE CONTENT

Unit A: Introduction

Sources and generation of solid waste, their classification and chemical composition; characterization of municipal solid waste; hazardous waste and biomedical waste.

Unit B: Effect of solid waste disposal on environment

8lectures

Impact of solid waste on environment, human and plant health; effect of solid waste and industrial effluent discharge on water quality and aquatic life; mining waste and land degradation; effect of landfill each ate on soil characteristics and ground water pollution.

Unit C: Solid waste Management

Different techniques used in collection, storage, transportation and disposal of solid waste (municipal, hazardous and biomedical waste); landfill (traditional and sanitary landfill design); thermal treatment (pyrolysis and incineration) of waste material; drawbacks in waste management techniques.

Unit D: Industrial waste management

8lectures

Types of industrial waste: hazardous and non-hazardous; effect of industrial waste on air, water andsoil; industrial waste management and its importance; stack emission control and emission monitoring; effluent treatment plant and sewage treatment plant.

UnitE:ResourceRecovery

6lectures

5lectures

9lectures

4R-reduce, reuse, recycle and recover; biological processing-composting, anaerobic digestion, aerobic treatment; reductive dehalogenation; mechanical biological treatment; green techniques for waste treatment.

Unit F: Waste- to-energy (WTE)

6lectures

Concept of energy recovery from waste; refuse derived fuel (RDF); different WTE processes: combustion, pyrolysis, landfillgas (LFG) recovery; anaerobic digestion; gasification.

UnitG: Integratedwastemanagement

6lectures

Concept of Integrated waste management; waste management hierarchy; methods and importance ofIntegratedwastemanagement.

UnitH: Life cycle assessment (LCA)

6lectures

Cradle to grave approach; lifecycle inventory of solid waste; role of LCA in waste management; advantage and limitation of LCA; case study onLCA of aproduct.

UnitI:Policiesforsolidwastemanagement 6lectures

Municipal Solid Wastes (Management and Handling) Rules 2000; Hazardous Wastes Management and Handling Rules 1989; Bio-Medical Waste (Management and Handling) Rules 1998; Ecofriendly orgreenproducts.

Suggested Text Books

1. "Solid Waste Management: Engineering Principles and Management Issues" by Tchobanoglous et al.

2. "Introduction to Solid Waste Management" by Tebbutt.

Reference Books/Materials

- 3. "Waste Management Practices: Municipal, Hazardous, and Industrial" by Pichtel.
- 4. "Handbook of Solid Waste Management" by Kreith and Tchobanoglous.
- 5. "Waste Management and Sustainable Consumption" by Cooper.
- 6. "Hazardous Waste Management" by LaGrega et al.
- 7. "Biomedical Waste Management: Principles and Case Study" by Bhandari.

Open Educational Resources (OER)

- <u>https://openstax.org/details/books/earth-science</u>
- <u>https://www.nssl.noaa.gov/education</u>
- <u>https://www.usgs.gov/natural-hazards/earthquake-hazards</u>
- <u>https://volcano.si.edu</u>

- <u>https://www.fema.gov/</u>
- <u>http://www.unesco.org/new/en/education/themes/education-building-blocks/disaster-</u>

risk-reduction/resources/

- <u>https://open.umn.edu/opentextbooks/textbooks?subject_area_id=28</u>
- <u>https://www.merlot.org/merlot/materials.htm?category=2665</u>

Assessment & Evaluation

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Evaluation component	ts	Wei	ight	age		
Internal marks (Theor	y)	30 1	Marl	cs		
I. Continuous assessm	ent (30 marks)					
All the components to b	be evenly spaced					
Project/quizzes/assignm	nent and essays/presentation/					
II. Internal marks(The	eory):	20 I	Mar	ks		
Mid Term Examination						
III. External Marks (T	Theory):	50 I	Mar	ks		
End Term Examination POOL OF ENVIRON	MENTAL SCIENCE AS MINOR					
UEV107	Environmental Impact assessment and R	Risk	L	Т	Р	С
Version 3.0			4	0	0	4
Total Contact Hours	60				<u> </u>	<u> </u>
Pre-	Environmental risk assessment					
Co-requisites						

Course Perspective

This course offers detailed exploration of Environmental Impact Assessment (EIA) and risk management, focusing on evaluating and mitigating environmental impacts to support sustainable development. Students will learn the fundamentals of EIA, including its methodologies, stakeholder roles, and the creation of Environmental Impact Statements (EIS) and Environmental Management Plans (EMP). The course also covers advanced topics such as Rapid EIA, Strategic Environmental Assessment, Social Impact Assessment, and life cycle

assessments, alongside the principles of environmental management and sustainable development. Additionally, students will examine EIA regulations in India, current challenges, and case studies of major projects, while developing skills in risk assessment, including hazard identification and risk communication. This comprehensive approach prepares students for careers in environmental consulting, project management, and policy development.

COURSE OUTCOMES (CO)

On completion of this course, the student will be:

CO1: Understanding the concepts and methodologies of Environmental Impact Assessment (EIA) and its significance in project planning.

CO2: Applying various assessment techniques, such as Rapid EIA and Social Impact Assessment, to evaluate environmental effects.

CO3: Analyzing EIA regulations and practices in India, identifying current challenges through case studies.

CO4: Evaluating risk assessment processes, including exposure assessment and hazard identification in environmental monitoring.

CO5: Creating effective Environmental Management Plans (EMP) and Environmental Impact Statements (EIS) based on impact predictions and baseline data.

15

15

15

COURSE CONTENT

Unit 1: Environmental impact assessment (EIA):

Lectures

Definitions, introduction and concepts; rationale and historical development of EIA; scope and methodologies of EIA; role of project proponents, project developers and consultants; Terms of Reference; impact identification and prediction; baseline data collection; Environmental Impact Statement (EIS), Environmental Management Plan(EMP)

Unit 2:

Lectures

Rapid EIA; Strategic Environmental Assessment; Social Impact Assessment; Cost-Benefitanalysis; Life cycle assessment; environmental appraisal; environmental management principles, problems and strategies; environmental planning; environmental audit; introduction to ISO and ISO14000; sustainable development.

Unit3:

Lectures

EIA regulations in India; status of EIA in India; current issues in EIA; case study of hydropower projects/ thermal projects.

Unit 4:

Lectures

Risk assessment: introduction and scope; project planning; exposure assessment; toxicity assessment; hazard identification and assessment; risk characterization; risk communication; environmental monitoring; community involvement; legal and regulatory framework; human and ecological risk assessment.

Learning Experience

This course combines lectures, interactive sessions, and hands-on activities for understanding environmental impact and risk assessment.

Instruction Methods:

- Lectures: Multimedia presentations on core concepts.
- Interactive Sessions: Q&A, quizzes, and discussions.

Technology Use:

• **Online Platforms:** LMS for resources and discussions.

Assessments:

- **Formative:** Quizzes and discussions.
- **Summative:** Exams, peer reviews, presentations.

Support: Instructor guidance, peer collaboration, and regular feedback.

Text Book

Barrow, C.J. 2000. Social Impact Assessment: An Introduction. Oxford University Press.

Reference Books/Materials

Glasson, J., Therivel, R., Chadwick, A. 1994. Introductionto EnvironmentalImpactAssessment. London,ResearchPress,UK.Judith, P. 1999. Handbook of EnvironmentalImpactAssessment. BlackwellScience.Marriott, B. 1997. EnvironmentalImpactAssessment: APracticalGuide. McGraw-Hill, NewYork,USA.

Open Educational Resources (OER)

- <u>United Nations Environment Programme (UNEP) EIA Training Resource Manual</u>
- International Association for Impact Assessment (IAIA) Resources
- World Bank Environmental and Social Framework
- Environmental Protection Agency (EPA) EIA Resources
- Asian Development Bank (ADB) Environmental Assessment Sourcebook

• <u>United Nations Economic Commission for Europe (UNECE) EIA Training</u>

<u>Materials</u>

• <u>World Health Organization (WHO) Environmental Impact Assessment</u> <u>Guidelines</u>

- Environmental Law Institute (ELI) EIA Resources
- <u>International Finance Corporation (IFC) EIA Guidelines</u>
- <u>United Nations Development Programme (UNDP) EIA Toolkit</u>

Assessment & Evaluation

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous accessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

POOL OF ENVIRONMENT	AL SCIENCE AS MINOR				
UEV108	SDG'S AND CLIMATE CHANGE	L	Т	Р	С
Version 1.0		4	0	0	4
Total Contact Hours	60				
Pre-requisites/Exposure	Basics of Chemistry				
Co-requisites					

Course Perspective

This course offers a comprehensive examination of global warming, climate change, and sustainable development, focusing on both theoretical frameworks and practical applications. Students will explore the socio-economic and environmental drivers of change, adaptation and

mitigation strategies, and the international and national efforts to reduce carbon emissions. The course also covers the Sustainable Development Goals, the challenges and opportunities of sustainable business practices, and the importance of governance in achieving sustainability. Through case studies, students will gain insights into climate risks, vulnerability assessments, and the role of responsible management in addressing environmental and societal challenges.

COURSE OUTCOMES

Upon completion of this course, students will be:

CO1: Understanding the causes and effects of global warming and climate change, including the role of human activities and socio-economic factors.

CO2: Applying adaptation and mitigation strategies for climate change at international and national levels, focusing on sustainable development practices.

CO3: Analyzing climate risks and vulnerabilities in India, using assessment tools and creating vulnerability maps for specific areas.

CO4: Evaluating the principles of sustainable development, including the Sustainable Development Goals (SDGs) and the dynamics involved in achieving them.

CO5: Assessing governance strategies for sustainable development, incorporating environmental management, corporate social responsibility, and risk management practices.

Course Content

Unit A:

lectures

Global Warming and Climate Change, Debate on Climate Change – the manifestations of Climate Change; Natural and anthropogenic (human interventions), Relationship between socioeconomic and environmental drivers of change (e.g. globalization, urbanization, land degradation, inefficient use of water, climate change), Climate change: Adaptation and Mitigation Strategies at International and national contexts, International and National Efforts at Carbon Emission Reductions, Global (environmental) change and sustainable development, and sustainable development with a focus on the specific situation in Central Asia

Case study 1: Assessment of climate risks and vulnerability in India, Presentation of national assessment results and vulnerability maps and preparation of an assessment in the pilot area.

Unit B

12 lectures

10

Sustainable Development in theory and practice, Global Responses to Sustainable Development, Sustainable Development Goals (vs Millennium Development Goals), The Paris and Post-Paris Convention on Climate Change and Sustainable Development, Triple Bottom line of Sustainability: Food, Water, Energy nexus, Potential and Barriers to Sustainable Business, Sustainable rural and urban livelihoods, Laying Out Actors and Dynamics in the 2030 Agenda for Sustainable Development

Unit C

8 lectures

Climate Risks and Vulnerability Assessment of India, Why environment and natural resources are prone to market failure, Values (Economic or otherwise) of Environment and Natural Resources: Use, Option, Existence, Signals of Natural Resource Depletion/ scarcity and valuation methods (Health cost, amenities and Hedonic Pricing, Travel Cost methods, Contingent Valuation Methods, Choice Experiments, Limitations of these signals), Payment for Ecosystem Services (PES), Combining Theories of Governing Societal Change towards Sustainability

Unit D

12 lectures

Governance Pillars and Competences: Power, Knowledge and Norms as Cross-Cutting Issues in Governance for the SDGs, Socially and Environmentally Responsible Business Management, The relevance of Green Growth Green Business paradigms, Environmental Values of Business, Corporate Social Responsibility and Environmental Impacts, Environmental Risk Management & Environmental Strategy, Environmental and Ecological Stewardship, Inferences on Improving Integrative Sustainability Governance

Case Study 1: Sustainable Disaster Risk Reduction in Mountain Agriculture: Agroforestry Experiences in Kaule, Mid-Hills of Nepal

Case study 2: Climate Change 2014, Impacts, Adaptation, and Vulnerability Part A: Global and Sectoral Aspects. Working Group II Contribution to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change.

Case study 3: Influence of Climate Change on Environmental Hazards and Human Well-Being in the Urban Areas—Warsaw Case Study Versus General Problems

Learning Experience

This course integrates lectures, interactive discussions, and case studies to explore climate change, sustainable development, and governance.

Instruction Methods:

- Lectures: Core topics delivered through presentations.
- Interactive Sessions: Discussions and case studies to apply concepts.

Technology Use:

• **Online Platforms:** LMS for resources.

Assessments:

• **Formative:** Quizzes and discussions.

• **Summative:** Exams, case study presentations, and reports.

Support: Instructor support, peer collaboration, and regular feedback.

Textbook

 Jacob Thomas, Environmental Management – Text and Cases, Dorling Kindersley (India) Pvt. Ltd. 2014.

References books

 Environmental Management, Sustainable Development and Human Health, (Eds.) [2009], Eddie N. Laboy-Nieves & Fred C. Schaffner; Ahmad H. Abdelhadi; Mattheus F. A. Goosen, CRC Press/Balkema is an imprint of the Taylor & Francis Group, London, UK, 596p.

Open Educational Resources (OER)

- <u>https://www.climate.gov/</u>
- <u>https://unfccc.int/</u>
- <u>https://sustainabledevelopment.un.org/</u>
- <u>https://ocw.mit.edu/index.htm</u>
- <u>https://www.open.edu/openlearn/science-maths-technology/environmental-</u>

studies/climate-change-and-global-warming/content-section-0

- <u>https://open.umn.edu/opentextbooks/textbooks/environmental-science</u>
- <u>https://www.unep.org/library</u>
- <u>https://www.greengrowthknowledge.org/</u>
- <u>https://www.wri.org/</u>
- <u>https://openknowledge.worldbank.org/</u>

Modes of Evaluation:

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Minor - Data Science

UDT101	Data Analytics using SQL	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co	- Nil				
Requisites					

Course Perspective

This course is designed to provide students with essential skills in SQL, a fundamental tool in data analysis and data science. Students will learn to effectively retrieve, clean, manipulate, and analyze data stored in relational databases, supporting data-driven decision-making in various domains. The course emphasizes practical application, equipping students with the ability to use SQL to solve real-world problems in business, finance, marketing, healthcare, and more. By mastering SQL, students will gain a strong foundation in data analytics, enabling them to make meaningful contributions in their careers.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and constructing complex SQL queries to retrieve, filter, and aggregate data from relational databases.

CO2: Applying SQL commands to clean and preprocess data, including handling missing values, duplicates, and performing data transformations.

CO3: Analyzing datasets using SQL queries to identify patterns and summarize key statistics for initial insights.

CO4: Evaluating and interpreting query results by visualizing data with tools or libraries to create meaningful charts, graphs, and plots that enhance understanding.

Course Content

Unit 1: Introduction to SQL and Database Management Contact Hours: 16

- Introduction to Data Science
- Introduction to SQL Server
- Understanding Data & Information
- Database Concepts
- DBMS and RDBMS
- Database Design Principles
- Types of Databases
- SQL Server Versions
- Creating Databases
- Sub-languages of T-SQL: DDL, DML, TCL, DCL, DQL
- Creating Tables
- Data Manipulation (Insert, Delete, Update)
- Normalization
- Constraints (Unique, Not Null, Primary Key, Check, Default, Foreign Key)

Unit 2: SQL Queries and Data Manipulation

• Working with Single Table Queries

- Writing Queries using SELECT Statement
- Understanding Query Flow
- Operators in SQL Server

Contact Hours: 22

- Clauses in SQL Server (WHERE, ORDER BY, DISTINCT, TOP)
- Filtering and Sorting Data
- DML Commands (Insert, Update, Delete)
- DDL Commands (Create, Alter, Drop, Truncate)
- Delete vs Truncate

Unit 3: SQL Functions and Aggregation

Contact Hours: 10

- Built-in Functions
- Scalar Functions (String, Date, ISNULL, etc.)
- Group Functions (Aggregate Functions: COUNT, MAX, MIN, AVG, SUM)
- Usage of Functions in Data Analysis

Unit 4: Advanced SQL Queries: Subqueries and Joins Contact Hours: 16

- Subqueries: Importance and Types
- Nested Queries

• JOINS: Importance and Types (Inner Join, Outer Joins, Left, Right Outer Joins)

List of Practicals

- Create a student table with student ID, name, and marks as attributes where student ID is the primary key.
- Insert the details of a new student in the above table.
- Delete the details of a student in the above table.
- Use the SELECT command to get the details of students with marks more than 80.
- Find the min, max, sum, and average of marks in a student marks table.
- Find the total number of customers from each country using GROUP BY.
- Write a SQL query to order the (student ID, marks) table in descending order of marks.

- Write a SQL query to display marks without decimal places, the remainder after dividing marks by 3, and the square of marks.
- Write a SQL query to display names in capital letters, small letters, first 3 letters of the name, last 3 letters of the name, and the position of the letter 'A' in the name.
- Remove extra spaces from left, right, and both sides of the text "SQL for Data Science".
- Display today's date in "Date/Month/Year" format.
- Display the day name, month name, day, day name, day of the month, and day of the year for today's date.

Learning Experience

This course will integrate lectures, interactive sessions, and hands-on projects to deepen understanding of SQL, data manipulation, and data analysis.

Instruction Methods:

- Lectures: Core SQL concepts will be taught using multimedia presentations and realworld examples.
- **Interactive Sessions:** Q&A, live coding exercises, and group discussions will actively engage students.
- Technology Use:
- **Online Platforms:** An LMS will host resources, recorded lectures, assignments, and discussion forums to facilitate extended learning.
- Assessments:
- Formative: Regular quizzes, assignments, and online discussions will provide continuous feedback.
- **Summative:** Exams, project presentations, and peer reviews will assess students' mastery of the material.

Support: The course instructor will offer additional guidance, with peer collaboration encouraged through group work and review sessions. Continuous feedback will ensure students' progress and improvement in achieving course outcomes.

Textbooks

- 1. "Learning SQL" by Alan Beaulieu
- 2. "SQL for Dummies" by Allen G. Taylor

Suggested Readings

- 1. "SQL in 10 Minutes, Sams Teach Yourself" by Ben Forta
- 2. "SQL Pocket Guide" by Jonathan Gennick

3. "The Practical SQL Handbook" by Judith S. Bowman, Sandra L. Emerson, and Marcy Darnovsky

Open Educational Resources (OER)

- 1. <u>https://www.w3schools.com/sql/</u>
- 2. <u>https://www.khanacademy.org/computer-programming/new/sql</u>
- 3. <u>https://www.coursera.org/learn/sql-for-data-science</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
An me components to be evenily spaced	

Project/quizzes/assignment and essays/presentation/	1
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

SEMESTER II					
UDT102	Data Analytics using R	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co	- Basic concepts of Statistics				
Requisites					

Course Perspective

The course "Data Analytics using R" is designed to equip students with the foundational skills in R programming necessary for data analysis in various domains. By engaging with this course, students will gain hands-on experience in data manipulation, visualization, and statistical analysis using R, making them proficient in handling real-world data challenges. The knowledge acquired in this course is applicable across industries where data-driven decisionmaking is key. The skills developed will enhance students' academic prowess and prepare them for careers in data science, analytics, and research. For example, students will be able to create insightful visualizations to present data-driven solutions, identify trends, and model data effectively, which are essential skills in today's data-centric job market.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and applying R programming concepts to perform basic data manipulation and visualization tasks.

CO2: Analyzing data sets by employing appropriate R data structures, such as vectors, matrices, and data frames.

CO3: Creating custom R functions and utilizing control structures to automate data analysis processes.

CO4: Evaluating and interpreting data trends through various graphical representations in R.

Course Content

Unit I: Fundamentals of R: (No. of Hours: 16)

- Introduction to R: Features of R, Environment, R Studio
 - Basics of R: Assignment, Modes, Operators, Logical values, Basic Functions
 - R Data Structures: Vectors, Lists, Matrices, Data Frames, Factors
 - Control Structures: if-else, loops, and functions

Unit II: Data Structures in R: (No. of Hours: 16)

- Vectors: Definition, Declaration, Operations
- Matrices: Creating, Reshaping, Operations
- Lists: Creating, General Operations
- Data Frames: Creating, Accessing, Merging, Special Functions

Unit III: Working with Data in R:

(No. of Hours: 16)

- Reading and Writing Data: CSV, Excel, Text Files
- String Operations: Regular Expressions, Dates in R

• Data Preprocessing: Descriptive Statistics, Handling Missing Values, Normalization

• Exploratory Data Analysis: Summarizing Data, Identifying Patterns

Unit IV: Data Visualization with R: (No. of Hours: 16)

- Basic Visualization Tools: Bar Charts, Histograms, Pie Charts, Scatter Plots, Line Plots
- Introduction to ggplot2: Creating Simple Plots, Customization Techniques
- Project on R and related discussion

Learning Experience

This course will be conducted through a blend of lectures, practical sessions, and interactive activities. Students will engage in hands-on learning using R software, working on real-world data sets to apply concepts learned in class. Methods of instruction will include case studies, group work, and individual assignments.

Instruction Methods:

- Lectures: Core R programming concepts will be taught through multimedia presentations and coding examples.
- Hands-on Sessions: Students will work on real-world data sets using R, applying concepts through practical exercises.
- **Group Work and Case Studies:** Collaborative projects and case studies will reinforce learning and promote teamwork.
- Technology Use:
- **R and RStudio:** Students will use R and RStudio for data manipulation, visualization, and analysis.
- Shiny: For creating interactive web applications and visualizations.
- Online Platforms: LMS for accessing resources, recorded lectures, and submitting assignments.

- Assessments:
- Formative: Regular quizzes, coding exercises, and assignments for continuous feedback.
- Summative: Project presentations, case study analyses, and a final assessment to evaluate students' mastery of R programming and data analysis.

Support: The course instructor will provide continuous guidance, with opportunities for students to collaborate through group work and peer reviews. Regular feedback will be given on assignments and projects, and students are encouraged to seek help as needed to enhance their learning experience.

Textbooks

- 1. Cognitive Computing with IBM Watson by Rob High, Tanmay Bakshi (1st edition)
- Nina Zumel, John Mount, Practical Data Science with R, Manning Publications, 2014

Suggested Readings

- Mark Gardener, Beginning R: The Statistical Programming Language, John Wiley & Sons, 2012
- 2. Nathan Yau, Visualize This: The FlowingData Guide to Design, Visualization, and Statistics, Wiley, 2011

Open Educational Resources (OER)

- 1. "Introduction to Data Science with R" (HarvardX Data Science Series on edX)
- 2. "R Programming" (Coursera by Johns Hopkins University)
- 3. "Advanced R" by Hadley Wickham (available online at Advanced R)

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

Students will participate in activities such as data cleaning, summarization, and visualization tasks. They will engage in assignments, quizzes, and group discussions, focusing on applying the concepts learned to real-time data. These activities will reinforce the theoretical knowledge acquired and provide practical experience in data analytics.

SEMESTER III					
UDT103	Python for Data Science	L	T	Р	С
Version		2	0	4	4

Category of Course	Minor
Total Contact Hours	64 Hours
Pre-Requisites/ Co-	Nil
Requisites	

Course Perspective

"Python for Data Science" is designed to equip students with the foundational skills necessary for data analysis and manipulation using Python, a leading programming language in the data science field. The course contributes to students' academic and professional development by providing them with essential tools and techniques to solve real-world data problems. Students will gain knowledge in Python programming, data manipulation using NumPy and Pandas, and data cleaning and visualization techniques, making them well-prepared for careers in data science, analytics, and related fields. The skills learned in this course are directly applicable to analyzing large datasets, performing complex data operations, and generating meaningful insights, which are crucial in various industries such as finance, healthcare, marketing, and technology.

Course Outcomes

Upon completion of this course, the learner will be:

CO1: Understanding and remembering Python's built-in data types and methods to solve basic data-related problems.

CO2: Applying efficient data storage and operations using NumPy arrays for numerical data processing.

CO3: Analyzing data using Pandas for advanced data manipulation tasks, identifying trends and patterns in datasets.

CO4: Evaluating data pre-processing techniques and creating visualizations using Pandas to communicate insights effectively.

Course Content

Unit 1: Introduction to Data Science and Python Programming (No. of Hours: 16)

- Introduction to Data Science
- Why Python?
- Essential Python libraries
- Python Introduction: Features, Identifiers, Reserved words, Indentation, Comments
- Built-in Data types and their Methods: Strings, List, Tuples, Dictionary, Set
- Type Conversion
- Operators, Decision Making, Looping, Loop Control statement
- Math and Random number functions
- User-defined functions: function arguments & its types

Practical Component:

- 1. Implement basic Python programs for reading input from the console.
- 2. Perform operations on Python built-in data types: Strings, List, Tuples, Dictionary, Set.
- 3. Solve problems using decision and looping statements.
- 4. Handle numerical operations using math and random number functions.
- 5. Create user-defined functions with different types of function arguments.

Unit 2: Introduction to NumPy

(No. of Hours: 16)

- Arrays and Vectorized Computation
- The NumPy ndarray
- Creating ndarrays
- Data Types for ndarrays
- Arithmetic with NumPy Arrays

- Basic Indexing and Slicing
- Boolean Indexing
- Transposing Arrays and Swapping Axes
- Universal Functions: Fast Element-Wise Array Functions
- Mathematical and Statistical Methods
- Sorting, Unique and Other Set Logic

Practical Component:

- 1. Create NumPy arrays from Python Data Structures and Random Functions.
- 2. Manipulate NumPy arrays: Indexing, Slicing, Reshaping, Joining, and Splitting.
- 3. Perform computations using Universal Functions and Mathematical methods.
- 4. Import and analyze data from CSV files using NumPy.
- 5. Manipulate images using NumPy.

Unit 3: Data Manipulation with Pandas

(No. of Hours: 16)

- Introduction to Pandas Data Structures: Series, DataFrame
- Essential Functionality: Dropping Entries, Indexing, Selection, and Filtering
- Function Application and Mapping
- Sorting and Ranking
- Summarizing and Computing Descriptive Statistics
- Unique Values, Value Counts, and Membership
- Reading and Writing Data in Text Format

Practical Component:

- 1. Create Pandas Series and DataFrames from various inputs.
- 2. Perform data operations on CSV files using Pandas.
- 3. Conduct statistical analysis and operations on DataFrames.
- 4. Handle categorical data using Pandas.
- 5. Rename columns and restructure data using Pandas.

Unit 4: Data Cleaning, Preparation, and Visualization

(No. of Hours: 16)

- Handling Missing Data
- Data Transformation: Removing Duplicates, Transforming Data Using a Function or Mapping, Replacing Values
- Detecting and Filtering Outliers
- String Manipulation: Vectorized String Functions in Pandas
- Plotting with Pandas: Line Plots, Bar Plots, Histograms, Density Plots, Scatter Plots

Practical Component:

- 1. Handle missing data and perform data transformations using Pandas.
- 2. Detect and filter outliers in datasets.
- 3. Execute vectorized string operations in Pandas.
- 4. Visualize data using various plotting techniques.

Learning Experience

This course will combine lectures, hands-on sessions, and interactive activities to equip students with Python programming skills for data science.

Instruction Methods:

- Lectures: Core Python programming concepts and data science principles will be introduced through multimedia presentations and live coding demonstrations.
- Hands-on Sessions: Students will work on real-world data sets using Python, applying concepts learned in class through practical exercises.
- **Group Work and Case Studies:** Collaborative projects will reinforce learning, with case studies to address real-world data science challenges.
- Technology Use:
- **Python, NumPy, Pandas:** These tools will be used for data analysis, manipulation, and visualization.
- Jupyter Notebooks: For executing and documenting Python code.

- Online Platforms: LMS for accessing resources, recorded lectures, and submitting assignments.
- Assessments:
- Formative: Regular quizzes, coding exercises, and assignments will provide continuous feedback on students' progress.
- **Summative:** Project presentations, case study analyses, and a final assessment will evaluate students' understanding and application of Python for data science.

Support: The course instructor will offer continuous guidance and feedback. Peer collaboration will be encouraged through group work and review sessions. Students will have access to online resources and office hours to seek additional help when needed. Regular feedback will ensure that students meet the course outcomes effectively.

Textbooks

- 1. Y. Daniel Liang, "Introduction to Programming using Python," Pearson, 2012.
- Wes McKinney, "Python for Data Analysis: Data Wrangling with Pandas, NumPy, and IPython," O'Reilly, 2nd Edition, 2018.
- 3. Jake VanderPlas, "Python Data Science Handbook: Essential Tools for Working with Data," O'Reilly, 2017.

4.

Suggested Readings

- 1. Wesley J. Chun, "Core Python Programming," Prentice Hall, 2006.
- 2. Mark Lutz, "Learning Python," O'Reilly, 4th Edition, 2009.
- Joel Grus, "Data Science from Scratch: First Principles with Python," O'Reilly, 2015.

Open Educational Resources (OER)

- 1. NPTEL Python for Data Science
- 2. Kaggle's Python for Data Science
- 3. Awesome Python for Data Science (GitHub)

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

UDT104	Data	Pre-processing	andL	Т	Р	С
	Visualiz	ation using Python				
Version			2	0	4	4
Version Category of Course	Minor		2	0	4	

Total Contact Hours	64 Hours
Pre-Requisites/ Co-	Basics of Python Programming
Requisites	

Course Perspective

This course is integral to the data science curriculum as it provides students with the foundational skills necessary for effective data analysis and visualization. Mastery of data preprocessing ensures that students can clean and prepare datasets, which is crucial for generating accurate and reliable insights in any data-driven field. Visualization techniques taught in this course empower students to communicate their findings effectively, making complex data understandable to a wide audience. The course emphasizes real-world applicability, allowing students to work with diverse datasets and leverage popular Python libraries to create visualizations that are both informative and aesthetically pleasing. The skills and knowledge gained from this course are essential for careers in data analysis, business intelligence, and any profession requiring data-driven decision-making.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Explaining the significance of data pre-processing in the data analysis pipeline and its role in enhancing the quality of data for machine learning models.

CO2: Identifying and applying appropriate techniques for handling missing data, duplicates, and outliers to ensure data integrity.

CO3: Implementing data transformation processes such as normalization, scaling, and encoding to prepare datasets for analysis.

CO4: Analyzing datasets using exploratory data analysis (EDA) techniques and creating visualizations to uncover patterns, correlations, and trends.

Course Content

Unit 1: Introduction to Data Preprocessing Understanding the importance of data preprocessing Steps involved in data preprocessing

• Handling missing data and outliers

Unit 2: Data Cleaning and Transformation No. of Hours: 16

- Removing duplicates and dealing with data inconsistencies
- Data normalization, scaling, and encoding techniques
- Handling categorical variables

Unit 3: Exploratory Data Analysis (EDA)

- Data summarization and descriptive statistics
- Data visualization techniques: histograms, box plots, scatter plots

No. of Hours: 16

• Correlation analysis, heatmaps, and pair plots

Unit 4: Data Visualization Libraries and Applications No. of Hours: 16

- Introduction to Python libraries: Matplotlib, Seaborn, Plotly
- Creating and customizing plots
- Interactive visualizations and real-world data applications through Project.

Learning Experience

This course will blend lectures, practical sessions, and interactive activities to develop skills in data preprocessing and visualization using Python.

Instruction Methods:

- Lectures: Key concepts of data preprocessing and visualization will be taught through multimedia presentations and theoretical explanations.
- Hands-on Sessions: Practical exercises using Python libraries will allow students to apply data cleaning, transformation, and visualization techniques.

- **Group Work and Projects:** Collaborative projects and case studies will provide realworld data analysis experience and encourage teamwork.
- Technology Use:
- **Python Libraries:** NumPy, Pandas, Matplotlib, Seaborn, Plotly for data manipulation and visualization.
- Jupyter Notebooks: For coding exercises and project documentation.
- **Online Platforms:** LMS for accessing course materials, recorded lectures, and submitting assignments.
- Assessments:
- **Formative:** Regular quizzes, coding exercises, and practical assignments for ongoing feedback.

• **Summative:** Project presentations, case study analyses, and a final assessment to evaluate mastery of data preprocessing and visualization techniques.

Support: The course instructor will provide continuous feedback and be available during office hours. Peer collaboration and group activities will be encouraged to enhance learning. Students will have access to online resources and additional help as needed to achieve course outcomes effectively.

Textbooks

- 1. Claus Wilke, "Fundamentals of Data Visualization: A Primer on Making Informative and Compelling Figures", 1st edition, O'Reilly Media Inc, 2019.
- 2. Jacqueline Kazil, Katharine Jarmul, "Data Wrangling with Python," O'Reilly Media.

Suggested Readings

- 1. **Python Data Science Handbook** by Jake VanderPlas.
- 2. Effective Data Visualization by Stephanie D. H. Evergreen.
- 3. **Practical Statistics for Data Scientists** by Peter Bruce and Andrew Bruce.

Open Educational Resources (OER)

- 1. <u>Data Visualization Netquest eBook</u>
- 2. <u>Coursera: Data Visualization</u>
- 3. <u>Coursera: Python for Data Visualization</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
Include assignments, lab work, quizzes, and a final project, focusing on the practical application of data preprocessing and visualization techniques.	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER V					
UDT105	Time series analysis and forecasting using Python	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor	•	•	•	•

Total Contact Hours	60 Hours
Pre-Requisites/ Co-	Nil
Requisites	

Course Perspective

This course is designed to equip students with essential skills in Time Series Analysis and Forecasting, crucial for making accurate predictions based on temporal data. Students will learn to analyze time series data, apply various forecasting models, and evaluate their performance to support decision-making across diverse fields. The course covers foundational concepts, including autocorrelation, statistical inference in regression models, and advanced techniques such as ARIMA and seasonal ARIMA models. By integrating theoretical knowledge with practical application, students will be able to address real-world forecasting challenges, enhancing their analytical capabilities and making informed contributions in areas such as finance, economics, and business strategy.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding the fundamental concepts of time series data, including key patterns, trends, and seasonality, as well as various statistical methods used for time series analysis.

CO2: Applying forecasting models such as regression models and ARIMA to predict future values based on historical time series data in practical scenarios.

CO3: Analyzing and interpreting time series data through graphical displays, numerical descriptions, and techniques such as smoothing, transformations, and adjustments for enhanced analysis.

CO4: Evaluating the accuracy and performance of different forecasting models using statistical techniques, ensuring reliable predictions and continuous model monitoring.

Course Content

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Unit 1: Introduction of Time series Analysis Statistical Methods: No. of

- Introduction to Time Series and Forecasting
- Different types of data
- Internal structures of time series
- Models for time series analysis
- Autocorrelation and Partial autocorrelation
- Examples of Time series Nature and uses of forecasting
- Forecasting Process
- Data for forecasting
- Resources for forecasting

Unit 2: Statistics Background for Forecasting:

- Graphical Displays
- Time Series Plots
- Plotting Smoothed Data
- Numerical Description of Time Series Data
- Use of Data Transformations and Adjustments
- General Approach to Time Series Modeling and Forecasting
- Evaluating and Monitoring Forecasting Model Performance

Unit 3: Time Series Regression Model:

- Introduction Least Squares Estimation in Linear Regression Models
- Statistical Inference in Linear Regression
- Prediction of New Observations
- Model Adequacy Checking
- Variable Selection Methods in Regression
- Generalized and Weighted Least Squares
- Regression Models for General Time Series Data
- Exponential Smoothing, First order and Second order.

Unit 4: Autoregressive Integrated moving average (ARIMA) Models: No. of Hours: 15

- Autoregressive Moving Average (ARIMA) Models
- Stationarity and Invertibility of ARIMA Models

No. of Hours: 15

No. of Hours: 15

No. of Hours: 15

- Checking for Stationarity using Variogram
- Detecting Nonstationarity
- Autoregressive Integrated Moving Average (ARIMA) Models
- Forecasting using ARIMA
- Seasonal Data
- Seasonal ARIMA Models Forecasting using Seasonal ARIMA Models Introduction
- Finding the "BEST" Model
- Example: Internet Users Data Model Selection Criteria
- Impulse Response Function to Study the Differences in Models Comparing Impulse Response Functions for Competing Models

List of Practicals

- **Time Series Data Cleaning:** Apply techniques to clean time series data, including handling missing values and outliers.
- Loading and Handling Time Series Data: Import time series data from various sources and manage it using appropriate tools.
- **Preprocessing Techniques:** Implement preprocessing techniques such as normalization and transformation to prepare time series data for analysis.
- How to Check Stationarity: Use statistical tests (e.g., ADF test) to determine if a time series is stationary.
- Making a Time Series Stationary: Apply techniques such as differencing and transformation to achieve stationarity.
- Estimating & Eliminating Trend: Use aggregation, smoothing, and polynomial fitting to estimate and remove trends.
- Eliminating Seasonality: Apply decomposition methods to separate and remove seasonal effects from the time series.
- **Moving Average Time Analysis:** Apply moving average techniques to smooth time series data and identify patterns.
- **Smoothing Time Series Data:** Use various smoothing methods (e.g., simple, weighted) to reduce noise and highlight trends.

- Checking Linear and Non-Linear Trends: Analyze time series data to identify and model both linear and non-linear trends.
- Creating a Time Series Model: Develop a time series model based on observed patterns and trends.
- Moving Average Model: Implement and evaluate a moving average model to forecast time series data.
- **Exponential Smoothing:** Apply exponential smoothing methods to forecast time series data and assess model performance.
- **ARIMA Model:** Develop and validate an ARIMA model for time series forecasting.
- Seasonal ARIMA Model (ARIMA): Create and test a SARIMA model to account for seasonality in time series data.

Learning Experience

This course will blend lectures, interactive sessions, and hands-on projects to deepen understanding and application of time series analysis and forecasting techniques.

Instruction Methods:

- Lectures: Core time series analysis and forecasting concepts will be taught using multimedia presentations and real-world case studies.
- **Interactive Sessions:** Q&A, practical exercises, and group discussions will actively engage students in applying time series models and forecasting techniques.
- Technology Use:
- **Python:** Primary tool for statistical analysis and data visualization.
- Online Platforms: LMS for accessing resources, recorded lectures, and discussion forums.
- Assessments:
- Formative: Regular quizzes, practical exercises, and assignments for continuous feedback.
- **Summative:** Case study analyses, project presentations, and a final exam to evaluate students' grasp of statistical methods.

Support: The course instructor will be available for guidance during office hours, and students are encouraged to collaborate through peer reviews and group work. Regular feedback will be provided to help students refine their skills and meet course outcomes effectively.

Textbooks

- Introduction To Time Series Analysis And Forecasting, 2nd Edition, Wiley Series In Probability And Statistics, By Douglas C. Montgomery, Cheryl L. Jen(2015)
- 2. Master Time Series Data Processing, Visualization, And Modeling Using Python Dr. Avishek Pal Dr. Pks Prakash (2017)

Suggested Readings

- 1. Time Series Analysis and Its Applications: With R Examples by Robert H. Shumway and David S. Stoffer.
- 2. Applied Time Series Analysis by Wayne A. Woodward, Henry L. Gray, and Alan C. Elliott.
- 3. The Elements of Statistical Learning: Data Mining, Inference, and Prediction by Trevor Hastie, Robert Tibshirani, and Jerome Friedman.

Open Educational Resources (OER)

- 1. Khan Academy Time Series Analysis
- 2. <u>Time Series Regression Analysis University of California, Irvine</u>
- 3. <u>Time Series Analysis and Forecasting Coursera</u>

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Test/Project/quizzes/assignment and essays/presentation/	
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

UDT106	Fundamental of Machine	L	Т	Р	С
	Learning				
Version		2	0	4	4
Category of Course	Minor		1	I	1
Total Contact Hours	64 Hours				
Pre-Requisites/ Co-	Basic concepts of Statistics				
Requisites					

Course Perspective

This course provides essential skills in machine learning, covering both supervised and unsupervised techniques using Python. Students will learn to preprocess data, apply regression and classification models, and utilize clustering methods. The course emphasizes practical application, preparing students to tackle real-world problems and make impactful contributions in fields like business, healthcare, and technology.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Remembering and understanding machine learning concepts to identify suitable applications, distinguishing between supervised and unsupervised learning, and utilizing appropriate Python libraries for machine learning tasks.

CO2: Applying preprocessing techniques to prepare data for analysis by handling missing values, encoding categorical data, and applying normalization, standardization, and Principal Component Analysis (PCA) techniques.

CO3: Analyzing and developing supervised learning models, including linear and nonlinear regression, K-Nearest Neighbor, Decision Trees, Logistic Regression, and Support Vector Machines, to address various predictive and classification problems.

CO4: Evaluating the performance of machine learning models using appropriate evaluation methods to ensure the development of accurate and effective predictive models.

Course Content

Unit I: Introduction to Machine Learning No. of Hours: 16 Hours

- Application of Machine Learning
- Supervised vs Unsupervised Learning
- Python libraries suitable for Machine Learning

Unit II: Data Pre-Processing and Data No. of Hours: 16 Hours

- Identifying and handling the missing values
- Encoding the categorical data
- Normalization
- Standardization

• PCA

Unit III: Supervised Learning Regression and Classification No. of Hours: 16 Hours

- Linear Regression
- Non-Linear Regression
- Model evaluation methods
- K-Nearest Neighbour
- Decision Tree
- Logistic Regression
- Support Vector Machines,
- Model Evaluation

Unit IV: Unsupervised Learning

No. of Hours: 16 Hours

- K-means Clustering
- Hierarchical Clustering
- Density-Based Clustering

Learning Experience

This course will combine lectures, interactive sessions, and hands-on projects to enhance understanding of machine learning concepts, data preprocessing, and model implementation. Students will engage in practical exercises to apply supervised and unsupervised learning techniques, ensuring comprehensive learning experience.

Lectures:

Instruction Methods:

• Lectures: Core machine learning concepts will be taught using multimedia presentations and real-world case studies.

- **Interactive Sessions:** Q&A, coding exercises, and group discussions will actively engage students in applying machine learning techniques and solving practical problems.
- **Group Work and Case Studies:** Collaborative projects and case studies will reinforce learning and promote teamwork.
- Technology Use:
- **R and RStudio:** Students will use R and RStudio for data manipulation, visualization, and analysis.
- Shiny: For creating interactive web applications and visualizations.
- **Online Platforms:** LMS for accessing resources, recorded lectures, and submitting assignments.
- Assessments:
- Formative: Regular quizzes, coding exercises, and assignments for continuous feedback.
- **Summative:** Project presentations, case study analyses, and a final assessment to evaluate students' mastery of R programming and data analysis.

Support: The course instructor will provide continuous guidance, with opportunities for students to collaborate through group work and peer reviews. Regular feedback will be given on assignments and projects, and students are encouraged to seek help as needed to enhance their learning experience.

Textbooks

- 1. Machine Learning Tom M. Mitchell
- 2. Python Machine Learning Sebastian, Raschka and Vahid Mirjalili

Suggested Readings

- 1. Understanding Machine Learning Shai Shalev-Shwartz and Shai Ben-David La
- 2. Hands-On Machine Learning with Scikit-Learn and TensorFlow: Concepts, Tools, and Technique to Build Intelligent Systems-AurelienGeron

Open Educational Resources (OER)

https://www.coursera.org/learn/machine-learning

https://www.datacamp.com/tutorial/introduction-machine-learning-python

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	/
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

Students will participate in activities such as data cleaning, summarization, and visualization tasks. They will engage in assignments, quizzes, and group discussions, focusing on applying the concepts learned to real-time data. These activities will reinforce the theoretical knowledge acquired and provide practical experience in data analytics.

UDT107	Data DrivenL	Т	Р	С	
001107	Applications	I	r	C	
Version	2	0	4	4	
Category	of Minor VII (Practical)			
Course					
Total	60 Hours				
Contact					
Hours					
Pre-					
Requisites/					
Co-					
Requisites					

Course Perspective

The undergraduate course "Data Driven Applications" focuses on utilizing Power BI to design and manage data-driven reports and visualizations. It starts with an introduction to Power BI's architecture, installation, and cloud capabilities, covering essentials such as Power BI Desktop, mobile editions, and report rendering options. Students will learn to create and design interactive reports using various data sources and visualization tools, exploring report design elements, auto filters, and multiple visualization types. The course also delves into advanced features, including custom visualizations, real-time data access, and comprehensive report formatting and analytics. Overall, it provides a solid foundation in leveraging Power BI for effective data analysis and business intelligence.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Observing and identifying the fundamental components and features of Power BI, architecture, installation procedures, and basic functionalities.

CO2: Imitating best practices in report design by replicating sample reports and visualizations using Power BI's tools and effective data representation.

CO3: Practicing and creating interactive Power BI reports by utilizing various data sources, applying filters, and experimenting with different visualization tools to effectively communicate insights.

Course Content:

Unit 1 : Introduction to Power Bi

Contact Hours: 15

Introduction to Power BI - Need, Importance, Power BI - Advantages and Scalable Options, History - Power View, Power Query, Power Pivot, Power BI Data Source Library and DW Files, Cloud Collaboration and Usage Scope, Business Analyst Tools, MS Cloud Tools, Power BI Installation and Cloud Account, Power BI Cloud and Power BI Service, Power BI Architecture and Data Access, On-premise Data Access and Microsoft On Drive, Power BI Desktop - Installation, Usage, Sample Reports and Visualization Controls, Power BI Cloud Account Configuration, Understanding Desktop & Mobile Editions, Report Rendering Options and End User Access, Power View and Power Map. Power BI Licenses, Course Plan - Power BI Online Training.

Unit 2 : Creating Power Bi Reports, Auto FiltersContact Hours: 15

Report Design with Legacy & .DAT Files, Report Design with Database Tables, Understanding Power BI Report Designer, Report Canvas, Report Pages: Creation, Renames, Report Visuals, Fields and UI Options, Experimenting Visual Interactions, Advantages, Reports with Multiple Pages and Advantages, Pages with Multiple Visualizations. Data Access, PUBLISH Options and Report Verification in Cloud, "GET DATA" Options and Report Fields, Filters, Report View Options: Full, Fit Page, Width Scale, Report Design using Databases & Queries, Query Settings and Data Preloads, Navigation Options and Report Refresh, stacked bar chart, Stacked column chart, Clustered bar chart, Clustered column chart, Adding Report Titles. Report Format Options, Focus Mode, Explore and Export Settings.

Unit 3 : Report Visualizations and Properties

Contact Hours: 15

Power BI Design: Canvas, Visualizations and Fields, Import Data Options with Power BI Model, Advantages, Direct Query Options and Real-time (LIVE) Data Access, Data Fields and Filters with Visualizations, Visualization Filters, Page Filters, Report Filters, Conditional Filters and Clearing. Testing Sets, Creating Customised Tables with Power BI Editor, General Properties, Sizing, Dimensions, and Positions, Alternate Text and Tiles. Header (Column, Row) Properties, Grid Properties (Vertical, Horizontal) and Styles, Table Styles & Alternate Row Colours - Static, Dynamic, Sparse, Flashy Rows, Condensed Table Reports. Focus Mode, Totals Computations, Background. Bordes Properties, Column Headers, Column Formatting, Value Properties, Conditional Formatting Options - Colour Scale, Page Level Filters and Report Level Filters, Visual-Level Filters and Format Options, Report Fields, Formats and Analytics, Page-Level Filters and Column Formatting, Filters, Background Properties, Borders and Lock Aspect.

Unit 4: Chart and Map Report Properties

Contact Hours: 15

Chart report types and properties, stacked bar chart, stacked column chart clustered bar chart, clustered column chart, 100% stacked bar chart, 100% stacked column chart, line charts, area charts, stacked area charts, line and stacked row charts, line and stacked column charts, waterfall chart, scatter chart, pie chart, Field Properties: Axis, Legend, Value, Tooltip, Field Properties: Colour Saturation, Filters Types, Formats: Legend, Axis, Data Labels, Plot Area, Data Labels: Visibility, Colour and Display Units, Data Labels: Precision, Position, Text Options, Analytics: Constant Line, Position, Labels, Working with Waterfall Charts and Default Values, Modifying Legends and Visual Filters - Options, Map Reports: Working with Map Reports.

Learning Experience

In the "Data Driven Applications" course, students will actively engage through hands-on activities and real-world case studies using Power BI. The course will blend lectures with practical exercises on both the Power BI desktop and cloud platforms. Students will work on assignments and group projects, creating and presenting their own reports and visualizations.

They'll receive feedback and support from the course instructor and will have opportunities to collaborate with classmates. The goal is to apply what they learn in a practical way, with plenty of chances for peer interaction and guidance throughout the course.

Instruction Methods:

- Lectures: Core MATLAB/Mathematica concepts will be taught using multimedia presentations and real-world examples.
- Interactive Sessions: Q&A, live coding exercises, and group discussions will actively engage students.
- Technology Use:
- **Online Platforms:** An LMS will host resources, recorded lectures, assignments, and discussion forums to facilitate extended learning.
- Assessments:
- Formative: Regular quizzes, assignments, and online discussions will provide continuous feedback.
- **Summative:** Exams, project presentations, and peer reviews will assess students' mastery of the material.

Support: The course instructor will offer additional guidance, with peer collaboration encouraged through group work and review sessions. Continuous feedback will ensure students' progress and improvement in achieving course outcomes.

Textbooks

- "Beginning Power BI: A Practical Guide to Self-Service Data Analytics with Excel 2016 and Power BI Desktop" by Dan Clark
- 2. "Power BI Step-by-Step Part 1: Up and Running: Power BI Mastery through hands-on Tutorials (Power BI Step by Step)" by Grant Gamble
- 3. "Mastering Microsoft Power BI" by Brett Powell

Reference Books for Power BI

- 1. The Definitive Guide to DAX by Marco Russo and Alberto Ferrari
- 2. Microsoft Power BI Cookbook by Greg Deckler

3. Analyzing Data with Power BI and Power Pivot for Excel by Alberto Ferrari and Marco Russo

Open Educational Resources (OER)

- 1. https://learn.microsoft.com/en-us/power-bi/
- 2. <u>https://docs.microsoft.com/en-us/power-bi/guided-learning/</u>
- 3. <u>https://docs.microsoft.com/en-us/learn/paths/analyze-visualize-data-power-</u>

bi/

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	/
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER VII					
UDT108	Project and Case Study	L	Т	Р	С

Version	2 0 4 4
Category of Course	Minor
Total Contact Hours	64 Hours
Pre-Requisites/ Co	- Python Programming and ML techniques
Requisites	

Course Outcomes (COs)

Upon completion of the course, the learner will be:

CO1: Identifying and articulating a research problem relevant to data analysis, demonstrating an understanding of domain-specific challenges and opportunities.

CO2: Designing and executing a comprehensive data analysis project, applying appropriate methodologies, tools, and techniques learned throughout the course.

CO3: Analyzing and interpreting the results of data-driven investigations, critically evaluating the effectiveness and limitations of the chosen methods.

CO4: Evaluating findings through written reports and oral presentations, effectively communicating the significance, methodology, and outcomes of the project to both technical and non-technical audiences.

CO5: Collaborating effectively within a team environment to complete the project, contributing to group discussions, planning, and decision-making processes.

Learning Experience

This course will provide a hands-on approach to applying data analysis techniques through a project-based framework.

Instruction Methods:

• Workshops: The course will involve interactive sessions where students can discuss project ideas, share progress, and seek guidance from the instructor.

- Hands-on Project Work: Students will work on real-world projects, applying the skills and knowledge they have acquired in previous courses to solve practical problems.
- **Case Study Discussions**: Students will analyze relevant case studies to understand the application of data analysis techniques in different domains, encouraging critical thinking and problem-solving skills.
- Technology Use:
- Data Analysis Tools: Students will utilize tools such as Python, R, or SQL, depending on their project requirements.
- **Project Management Tools**: Platforms like Trello or Asana may be used for team collaboration and project tracking.
- **Presentation Tools**: Tools such as PowerPoint or Tableau for creating visual presentations of project findings.
- Assessments:
- **Formative**: Regular progress updates, peer reviews, and feedback sessions will be conducted to guide students throughout the project lifecycle.
- **Summative**: The final assessment will include a comprehensive project report and an oral presentation, evaluating the application of data analysis techniques and the effectiveness of communication.

Support:

The course instructor will provide continuous support through regular consultations and feedback. Peer collaboration will be encouraged to enhance the learning experience. Students will have access to online resources and office hours for additional help as needed to achieve the course outcomes effectively.

Open Educational Resources (OER)

1. **Kaggle Datasets and Competitions**: For practical project ideas and data sources.

- 2. **Coursera**: Courses on Project Management and Data Analysis.
- 3. **GitHub**: For exploring open-source projects and datasets.

Evaluation Scheme

valuation	aluation components		
1.	Project Proposal: 10% a. Initial identification of the problem, objectives, and methodology.	10 Marks	
2.	Mid-Term Presentation: 20% Progress report including initial findings, challenges, and adjustments.	20 Marks	
3.	Final Report : 30% Comprehensive documentation of the project including literature review, methodology, data analysis, results, and conclusions.		
4.	Final Presentation : 20% Oral presentation and defense of the project findings before peers and evaluators.	20 Marks	
5.	Peer Review and Team Contribution : 10% Assessment based on peer evaluations, participation in team activities, and overall contribution to the project.	10 Marks	
6.	Case Study Analysis: 10% Analysis and presentation of a case study relevant to the project topic, demonstrating the application of theoretical concepts in real-world scenarios.		

MINOR IN AI\ML

UDT101	Data Analytics using SQL	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co	p- Nil				
Requisites					

Course Perspective

This course is designed to provide students with essential skills in SQL, a fundamental tool in data analysis and data science. Students will learn to effectively retrieve, clean, manipulate, and analyze data stored in relational databases, supporting data-driven decision-making in various domains. The course emphasizes practical application, equipping students with the ability to use SQL to solve real-world problems in business, finance, marketing, healthcare, and more. By mastering SQL, students will gain a strong foundation in data analytics, enabling them to make meaningful contributions in their careers.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and constructing complex SQL queries to retrieve, filter, and aggregate data from relational databases.

CO2: Applying SQL commands to clean and preprocess data, including handling missing values, duplicates, and performing data transformations.

CO3: Analyzing datasets using SQL queries to identify patterns and summarize key statistics for initial insights.

CO4: Evaluating and interpreting query results by visualizing data with tools or libraries to create meaningful charts, graphs, and plots that enhance understanding.

Course Content

Unit 1: Introduction to SQL and Database Management

Contact Hours: 16

- Introduction to Data Science
- Introduction to SQL Server
- Understanding Data & Information
- Database Concepts
- DBMS and RDBMS
- Database Design Principles
- Types of Databases
- SQL Server Versions
- Creating Databases
- Sub-languages of T-SQL: DDL, DML, TCL, DCL, DQL
- Creating Tables
- Data Manipulation (Insert, Delete, Update)
- Normalization
- Constraints (Unique, Not Null, Primary Key, Check, Default, Foreign Key)

Unit 2: SQL Queries and Data Manipulation

- Working with Single Table Queries
- Writing Queries using SELECT Statement
- Understanding Query Flow
- Operators in SQL Server
- Clauses in SQL Server (WHERE, ORDER BY, DISTINCT, TOP)
- Filtering and Sorting Data
- DML Commands (Insert, Update, Delete)

Contact Hours: 22

- DDL Commands (Create, Alter, Drop, Truncate)
- Delete vs Truncate

Unit 3: SQL Functions and Aggregation Contact Hours: 10

- Built-in Functions
- Scalar Functions (String, Date, ISNULL, etc.)
- Group Functions (Aggregate Functions: COUNT, MAX, MIN, AVG, SUM)
- Usage of Functions in Data Analysis

Unit 4: Advanced SQL Queries: Subqueries and Joins Contact Hours: 16

- Subqueries: Importance and Types
- Nested Queries
- JOINS: Importance and Types (Inner Join, Outer Joins, Left, Right Outer

Joins)

List of Practicals

- Create a student table with student ID, name, and marks as attributes where student ID is the primary key.
- Insert the details of a new student in the above table.
- Delete the details of a student in the above table.
- Use the SELECT command to get the details of students with marks more than 80.
- Find the min, max, sum, and average of marks in a student marks table.
- Find the total number of customers from each country using GROUP BY.
- Write a SQL query to order the (student ID, marks) table in descending order of marks.
- Write a SQL query to display marks without decimal places, the remainder after dividing marks by 3, and the square of marks.

- Write a SQL query to display names in capital letters, small letters, first 3 letters of the name, last 3 letters of the name, and the position of the letter 'A' in the name.
- Remove extra spaces from left, right, and both sides of the text "SQL for Data Science".
- Display today's date in "Date/Month/Year" format.
- Display the day name, month name, day, day name, day of the month, and day of the year for today's date.

Learning Experience

This course will integrate lectures, interactive sessions, and hands-on projects to deepen understanding of SQL, data manipulation, and data analysis.

Instruction Methods:

- Lectures: Core SQL concepts will be taught using multimedia presentations and realworld examples.
- **Interactive Sessions:** Q&A, live coding exercises, and group discussions will actively engage students.
- Technology Use:
- **Online Platforms:** An LMS will host resources, recorded lectures, assignments, and discussion forums to facilitate extended learning.
- Assessments:
- Formative: Regular quizzes, assignments, and online discussions will provide continuous feedback.
- **Summative:** Exams, project presentations, and peer reviews will assess students' mastery of the material.

Support: The course instructor will offer additional guidance, with peer collaboration encouraged through group work and review sessions. Continuous feedback will ensure students' progress and improvement in achieving course outcomes.

Textbooks

- 2. "Learning SQL" by Alan Beaulieu
- 3. "SQL for Dummies" by Allen G. Taylor

Suggested Readings

- 2. "SQL in 10 Minutes, Sams Teach Yourself" by Ben Forta
- 3. "SQL Pocket Guide" by Jonathan Gennick
- 4. "The Practical SQL Handbook" by Judith S. Bowman, Sandra L. Emerson, and Marcy Darnovsky

Open Educational Resources (OER)

- 2. <u>https://www.w3schools.com/sql/</u>
- 3. <u>https://www.khanacademy.org/computer-programming/new/sql</u>
- 4. <u>https://www.coursera.org/learn/sql-for-data-science</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	,
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade.

UDT102	Data Analytics using R	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co-	Basic concepts of Statistics				
Requisites					

Course Perspective

The course "Data Analytics using R" is designed to equip students with the foundational skills in R programming necessary for data analysis in various domains. By engaging with this course, students will gain hands-on experience in data manipulation, visualization, and statistical analysis using R, making them proficient in handling real-world data challenges. The knowledge acquired in this course is applicable across industries where data-driven decisionmaking is key. The skills developed will enhance students' academic prowess and prepare them for careers in data science, analytics, and research. For example, students will be able to create insightful visualizations to present data-driven solutions, identify trends, and model data effectively, which are essential skills in today's data-centric job market.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and applying R programming concepts to perform basic data manipulation and visualization tasks.

CO2: Analyzing data sets by employing appropriate R data structures, such as vectors, matrices, and data frames.

CO3: Creating custom R functions and utilizing control structures to automate data analysis processes.

CO4: Evaluating and interpreting data trends through various graphical representations in R.

Course Content

Unit I: Fundamentals of R: (No. of Hours: 16)

- Introduction to R: Features of R, Environment, R Studio
 - Basics of R: Assignment, Modes, Operators, Logical values, Basic Functions
 - R Data Structures: Vectors, Lists, Matrices, Data Frames, Factors
 - Control Structures: if-else, loops, and functions

Unit II: Data Structures in R: (No. of Hours: 16)

- Vectors: Definition, Declaration, Operations
- Matrices: Creating, Reshaping, Operations
- Lists: Creating, General Operations
- Data Frames: Creating, Accessing, Merging, Special Functions

Unit III: Working with Data in R: (No. of Hours: 16)

- Reading and Writing Data: CSV, Excel, Text Files
- String Operations: Regular Expressions, Dates in R
- Data Preprocessing: Descriptive Statistics, Handling Missing Values, Normalization
 - Exploratory Data Analysis: Summarizing Data, Identifying Patterns

Unit IV: Data Visualization with R: (No. of Hours: 16)

- Basic Visualization Tools: Bar Charts, Histograms, Pie Charts, Scatter Plots, Line Plots
- Introduction to ggplot2: Creating Simple Plots, Customization Techniques
- Project on R and related discussion

Learning Experience

This course will be conducted through a blend of lectures, practical sessions, and interactive activities. Students will engage in hands-on learning using R software, working on real-world data sets to apply concepts learned in class. Methods of instruction will include case studies, group work, and individual assignments.

Instruction Methods:

- Lectures: Core R programming concepts will be taught through multimedia presentations and coding examples.
- Hands-on Sessions: Students will work on real-world data sets using R, applying concepts through practical exercises.
- **Group Work and Case Studies:** Collaborative projects and case studies will reinforce learning and promote teamwork.
- Technology Use:
- **R and RStudio:** Students will use R and RStudio for data manipulation, visualization, and analysis.
- Shiny: For creating interactive web applications and visualizations.
- Online Platforms: LMS for accessing resources, recorded lectures, and submitting assignments.
- Assessments:
- Formative: Regular quizzes, coding exercises, and assignments for continuous feedback.
- **Summative:** Project presentations, case study analyses, and a final assessment to evaluate students' mastery of R programming and data analysis.

Support: The course instructor will provide continuous guidance, with opportunities for students to collaborate through group work and peer reviews. Regular feedback will be given

on assignments and projects, and students are encouraged to seek help as needed to enhance their learning experience.

Textbooks

- 3. Cognitive Computing with IBM Watson by Rob High, Tanmay Bakshi (1st edition)
- Nina Zumel, John Mount, Practical Data Science with R, Manning Publications, 2014

Suggested Readings

- Mark Gardener, Beginning R: The Statistical Programming Language, John Wiley & Sons, 2012
- 4. Nathan Yau, Visualize This: The FlowingData Guide to Design, Visualization, and Statistics, Wiley, 2011

Open Educational Resources (OER)

- 4. "Introduction to Data Science with R" (HarvardX Data Science Series on edX)
- 5. "R Programming" (Coursera by Johns Hopkins University)
- 6. "Advanced R" by Hadley Wickham (available online at Advanced R)

Evaluation Scheme

Evaluation components	Weightage

Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	/
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

Students will participate in activities such as data cleaning, summarization, and visualization tasks. They will engage in assignments, quizzes, and group discussions, focusing on applying the concepts learned to real-time data. These activities will reinforce the theoretical knowledge acquired and provide practical experience in data analytics.

UDT103	Python for Data Science	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor	I			
Total Contact Hours	64 Hours				
Pre-Requisites/ C	o- Nil				
Requisites					

Course Perspective

"Python for Data Science" is designed to equip students with the foundational skills necessary for data analysis and manipulation using Python, a leading programming language in the data science field. The course contributes to students' academic and professional development by providing them with essential tools and techniques to solve real-world data problems. Students will gain knowledge in Python programming, data manipulation using NumPy and Pandas, and data cleaning and visualization techniques, making them well-prepared for careers in data science, analytics, and related fields. The skills learned in this course are directly applicable to analyzing large datasets, performing complex data operations, and generating meaningful insights, which are crucial in various industries such as finance, healthcare, marketing, and technology.

Course Outcomes

Upon completion of this course, the learner will be:

CO1: Understanding and remembering Python's built-in data types and methods to solve basic data-related problems.

CO2: Applying efficient data storage and operations using NumPy arrays for numerical data processing.

CO3: Analyzing data using Pandas for advanced data manipulation tasks, identifying trends and patterns in datasets.

CO4: Evaluating data pre-processing techniques and creating visualizations using Pandas to communicate insights effectively.

Course Content

- Introduction to Data Science
- Why Python?
- Essential Python libraries
- Python Introduction: Features, Identifiers, Reserved words, Indentation, Comments
- Built-in Data types and their Methods: Strings, List, Tuples, Dictionary, Set
- Type Conversion
- Operators, Decision Making, Looping, Loop Control statement
- Math and Random number functions
- User-defined functions: function arguments & its types

Practical Component:

- 6. Implement basic Python programs for reading input from the console.
- 7. Perform operations on Python built-in data types: Strings, List, Tuples, Dictionary, Set.
- 8. Solve problems using decision and looping statements.
- 9. Handle numerical operations using math and random number functions.
- 10. Create user-defined functions with different types of function arguments.

Unit 2: Introduction to NumPy

(No. of Hours: 16)

- Arrays and Vectorized Computation
- The NumPy ndarray
- Creating ndarrays
- Data Types for ndarrays
- Arithmetic with NumPy Arrays
- Basic Indexing and Slicing
- Boolean Indexing
- Transposing Arrays and Swapping Axes
- Universal Functions: Fast Element-Wise Array Functions
- Mathematical and Statistical Methods

• Sorting, Unique and Other Set Logic

Practical Component:

- 6. Create NumPy arrays from Python Data Structures and Random Functions.
- 7. Manipulate NumPy arrays: Indexing, Slicing, Reshaping, Joining, and Splitting.
- 8. Perform computations using Universal Functions and Mathematical methods.
- 9. Import and analyze data from CSV files using NumPy.
- 10. Manipulate images using NumPy.

Unit 3: Data Manipulation with Pandas

(No. of Hours: 16)

- Introduction to Pandas Data Structures: Series, DataFrame
- Essential Functionality: Dropping Entries, Indexing, Selection, and Filtering
- Function Application and Mapping
- Sorting and Ranking
- Summarizing and Computing Descriptive Statistics
- Unique Values, Value Counts, and Membership
- Reading and Writing Data in Text Format

Practical Component:

- 6. Create Pandas Series and DataFrames from various inputs.
- 7. Perform data operations on CSV files using Pandas.
- 8. Conduct statistical analysis and operations on DataFrames.
- 9. Handle categorical data using Pandas.
- 10. Rename columns and restructure data using Pandas.

Unit 4: Data Cleaning, Preparation, and Visualization (No. of Hours: 16)

- Handling Missing Data
- Data Transformation: Removing Duplicates, Transforming Data Using a Function or Mapping, Replacing Values
- Detecting and Filtering Outliers

- String Manipulation: Vectorized String Functions in Pandas
- Plotting with Pandas: Line Plots, Bar Plots, Histograms, Density Plots, Scatter Plots

Practical Component:

- 5. Handle missing data and perform data transformations using Pandas.
- 6. Detect and filter outliers in datasets.
- 7. Execute vectorized string operations in Pandas.
- 8. Visualize data using various plotting techniques.

Learning Experience

This course will combine lectures, hands-on sessions, and interactive activities to equip students with Python programming skills for data science.

Instruction Methods:

- Lectures: Core Python programming concepts and data science principles will be introduced through multimedia presentations and live coding demonstrations.
- Hands-on Sessions: Students will work on real-world data sets using Python, applying concepts learned in class through practical exercises.
- **Group Work and Case Studies:** Collaborative projects will reinforce learning, with case studies to address real-world data science challenges.
- Technology Use:
- **Python, NumPy, Pandas:** These tools will be used for data analysis, manipulation, and visualization.
- Jupyter Notebooks: For executing and documenting Python code.
- Online Platforms: LMS for accessing resources, recorded lectures, and submitting assignments.
- Assessments:
- Formative: Regular quizzes, coding exercises, and assignments will provide continuous feedback on students' progress.

• **Summative:** Project presentations, case study analyses, and a final assessment will evaluate students' understanding and application of Python for data science.

Support: The course instructor will offer continuous guidance and feedback. Peer collaboration will be encouraged through group work and review sessions. Students will have access to online resources and office hours to seek additional help when needed. Regular feedback will ensure that students meet the course outcomes effectively.

Textbooks

- 1. Y. Daniel Liang, "Introduction to Programming using Python," Pearson, 2012.
- 2. Wes McKinney, "Python for Data Analysis: Data Wrangling with Pandas, NumPy, and IPython," O'Reilly, 2nd Edition, 2018.
- 3. Jake VanderPlas, "Python Data Science Handbook: Essential Tools for Working with Data," O'Reilly, 2017.

Suggested Readings

- 1. Wesley J. Chun, "Core Python Programming," Prentice Hall, 2006.
- 2. Mark Lutz, "Learning Python," O'Reilly, 4th Edition, 2009.
- Joel Grus, "Data Science from Scratch: First Principles with Python," O'Reilly, 2015.

Open Educational Resources (OER)

- 1. NPTEL Python for Data Science
- 2. Kaggle's Python for Data Science
- 3. Awesome Python for Data Science (GitHub)

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	/
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

SEMESTER IV					
UDT109	Data Structures and Algorithms	L	Т	Р	С
Version		4	0	0	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co-	Nil				
Requisites					

Course Perspective

The Data Structures and Algorithms course provides students with a deep understanding of fundamental data structures and the algorithms used to manipulate them. Students will learn to design, analyze, and implement efficient algorithms to solve complex computational problems.

The course covers topics such as arrays, linked lists, trees, graphs, sorting, searching, and optimization techniques. By mastering these concepts, students will develop the skills to write optimized code, improve problem-solving abilities, and prepare for advanced studies or careers in computer science and software development.

Course Outcomes

Upon completion of the course, the learner will be:

Course Content

CO1: Understanding and remembering key data structures such as arrays, linked lists, stacks, queues, trees, and graphs for effective data management.

CO2: Applying knowledge to design, implement, and analyze algorithms for various computational tasks, assessing their efficiency in terms of time and space complexity.

CO3: Analyzing complex real-world problems and selecting appropriate data structures and algorithms to optimize performance and resource usage.

CO4: Evaluating and improving existing algorithms, enhancing their efficiency and preparing them for more advanced studies or professional challenges in computer science.

Unit I: Oops Concepts

12 hours

Class, Object, Constructors, type of variables, type of methods. Inheritance: single, multiple, multi-level, hierarchical, hybrid. Polymorphism: with functions and objects, with class methods, with inheritance. Abstraction: abstract classes.

Unit II: Introduction to Data Structures and Basic Algorithms 15 hours

Overview of Data structures and their importance. Introduction to arrays and listsunderstanding linear data structures. Implementing arrays and lists in Python. Basic operations on arrays and lists: insertion, deletion, searching. Implementing stack and queues in Python. Stack Operations: push, pop, peek. Queue Operations: enqueue, dequeue, peek. Time complexity, amortize time complexity and space complexity analysis: Big O notation, Big omega notation and Big theta notation.

Problem-Solving Exercise: Parenthesis Matching, Tower of Hanoi, implementing a stackbased algorithm (Reversing a string).

Unit III: Advanced Data Structures and Sorting Algorithms20 hours

Introduction to Linked Lists and trees, Implementing Linked lists and Binary trees in Python. Introduction to searching algorithms: Linear search, Binary search. Introduction to sorting algorithms: Bubble sort, selection sort, insertion sort. Divide and Conquer algorithms: Merge Sort and quick sort algorithms.

Problem-Solving Exercises: Longest Common Subsequence, Longest Increasing Subsequence, Word Break Problem, Subset Sum Problem, Binary Search, Merge Sort, Quick Sort.

Unit IV: Graph Algorithms and Dynamic Programming17 hours

Introduction to graphs: representation and traversal. Depth-First Search (DFS) and Breadth-First Search (BFS). Shortest Path Algorithms: Dijkstra's Algorithm, Bellman-Ford Algorithm. Introduction to Dynamic Programming: Principles and Applications. Solving problems using dynamic programming.

Problem-Solving Exercises: Travelling Salesman Problem, Floyd-Warshall Algorithm, Knapsack Problem, Longest Increasing Subsequence (LIS) using Dynamic Programming.

Learning Experience

The *Data Structures and Algorithms* course offers a dynamic learning experience focused on both theory and practical application. Students will engage in interactive lectures, hands-on programming exercises, and collaborative projects to master key data structures and algorithms. Algorithmic challenges and real-world case studies will enhance problem-solving skills and demonstrate practical applications. Continuous feedback through quizzes and coding reviews will guide students' progress. The course emphasizes the use of industry-standard tools, encouraging students to write, optimize, and reflect on their code, preparing them for advanced studies and professional challenges.

Textbooks

- 1. Michael T. Goodrich: Data structures and algorithms in Python
- Thomas H. Cormen, Charles E. Leiserson, Ronald L. Rivest, and Clifford Stein: Introduction to Algorithms

Suggested Readings

- Aaron M. Tenenbaum, Yedidyah Langsam and Moshe J. Augenstein "Data Structures Using C and C++", PHI Learning Private Limited, Delhi India
- 2. Berztiss, A.T.: Data structures, Theory and Practice :, Academic Press.
- 3. Jean Paul Trembley and Paul G. Sorenson, "An Introduction to Data Structures with applications", McGraw Hill.

Open Educational Resources (OER)

- 1. <u>https://www.coursera.org/specializations/data-structures-algorithms</u>
- 2. <u>https://www.khanacademy.org/computing/computer-science/algorithms</u>
- 3. <u>https://www.coursera.org/specializations/algorithms</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	,
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks

Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

Create a simple sorting algorithm, such as Bubble Sort or Insertion Sort, and implement it in your preferred programming language. Write a program to sort an array of integers and analyze its time complexity. After implementation, discuss with peers how different sorting algorithms compare in terms of efficiency and practical use cases. Submit both your code and a brief report on your findings.

UDT110	Fundamentals of Intelligence	of	Artificial	L	Т	Р	С
Version				2	0	4	4
Category of Course	Minor			1			
Total Contact Hours	64 Hours						
Pre-Requisites/ Co)-						
Requisites							

Course Perspective

The *Fundamentals of Artificial Intelligence* course introduces students to core concepts and techniques in AI, including machine learning, neural networks, natural language processing, and computer vision. Students will explore how AI systems are designed, trained, and evaluated, gaining practical experience with tools and algorithms used in the field. The course

emphasizes both theoretical understanding and hands-on application, preparing students for advanced studies or careers in AI. By the end of the course, students will have a solid foundation in AI principles and the ability to implement basic AI solutions.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding and explaining core AI concepts, including machine learning, neural networks, natural language processing, and computer vision.

CO2: Applying AI models and algorithms using popular frameworks and tools, demonstrating skills in data preprocessing, model training, and evaluation.

CO3: Analyzing data by developing custom R functions and utilizing control structures to automate data analysis processes effectively.

CO4: Evaluating the effectiveness of different data preprocessing techniques, selecting appropriate methods for cleaning and preparing data for analysis.

Course Content

UNIT I: Introduction to Artificial Intelligence

Definition of Intelligence, Artificial Intelligence, Historical overview, Importance of AI, Real time applications, Turing Test, key milestones in AI, State of the Art in AI Differentiating AI from human intelligence; Types of Knowledge, Intelligent Agents and their structure, Risk and Benefits of AI.

UNIT II: Informed and Uninformed Search Strategies 20 hours

Problem Representation techniques, Declarative and Procedural representations, Search algorithms for problem solving; Uninformed Search Strategies: Breadth-first search, Depth First search, Uniform Cost search; Informed search (Heuristic Based) strategies: Hill Climbing, Greedy best first search, A* Search – admissibility and optimality.

UNIT III: Artificial Intelligence and Python

270

12 hours

17 hours

Agent architectures and hierarchical controllers; Using Python to search in continuous spaces, search with non-deterministic actions, search in partially observable environments; online search agents; constraint satisfaction problems; Game theory; Reasoning, Learning, Planning with uncertainty; Reinforcement Learning

UNIT IV: Applications of AI

15 hours

AI in healthcare: Diagnosis, treatment, and medical imaging; AI in finance: Fraud detection, algorithmic trading, and risk assessment; AI in transportation: Autonomous vehicles and traffic optimization; AI in customer service and chatbots; AI in education: Personalized learning and intelligent tutoring systems; AI and creativity: Generative models and artistic applications; Ethical and Social Implications of AI

Learning Experience

The *Fundamentals of Artificial Intelligence* course offers an immersive learning experience with a blend of theoretical and practical approaches. Students will engage in interactive lectures covering core AI concepts, and hands-on projects to implement AI models using tools like TensorFlow and PyTorch. Real-world case studies will illustrate AI applications and challenges. Collaborative activities will enhance problem-solving skills, while continuous feedback through assignments and quizzes will support learning. The course aims to build a solid foundation in AI principles and their practical applications.

Textbooks

 Stuart Russell & Peter Norvig, Artificial Intelligence: A Modern Approach, Prentice-Hall, Third Edition (2009).

Suggested Readings

- 1. David L. Poole and Alan K. Mackworth, Python code for Artificial Intelligence Foundations of Computational Agents, Version 0.9.12 of January 18, 2024.
- Ian GoodFellow, Yoshua Bengio & Aaron Courville, Deep Learning, MIT Press (2016).

Open Educational Resources (OER)

- 1. https://www.elementsofai.com/
- 2. <u>https://www.iu.org/en-in/blog/ai-and-education/best-ai-tools-for-students/</u>
- 3. https://ep.jhu.edu/programs/artificial-intelligence/courses/

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	
Project/quizzes/assignment and essays/presentation/	,
participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

In the *Fundamentals of Artificial Intelligence* course, students will engage in activities such as implementing AI models through hands-on projects, analyzing case studies of AI applications, participating in group discussions on ethical implications, and completing coding assignments using AI frameworks. Interactive quizzes and practical exercises will reinforce learning and enhance their understanding of AI concepts.

UDT106	Fundamental of	MachineL	Т	Р	С
	Learning				
Version		2	0	4	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co	- Basic concepts of Statist	ics			
Requisites					

Course Perspective

This course provides essential skills in machine learning, covering both supervised and unsupervised techniques using Python. Students will learn to preprocess data, apply regression and classification models, and utilize clustering methods. The course emphasizes practical application, preparing students to tackle real-world problems and make impactful contributions in fields like business, healthcare, and technology.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Remembering and understanding machine learning concepts to identify suitable applications, distinguishing between supervised and unsupervised learning, and utilizing appropriate Python libraries for machine learning tasks.

CO2: Applying preprocessing techniques to prepare data for analysis by handling missing values, encoding categorical data, and applying normalization, standardization, and Principal Component Analysis (PCA) techniques.

CO3: Analyzing and developing supervised learning models, including linear and nonlinear regression, K-Nearest Neighbour, Decision Trees, Logistic Regression, and Support Vector Machines, to address various predictive and classification problems. **CO4**: Evaluating the performance of machine learning models using appropriate evaluation methods to ensure the development of accurate and effective predictive models.

Course Content

Unit I: Introduction to Machine Learning No. of Hours: 16 Hours

- Application of Machine Learning
- Supervised vs Unsupervised Learning
- Python libraries suitable for Machine Learning

Unit II: Data Pre-Processing and Data No. of Hours: 16 Hours

- Identifying and handling the missing values
- Encoding the categorical data
- Normalization
- Standardization
- PCA

Unit III: Supervised Learning Regression and Classification No. of Hours: 16 Hours

- Linear Regression
- Non-Linear Regression
- Model evaluation methods
- K-Nearest Neighbour
- Decision Tree
- Logistic Regression
- Support Vector Machines,
- Model Evaluation

Unit IV: Unsupervised Learning

No. of Hours: 16 Hours

- K-means Clustering
- Hierarchical Clustering
- Density-Based Clustering

Learning Experience

This course will combine lectures, interactive sessions, and hands-on projects to enhance understanding of machine learning concepts, data preprocessing, and model implementation. Students will engage in practical exercises to apply supervised and unsupervised learning techniques, ensuring a comprehensive learning experience.

Lectures:

Instruction Methods:

- Lectures: Core machine learning concepts will be taught using multimedia presentations and real-world case studies.
- **Interactive Sessions:** Q&A, coding exercises, and group discussions will actively engage students in applying machine learning techniques and solving practical problems.
- **Group Work and Case Studies:** Collaborative projects and case studies will reinforce learning and promote teamwork.
- Technology Use:
- **R and RStudio:** Students will use R and RStudio for data manipulation, visualization, and analysis.
- Shiny: For creating interactive web applications and visualizations.
- Online Platforms: LMS for accessing resources, recorded lectures, and submitting assignments.
- Assessments:
- Formative: Regular quizzes, coding exercises, and assignments for continuous feedback.
- **Summative:** Project presentations, case study analyses, and a final assessment to evaluate students' mastery of R programming and data analysis.

Support: The course instructor will provide continuous guidance, with opportunities for students to collaborate through group work and peer reviews. Regular feedback will be given on assignments and projects, and students are encouraged to seek help as needed to enhance their learning experience.

Textbooks

- 1. Machine Learning Tom M. Mitchell
- 2. Python Machine Learning Sebastian, Raschka and Vahid Mirjalili

Suggested Readings

- 1. Understanding Machine Learning Shai Shalev-Shwartz and Shai Ben-David La
- 2. Hands-On Machine Learning with Scikit-Learn and TensorFlow: Concepts, Tools, and Technique to Build Intelligent Systems-AurelienGeron

Open Educational Resources (OER)

- 1. https://www.coursera.org/learn/machine-learning
- 2. <u>https://www.datacamp.com/tutorial/introduction-machine-learning-python</u>

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
I. Continuous assessment (30 marks)	
All the components to be evenly spaced	

Project/quizzes/assignment and essays/prese	ntation/
participation/case studies/reflective journals (minimum	of five
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
III. External Marks (Theory):	50 Marks
End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

Students will participate in activities such as data cleaning, summarization, and visualization tasks. They will engage in assignments, quizzes, and group discussions, focusing on applying the concepts learned to real-time data. These activities will reinforce the theoretical knowledge acquired and provide practical experience in data analytics.

SEMESTER VII					
UDT111	Neural Network and Learning	DeepL	Т	Р	С
Version		2	0	4	4
Category of Course	Minor				
Total Contact Hours	64 Hours				
Pre-Requisites/ Co-	•				
Requisites					

Course Perspective

Neural Networks and Deep Learning explore the intricacies of artificial neural networks, focusing on how they mimic human brain processes to recognize patterns and make predictions. This course covers fundamental concepts like perceptrons, activation functions, and backpropagation, along with advanced topics such as convolutional and recurrent neural networks. Students will gain practical experience in designing, training, and evaluating models, enabling them to tackle complex problems across various domains, from image recognition to natural language processing.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding the fundamental principles of neural networks, including architecture, activation functions, and optimization techniques.

CO2: Applying knowledge to design, train, and evaluate different types of neural network models such as feed forward, convolutional, and recurrent networks.

CO3: Analyzing the performance of neural network models by interpreting outputs, performing error analysis, and refining models to improve accuracy and efficiency.

CO4: Evaluating deep learning frameworks and tools to solve real-world problems in areas like image classification, natural language processing, and time-series analysis.

Course Content

Unit I: Fundamentals of Neural Network and Training

Introduction to neural network and their importance, biological inspiration of neural network, Historical overview; Perceptron: The basic neural unit; multi-layer perceptron and need for

hidden layers, Activation units and their roles, Cost/Loss function and performance measurement.

Unit II: Feedforward Neural Networks

15 hours

12 hours

Feedforward neural network architecture, training a neural network, determining hidden layers, backpropagation for weight updates, stochastic, gradient descent, mini-batch gradient descent.

19 hours

Unit III: Convolution Neural Networks

Convolution neural networks for image data, CNN building blocks- convolution, pooling, LeNet, AlexNet – pioneer CNN architectures, VGGNet, ResNet, and other modern CNN architectures, Data Augmentation techniques for computer vision, Applications of CNN- image recognition, object detection.

Unit IV: Recurrent Neural Networks and Sequences 18 hours

Recurrent neural network architectures (RNNs), Handling sequential data like text, speech, time series, Long Short-Term Memory (LSTM) models, Gated Recurrent Units (GRUs), Transformer Model, Applications like machine translation, text generation.

Learning Experience

The learning experience in Neural Networks and Deep Learning is immersive and hands-on, combining theoretical knowledge with practical application. Students engage with foundational concepts through interactive lectures and exercises, exploring how neural networks mimic brain functions. They gain practical skills by designing, training, and evaluating various models using popular deep learning frameworks. Real-world projects and case studies enhance understanding, while regular feedback and peer collaboration foster a deeper grasp of complex topics. This approach equips students with both the theoretical insights and practical skills needed for success in the field.

Textbooks

 "Neural Networks and Deep Learning" by Michael Nielsen (Determination Press, 2015).

Suggested Readings

1. "Pattern Recognition and Machine Learning" by Christopher Bishop.

2. "Hands-on Machine Learning with Scikit-Learn, Keras and Tensor Flow" by Aurelien Geron (O Reilly, 2019).

Open Educational Resources (OER)

1 http://neuralnetworksanddeeplearning.com/

- 2. https://www.coursera.org/specializations/deep-learning/
- 3. https://cs231n.stanford.edu/

Evaluation Scheme

Evaluation components	Weightage
Internal marks (Theory)	30 Marks
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All the components to be evenly spaced	
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participation/case studies/reflective journals (minimum of five	
components to be evaluated)	
II. Internal marks (Theory):	20 Marks
Mid Term Examination	
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End Term Examination	

It is compulsory for a student to secure 40 % marks in Internal and End Term Examination separately to secure minimum passing grade

Student Activity

Develop a neural network model to classify images from a provided dataset using a deep learning framework like TensorFlow or PyTorch. Students will preprocess the data, design and train the model, and evaluate its performance. They will then present their results, discussing challenges faced, model accuracy, and potential improvements. This activity emphasizes hands-on experience, problem-solving skills, and effective communication of technical findings.

SEMESTER VIII					
UDT112	Natural Language Processing	L	Т	Р	С
Version		2	0	4	4
Category of Course	Minor		1	I	
Total Contact Hours	64 Hours				
Pre-Requisites/ Co-					
Requisites					

Course Perspective

This course explores the foundations and advancements in Natural Language Processing (NLP) and Generative AI, focusing on the intersection of language and machine learning. Students will learn to analyze, understand, and generate human language using various NLP techniques. The course covers essential topics such as text processing, language models, machine translation, and sentiment analysis, alongside cutting-edge generative AI models like GPT and transformers. By the end, students will be equipped with the skills to build intelligent systems that comprehend and generate human-like text.

Course Outcomes

Upon completion of the course, the learner will be:

CO1: Understanding the core principles and techniques of Natural Language Processing (NLP), including tokenization, parsing, and language models.

CO2: Applying machine learning models to solve language-related tasks such as sentiment analysis, text classification, and named entity recognition.

CO3: Analyzing linguistic data and evaluating the performance of models using metrics like precision, recall, and F1 score.

CO4: Evaluating ethical considerations and challenges associated with AI in language processing, such as bias, privacy, and fairness.

Course Content

Unit I: Introduction to Natural Language Processing

Natural language Processing, Applications of NLP (chatbots, machine translation, sentiment analysis, etc.), Basic Text processing: tokenization, stopword removal, stemming/lemmatization, Vector representation of text (bag-of-word, TF-IDF, word embeddings).

Unit II: Language Learning Models

Introduction to Language models and n-grams, Regular expressions and pattern matching, Text normalization and data cleaning, Exploratory data analysis for text data. Supervised vs. unsupervised learning for NLP tasks, Text classification with logistic regression and naïve bayes, Sequence labelling with conditional random files (CRF), Evaluation metrics for NLP (accuracy, F1-score, perplexity), Neural network basics (feedforward, backpropagation).

Unit III: Voice Processing and Speech Recognition

Introduction to voice processing and its importance in NLP, Fundamentals of speech signals and acoustics. Speech pre-processing techniques; noise reduction, normalization, and feature extraction (MFCC, spectrograms), Automatic Speech Recognition (ASR) systems: Hidden Markov Models (HMMs), Gaussian Mixture Models (GMMs). Text-to-Speech synthesis: WaveNet, Tacotron, and other modern architectures.

Unit IV: Deep Learning for NLP 10 hours

Recurrent neural networks (RNNs) for sequence modelling, Long Short-Term Memory (LSTM) and Gated Recurrent Units (GRUs), Encoders, decoders and sequence-to-sequence models, Attention mechanisms and Transformer architecture, Pretrained language models (ELMo, BERT, GPT) and transfer learning, Named Entity Recognition (NER) with deep learning.

Unit V: Generative AI and Applications

Introduction to generative models and their applications, Text generation with language models and beam search, Image captioning and multimodal tasks, Generative adversarial networks (GANs) for text and image generation, Limitations, ethical considerations and future of generative AI, Open ended conversational AI and chatbots, ChatGPT.

7 hours

15 hours

15 hours

20 hours

Learning Experience

The learning experience in this course is highly interactive and hands-on, blending theoretical knowledge with practical applications. Students will engage in coding exercises, real-world projects, and collaborative group work to reinforce their understanding of Natural Language Processing (NLP) and Generative AI. Through the use of contemporary tools and frameworks, they will build and deploy models, analyze language data, and explore the ethical implications of AI. This approach ensures that students gain both technical expertise and critical thinking skills.

Textbooks

1. "Conversational AI" by Lili Zheng and Honglak Lee (2022).

Suggested Readings

- "An Introduction to Natural Language Processing, Computational Linguistics, and Speech Recognition" by Dan Jurafsky, James H. Martin, Prentice Hall, (2009)
- 2. "Natural Language Processing with Python" by Steven Bird, Ewan Klein, and Edward Loper (2009).

Open Educational Resources (OER)

- 1. <u>https://www.geeksforgeeks.org/top-natural-language-processing-nlp-books/</u>
- 2. https://www.iu.org/en-in/blog/ai-and-education/best-ai-tools-for-students/
- 3. <u>https://ep.jhu.edu/programs/artificial-intelligence/courses/</u>

Evaluation Scheme

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Student Activity

Implement a basic sentiment analysis model using Python. Students will use a dataset of text reviews to train a model that classifies the sentiment as positive, negative, or neutral. They will preprocess the text by removing stopwords and applying tokenization. After training the model, students will evaluate its accuracy using a test set. Finally, they will visualize the results and discuss potential improvements, considering different machine learning algorithms and feature extraction techniques such as TF-IDF or word embedding.